## Structural study of the Wadsley-type vanadium oxides V<sub>2</sub>O<sub>5</sub> and V<sub>6</sub>O<sub>13</sub> at extreme conditions

**B. Viliam Hakala**<sup>1,2</sup>, Dimitra K. Manousou<sup>1,3</sup>, Karen Friese<sup>1,2</sup>, Andrzej Grzechnik<sup>1,2</sup>

The so-called Wadsley phases with the general formula  $V_nO_{2n+1}$  form a homologous series of compounds [1]–[3]. They have arisen much interest due to the observed metal-insulator transitions and their potential application as battery materials. [4]–[7] The crystal structures of the parent compound  $V_2O_5$  and of  $V_6O_{13}$  (n=3) at ambient conditions are closely related.

We have now studied single crystals of  $V_2O_5$  and  $V_6O_{13}$  as a function of pressure at Petra III, DESY. Our study [8] shows a complete irreversible amorphization of the  $\alpha$ - $V_2O_5$  sample above 7.3 GPa. Further investigation of the HP-HT behaviour of  $\alpha$ - $V_2O_5$  was performed at the large volume press at ID06 at the ESRF, where we followed the evolution of the sample with in-situ synchrotron radiation. Heating of the amorphous phase led to the formation of the  $\delta$ - $V_2O_5$  polymorph with  $Sb_2O_5$  structure, which can be recovered at ambient conditions. High-pressure single-crystal diffraction experiments  $\alpha$ - $V_6O_{13}$  show an anomalous behaviour between 2 and 3 GPa, yet the ambient pressure polymorph seems to be stable up to the highest pressures reached in the experiment.

Furthermore, we performed low-temperature single-crystal X-ray diffraction experiments on  $\alpha$ -V<sub>6</sub>O<sub>13</sub>. Our data allow for a conclusive determination of the crystal structure below the metal-to-insulator transition, especially as this structure has long been disputed [9], [10]. Additionally, we have detected a structural phase transition associated with antiferromagnetic (AFM) ordering at 55 K.

## References

- [1] U. Schwingenschlögl and V. Eyert, *Ann. der Phys.*, **13**, 9, 475–510, (2004)
- [2] R. Eguchi et al., Phys. Rev. B Condens. Matter Mater. Phys., 65, 20, 1–4, (2002)
- [3] H. Katzke et al., *Phys. Rev. B Condens. Matter Mater. Phys.*, **68**, 2, 1–7, (2003)
- [4] A. Bhatia, et al., *Chem. Mater.*, **34**, 3, 1203–1212, (2022)
- [5] T. Averianov and E. Pomerantseva, *J. Alloys Compd.*, **903**, 163929, (2022)
- [6] J. L. Andrews et al., *Chem*, 4, 3, 564–585, (2018)
- [7] P. He et al., J. Mater. Chem. A, 8, 20, 10370–10376, (2020)
- [8] B. V. Hakala et al., J. Alloys Compd., **911**, 164966, (2022)
- [9] I. Kawada et al., *Acta Crystallogr.*, **34**, 3, 1037–1039, (1978)
- [10] J. Höwing et al., Acta Crystallogr. Sect. B Struct. Sci., **59**, 6, 747–752, (2003)

E-mail of the corresponding author: b.hakala@fz-juelich.de

<sup>&</sup>lt;sup>1</sup> Jülich Centre for Neutron Science-2 and Peter Grünberg Institute-4 (JCNS-2/PGI-4), Forschungszentrum Jülich GmbH, 52425 Jülich, Germany

<sup>&</sup>lt;sup>2</sup> Institute of Crystallography, RWTH Aachen, Jägerstraße 17-19, 52066 Aachen, Germany

<sup>&</sup>lt;sup>3</sup> Section of Condensed Matter Physics, Department of Physics, National and Kapodistrian University of Athens, Panepistimioupolis 15784, Zografos, Athens, Greece