

CLOSED LOOP RECYCLING OF SOLID OXIDE CELLS

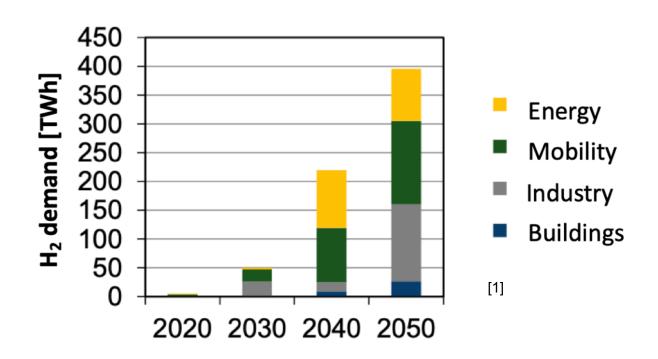
Hans-Walter-Hennicke Presentation Contest 2023

28/3/2023 I MARTIN HILGER I IEK-1, FORSCHUNGSZENTRUM JÜLICH





SOLID OXIDE CELLS FOR THE "ENERGIEWENDE"

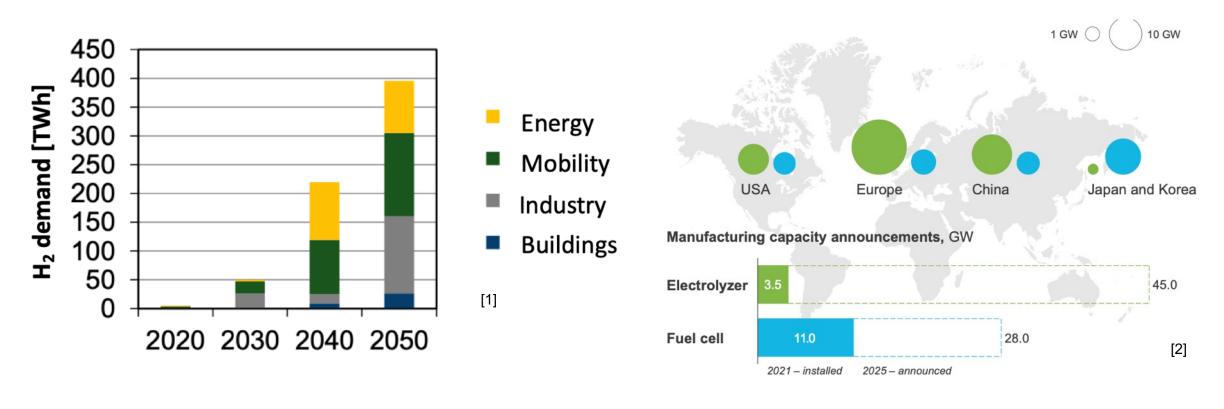


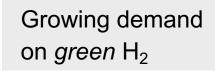
Growing demand on *green* H₂

[1] Robinius et al., Schriften des Forschungszentrums Jülich (Energy & Environment), 2020



SOLID OXIDE CELLS FOR THE "ENERGIEWENDE"







Electrolysis capacities



Resource consumption

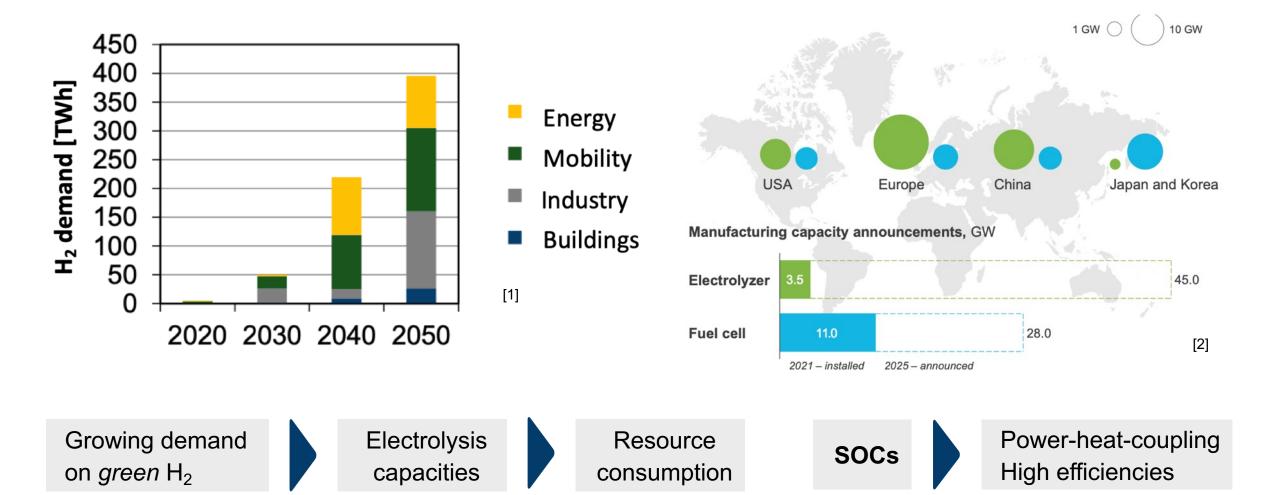
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[1] Robinius et al., Schriften des Forschungszentrums Jülich (Energy & Environment), 2020

[2] Hydrogen Insights report 2022, Hydrogen Council, McKinsey & Company



SOLID OXIDE CELLS FOR THE "ENERGIEWENDE"



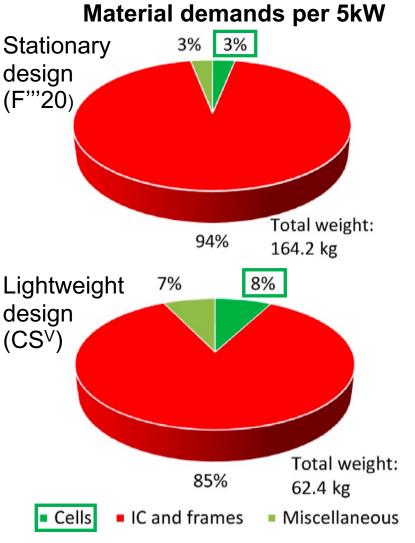
[1] Robinius et al., Schriften des Forschungszentrums Jülich (Energy & Environment), 2020

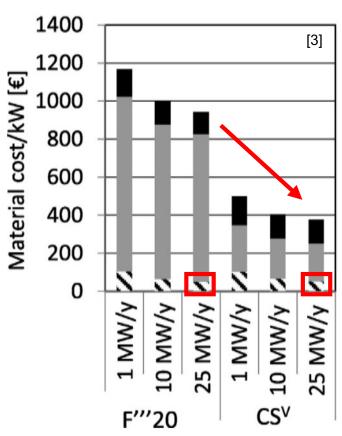
[2] Hydrogen Insights report 2022, Hydrogen Council, McKinsey & Company

SOCs = Solid oxide cells



WHY CLOSED LOOP RECYCLING OF SOCS?





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Lighter stack design

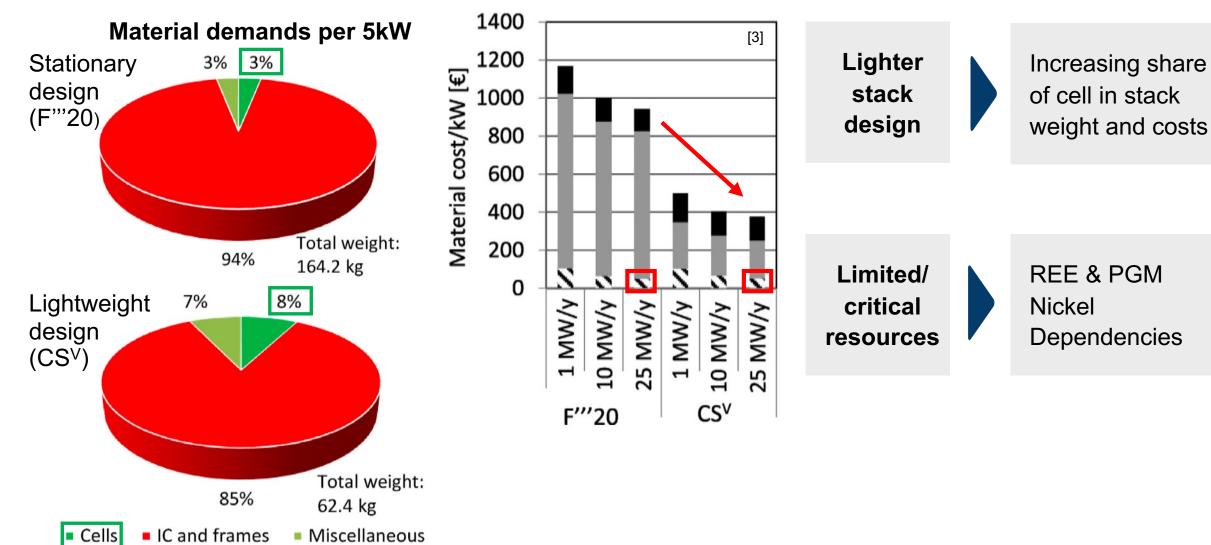


Increasing share of cell in stack weight and costs

[3] Harboe et al., Int. J. Hydrog. Energy, Vol. 45, 2020



WHY CLOSED LOOP RECYCLING OF SOCS?

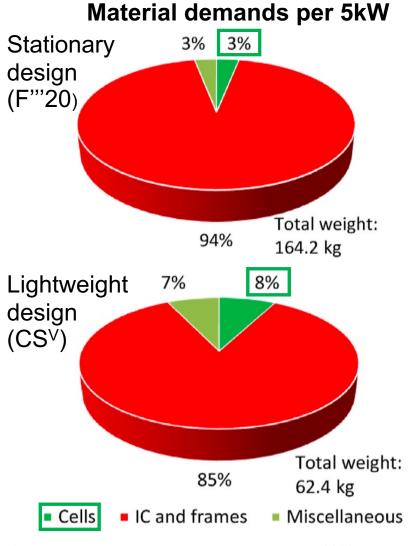


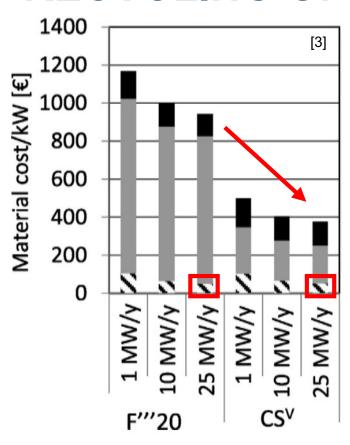
[3] Harboe et al., Int. J. Hydrog. Energy, Vol. 45, 2020

REE = Rare earth elements PGM = Platinum group metals



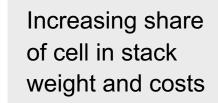
WHY CLOSED LOOP RECYCLING OF SOCS?





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Limited/ critical resources



REE & PGM Nickel Dependencies



Rising importance of closed loop recycling of cell material

[3] Harboe et al., Int. J. Hydrog. Energy, Vol. 45, 2020

REE = Rare earth elements PGM = Platinum group metals





Elcogen, Website, 2023

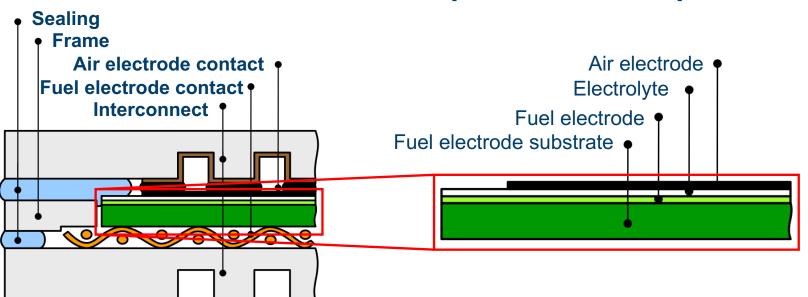
Mitglied der Helmholtz-Gemeinschaft



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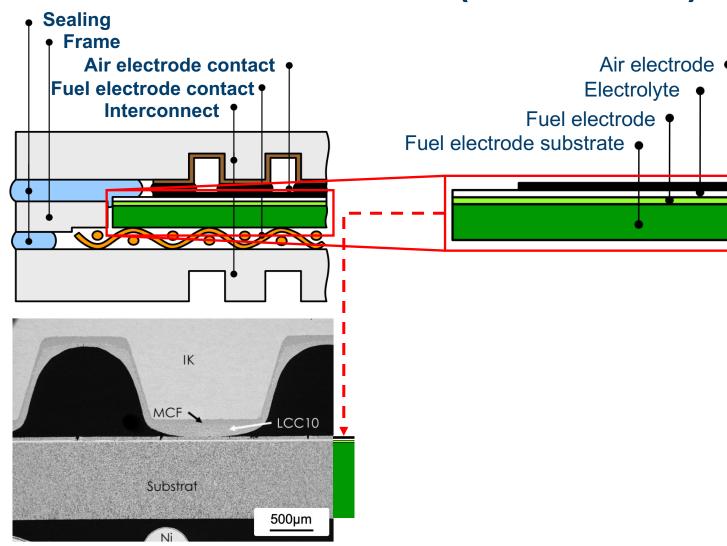
Elcogen, Website, 2023







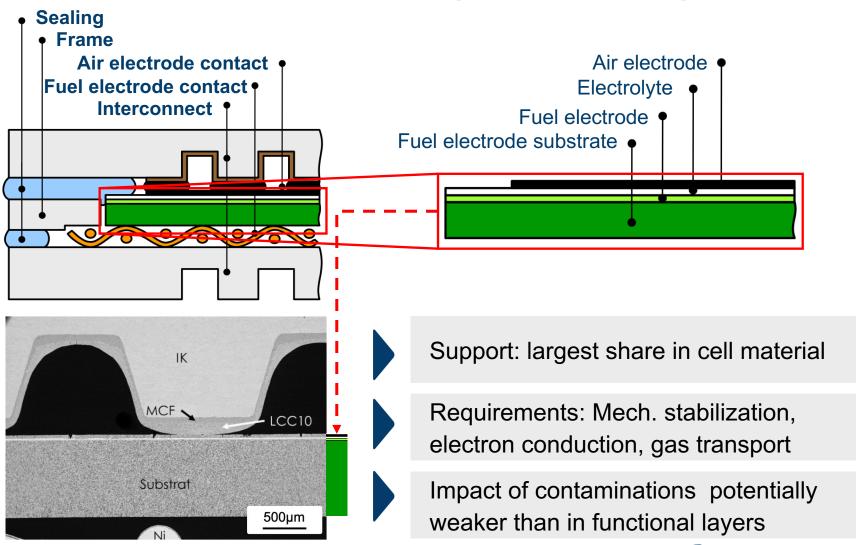
Elcogen, Website, 2023







Elcogen, Website, 2023





POSSIBLE RECYCLING CONCEPT (FZ JÜLICH)

Step Six

Stack testing or stack operation

Step Five

Integration into manufacturing process

Step Four

Powder preparation: Milling and sieving



Step One

Collecting EoL cell fragments

Step Two

Reoxidation of EoL cell fragments

Step Three

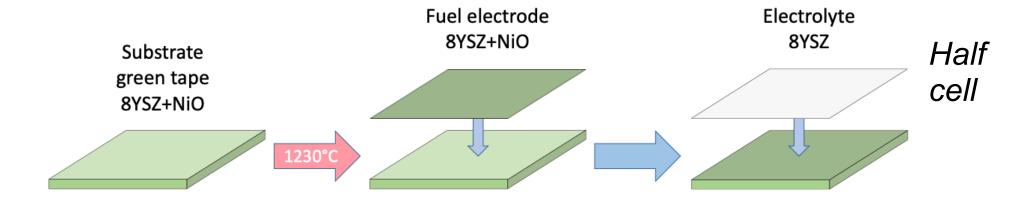
Separation: Mechanical or hydrometallurgical

Open-loop raw material recovery

Recycling concept developed by S. Sarner (IEK-1, Forschungszentrum Jülich)

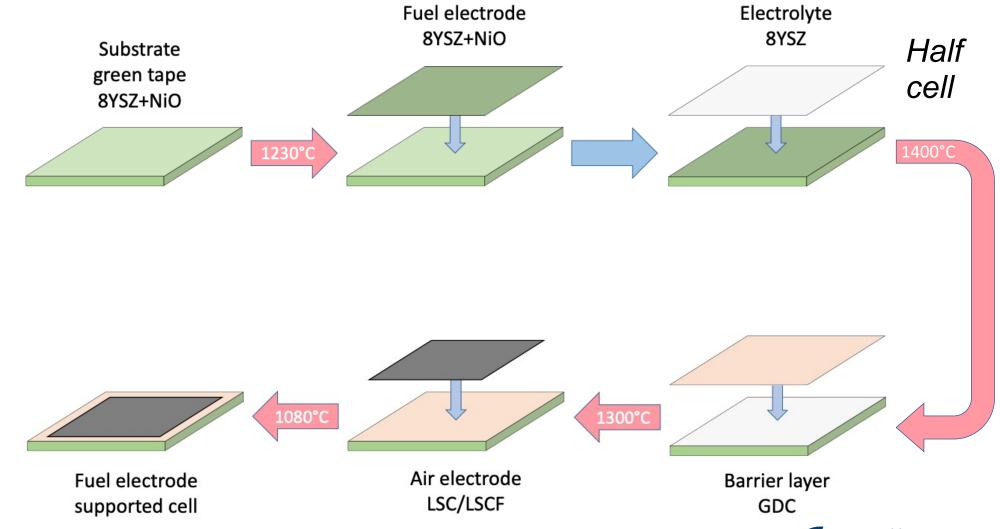








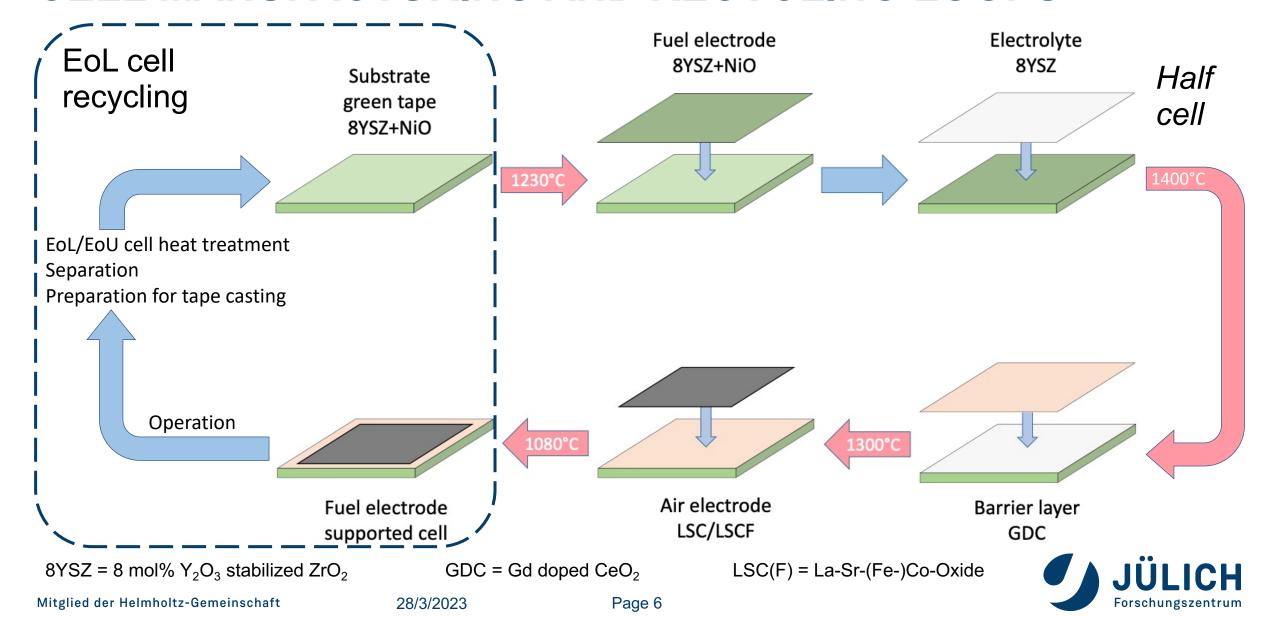
28/3/2023

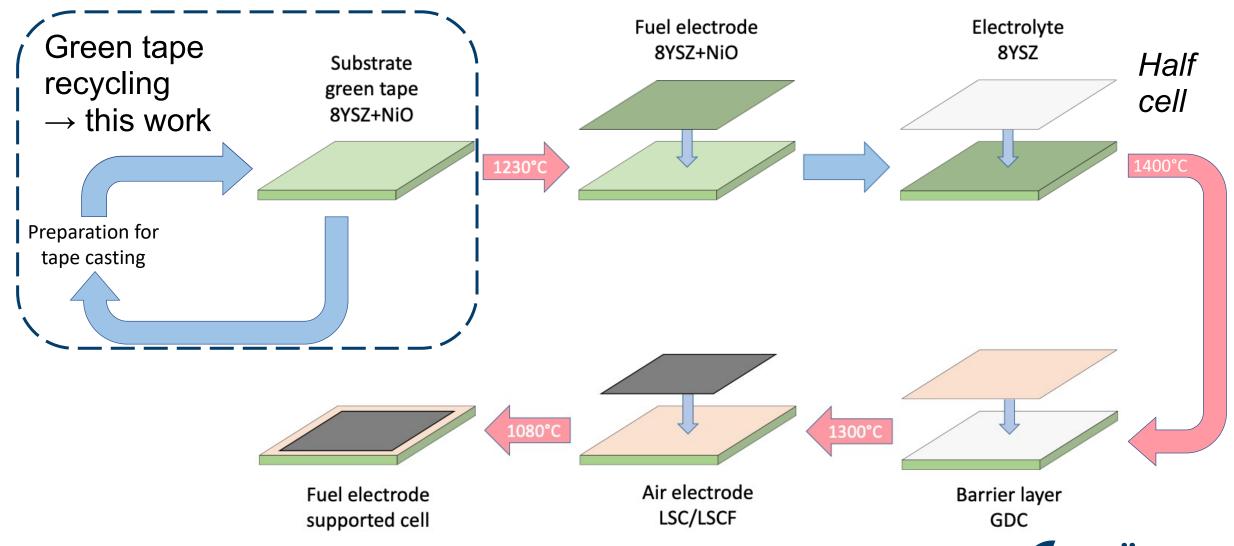


 $8YSZ = 8 \text{ mol}\% Y_2O_3 \text{ stabilized } ZrO_2$

 $GDC = Gd doped CeO_2$



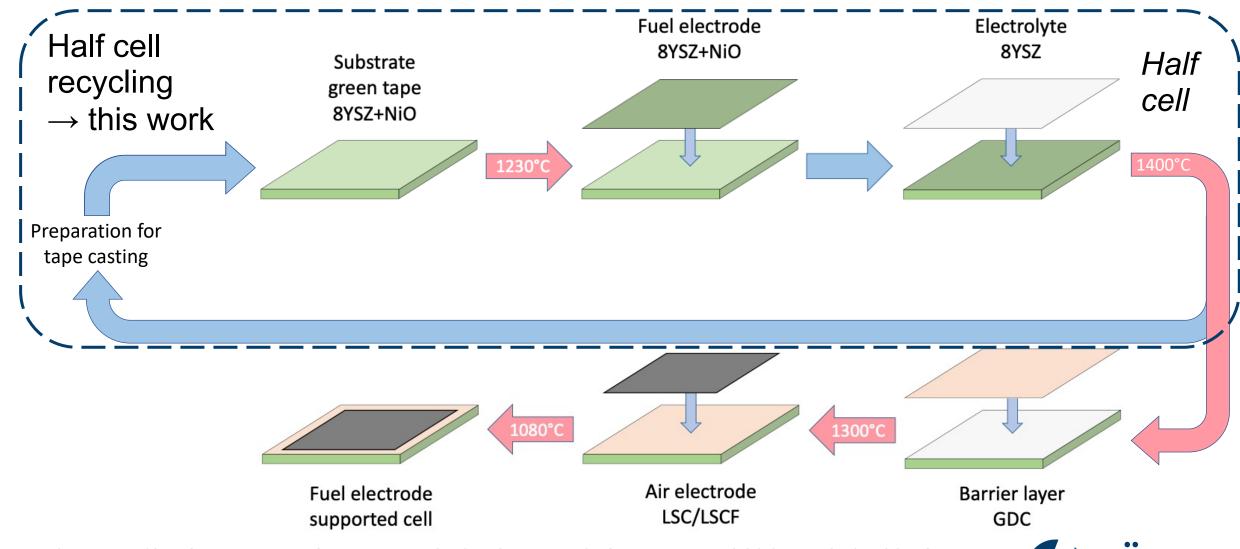




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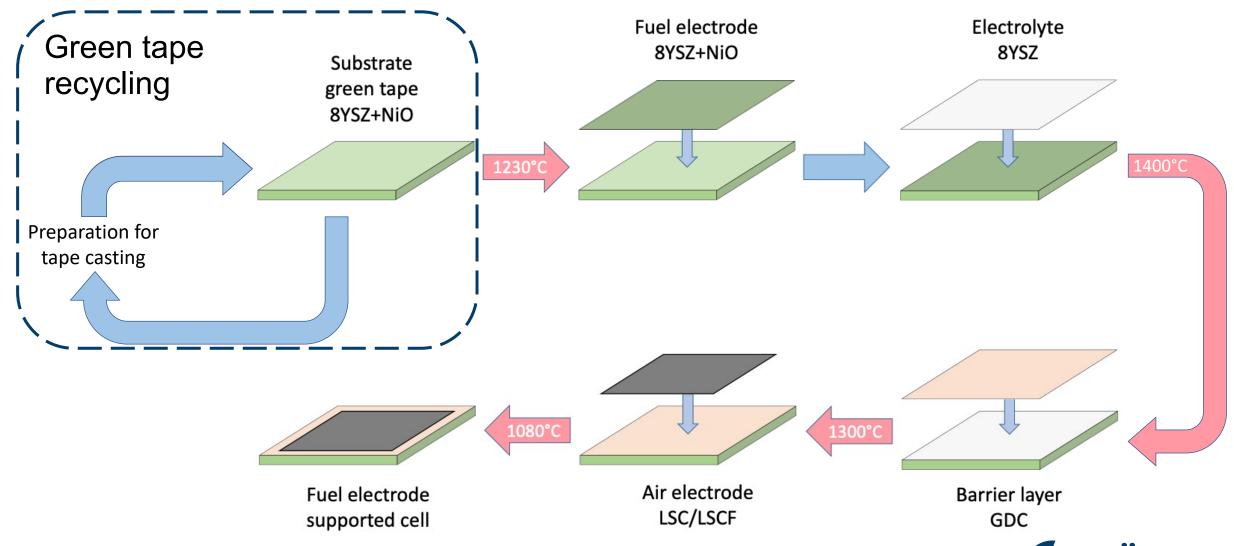


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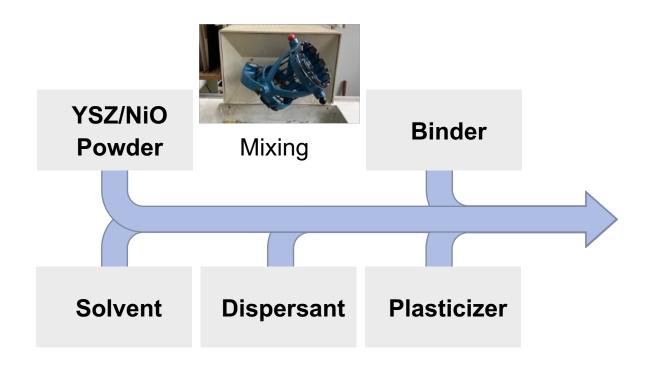




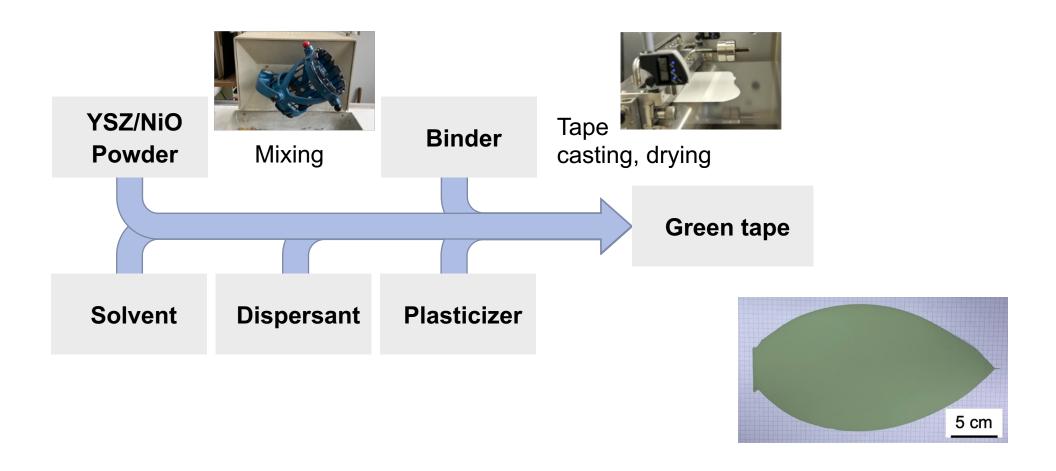
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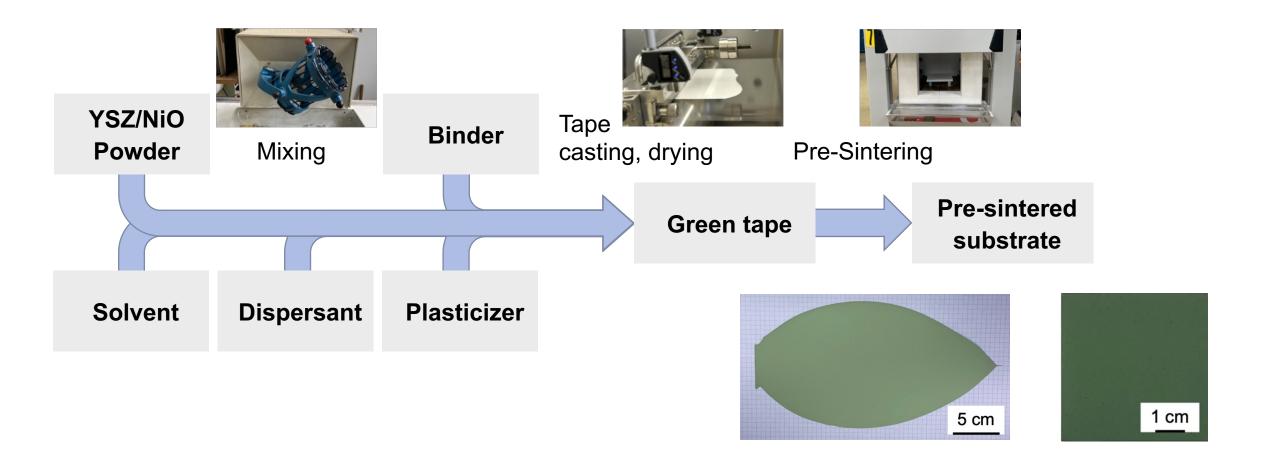




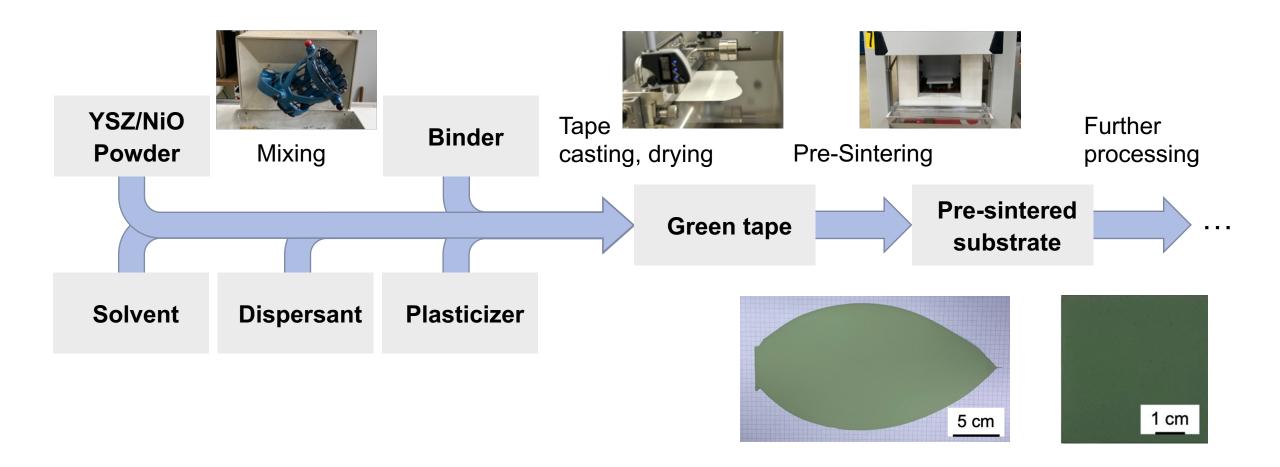




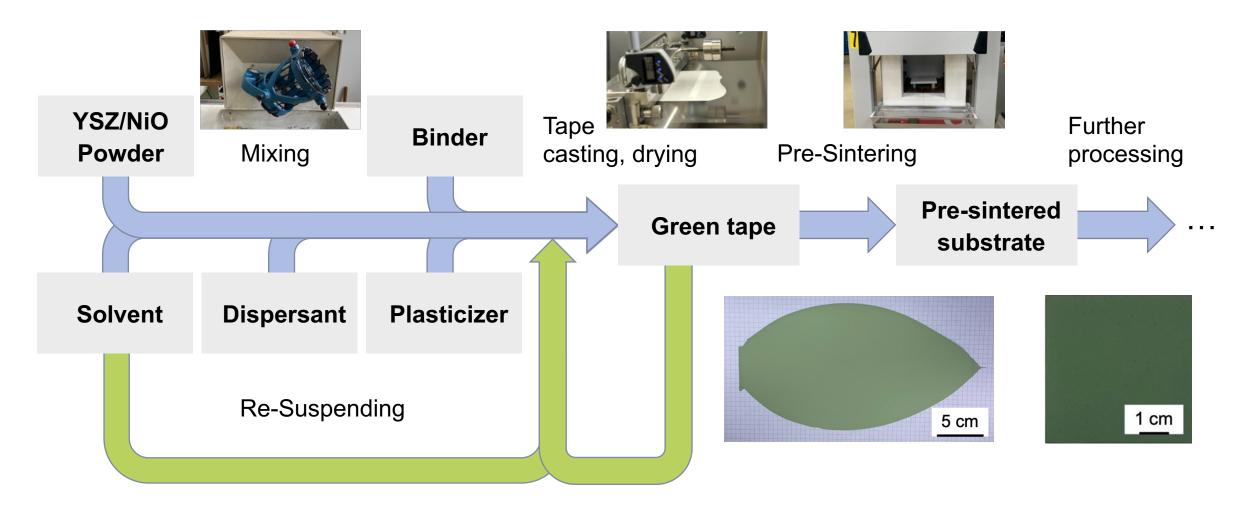




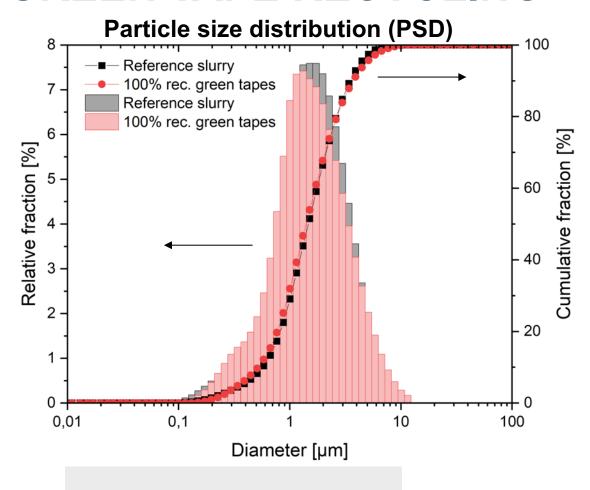






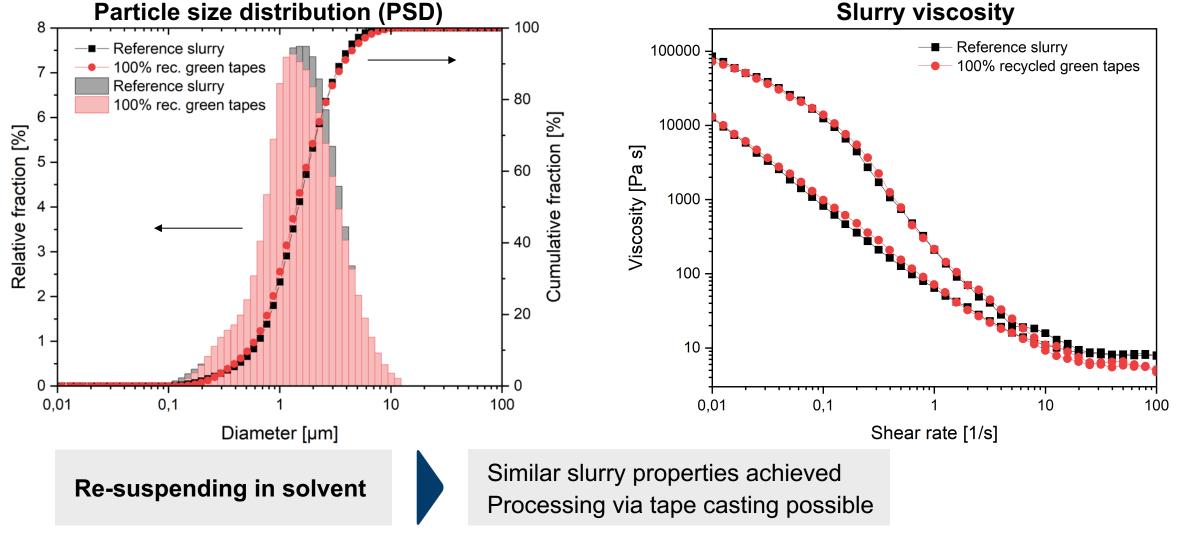




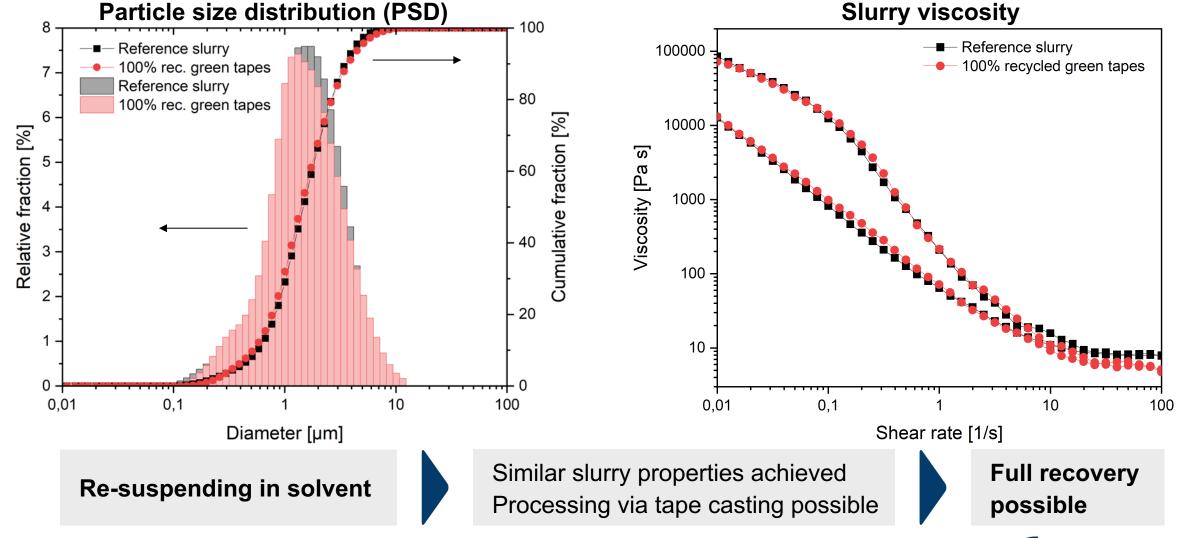


Re-suspending in solvent



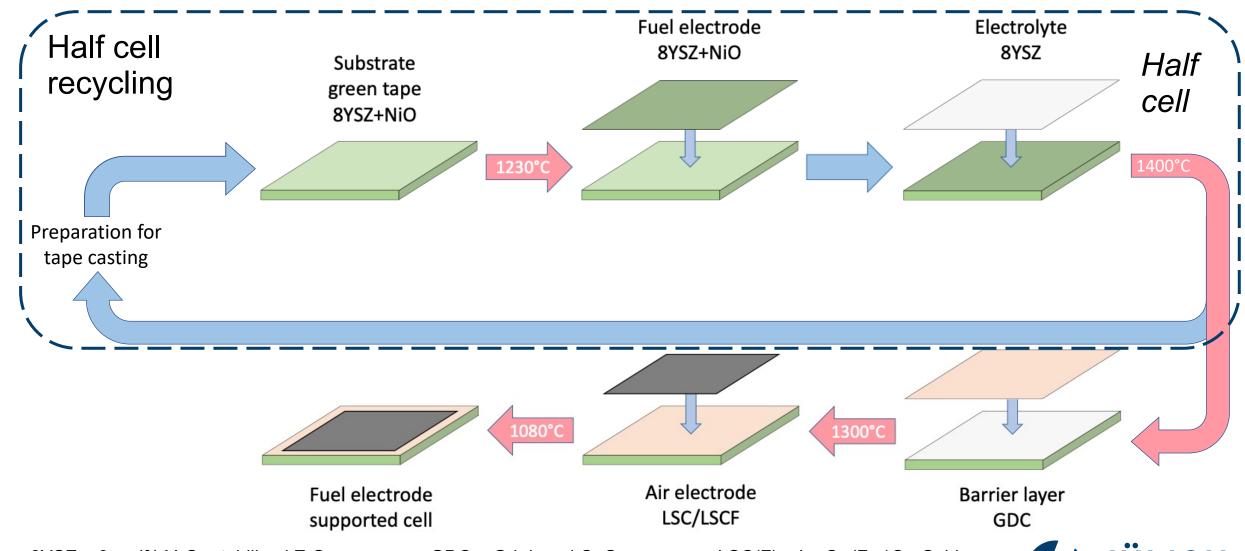








RECYCLING OF SINTERED HALF CELLS

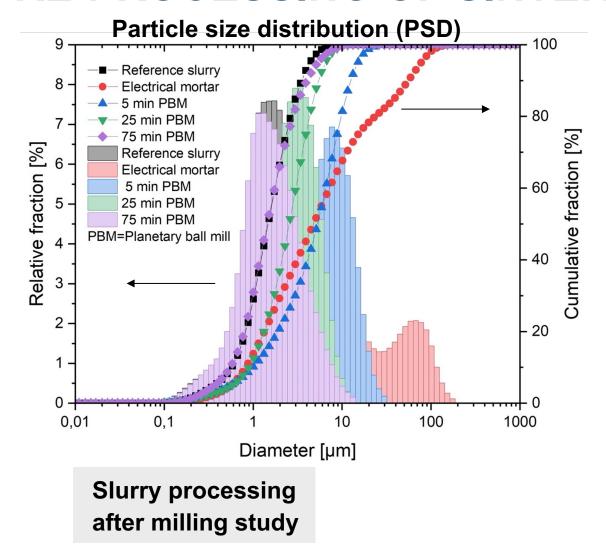


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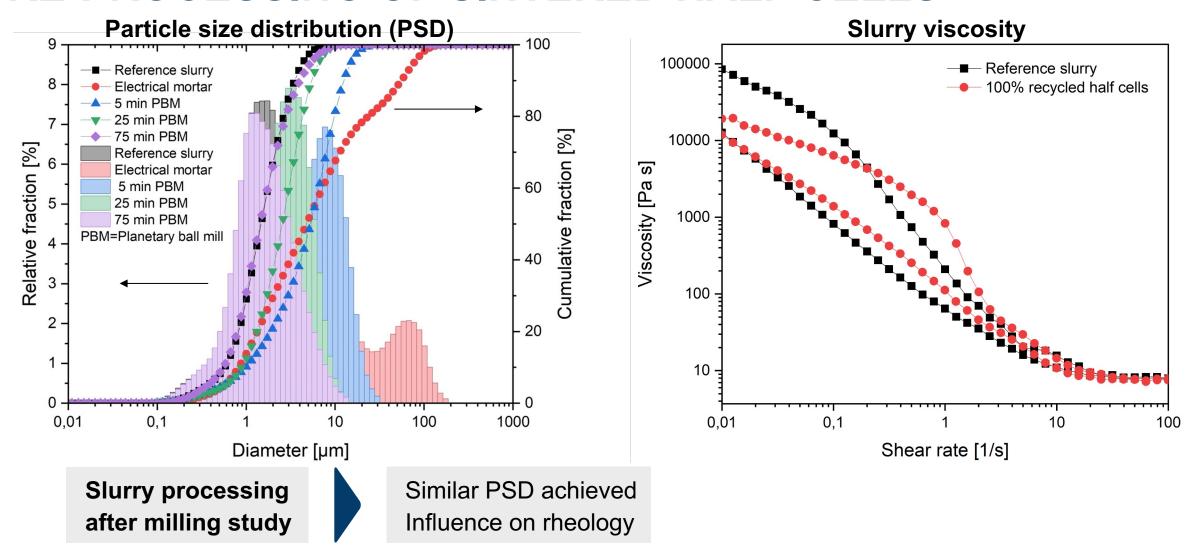


RE-PROCESSING OF SINTERED HALF CELLS



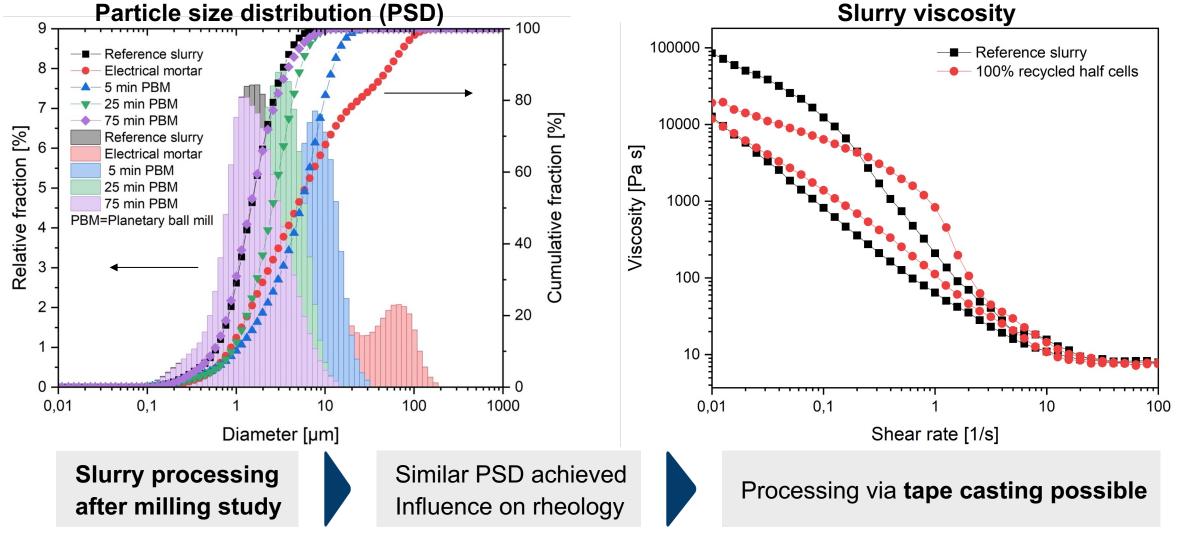


RE-PROCESSING OF SINTERED HALF CELLS



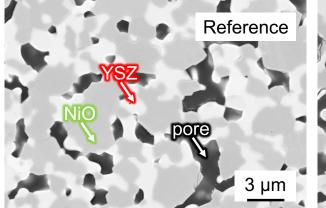


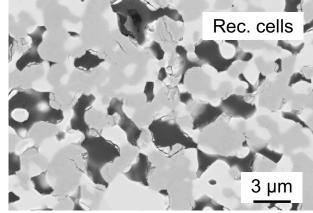
RE-PROCESSING OF SINTERED HALF CELLS





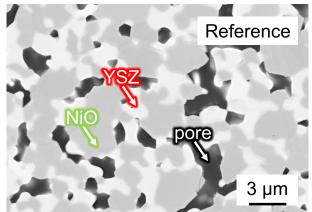
Sintered at 1400°C in air

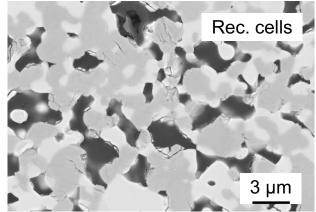






Sintered at 1400°C in air





Microstructure



Similar pore morphology

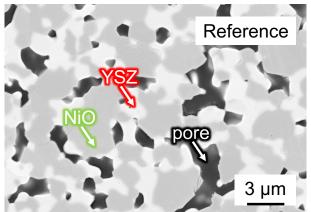
Sintering behavior

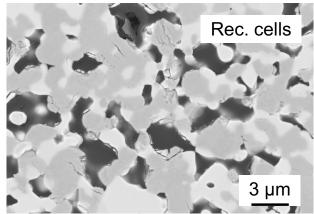


Lowered shrinkage: 13% (Reference 18%)



Sintered at 1400°C in air





Microstructure



Similar pore morphology

Sintering behavior



Lowered shrinkage: 13% (Reference 18%)



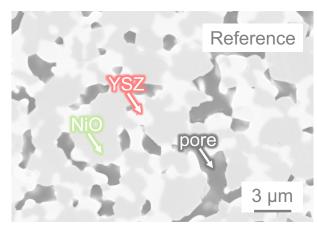
Affecting further cell processing route

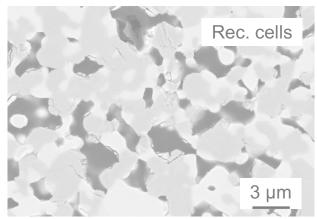


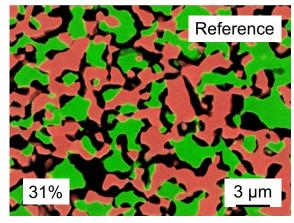
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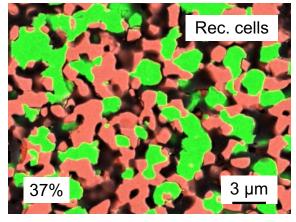
Sintered at 1400°C in air

Reduced at 900°C in H₂/Ar









Microstructure



Similar pore morphology

Sintering behavior



Lowered shrinkage: 13% (Reference 18%)



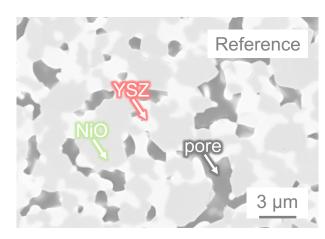
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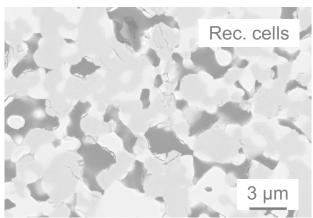


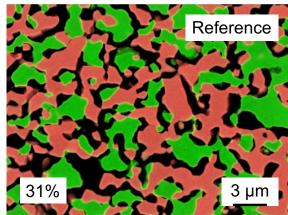


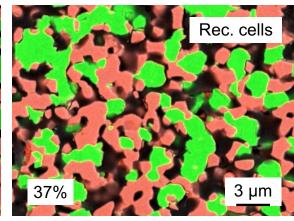
Sintered at 1400°C in air

Reduced at 900°C in H₂/Ar









Microstructure



Similar pore morphology

Microstructure



Higher porosity
Ni grains more spherical

Sintering behavior



Lowered shrinkage: 13% (Reference 18%)

Performance



Percolating Ni expected Reduced mech. stability



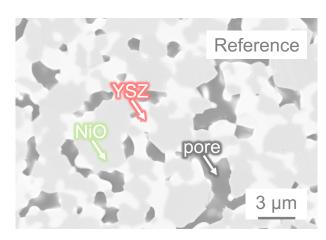
Affecting further cell processing route

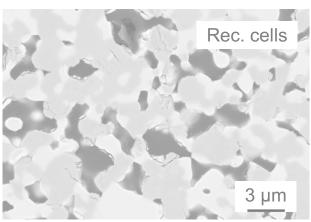
Green = Ni Red = Zr

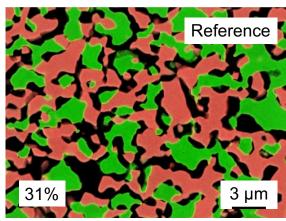


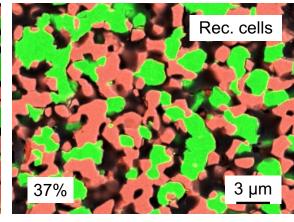
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Microstructure



Similar pore morphology

Microstructure



Higher porosity
Ni grains more spherical

Sintering behavior



Lowered shrinkage: 13% (Reference 18%)

Performance



Percolating Ni expected Reduced mech. stability

Affecting further cell processing route

100

100% recycling challenging

Green = Ni Red = Zr



Recycling of substrate green tapes

Re-suspending green tapes to slurry



Nearly no influence on slurry properties

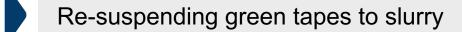


Full recovery of substrate in process





Recycling of substrate green tapes





Nearly no influence on slurry properties



Full recovery of substrate in process

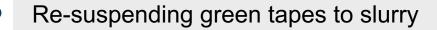


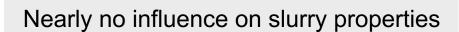
Promising to be implemented into manufacturing of SOCs





Recycling of substrate green tapes







Promising to be implemented into manufacturing of SOCs



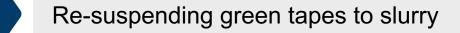
Recycling of sintered half cell material

Processing slurry from re-milled half cells





Recycling of substrate green tapes



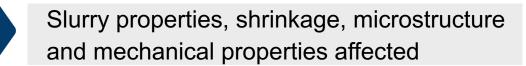




Promising to be implemented into manufacturing of SOCs

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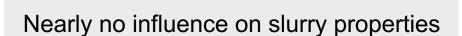
Full recovery of sintered cells difficult





Recycling of substrate green tapes





Full recovery of substrate in process

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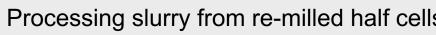
Slurry properties, shrinkage, microstructure and mechanical properties affected

Full recovery of sintered cells difficult

Recycling of defined shares realistic Adaptability for End-of-Life material?













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