

Elucidation of Barocaloric Effect in Spin Crossover Compounds with Inelastic Scattering Methods

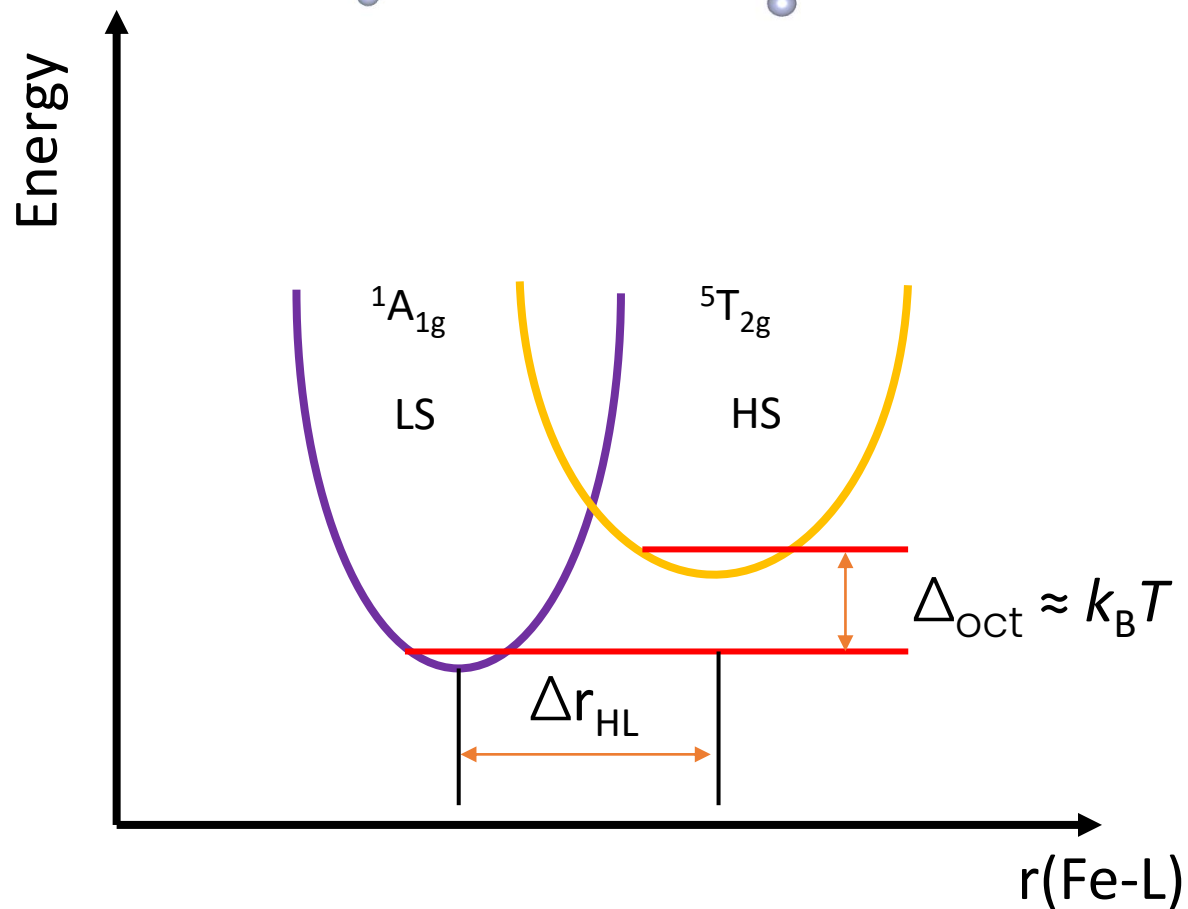
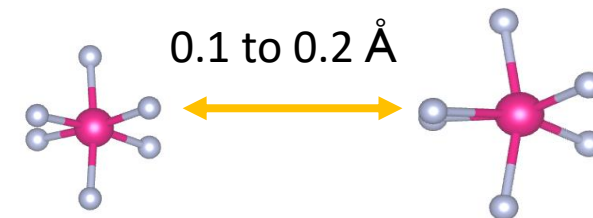
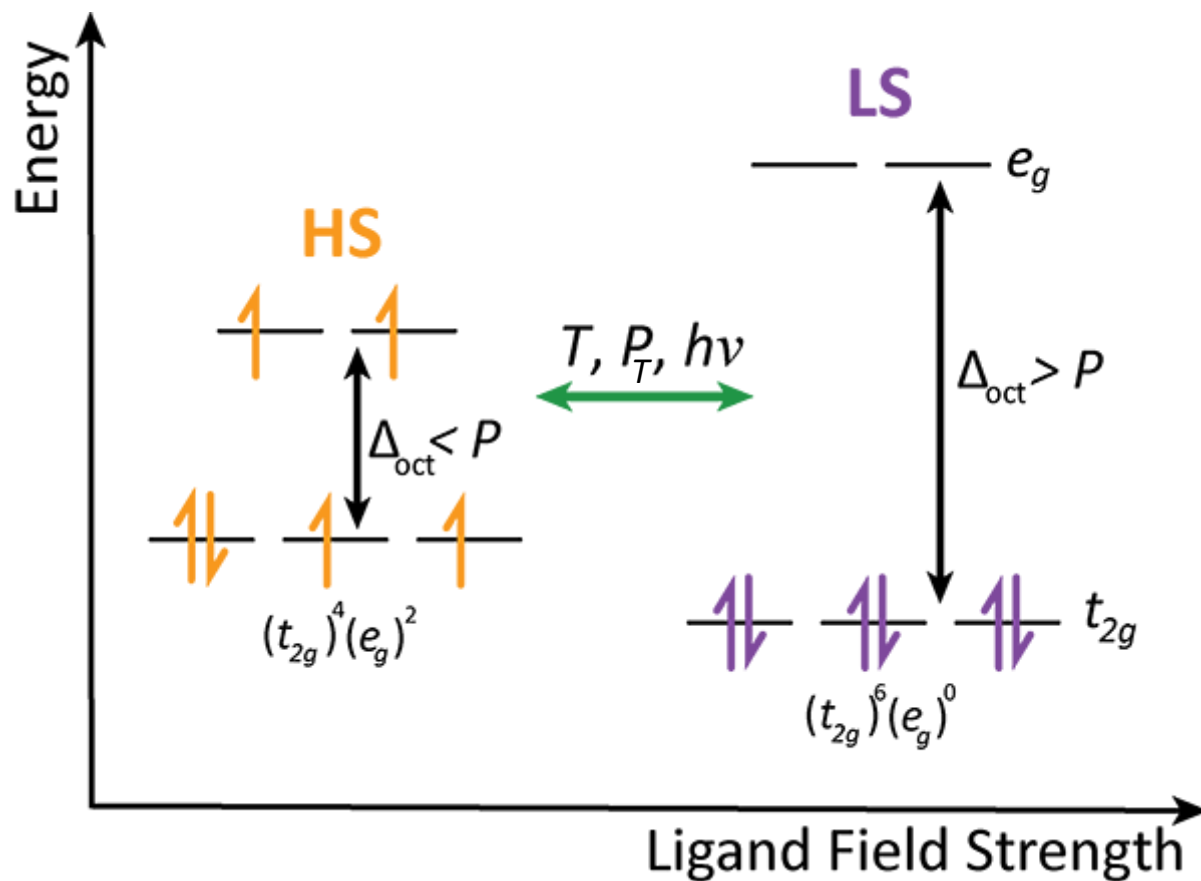
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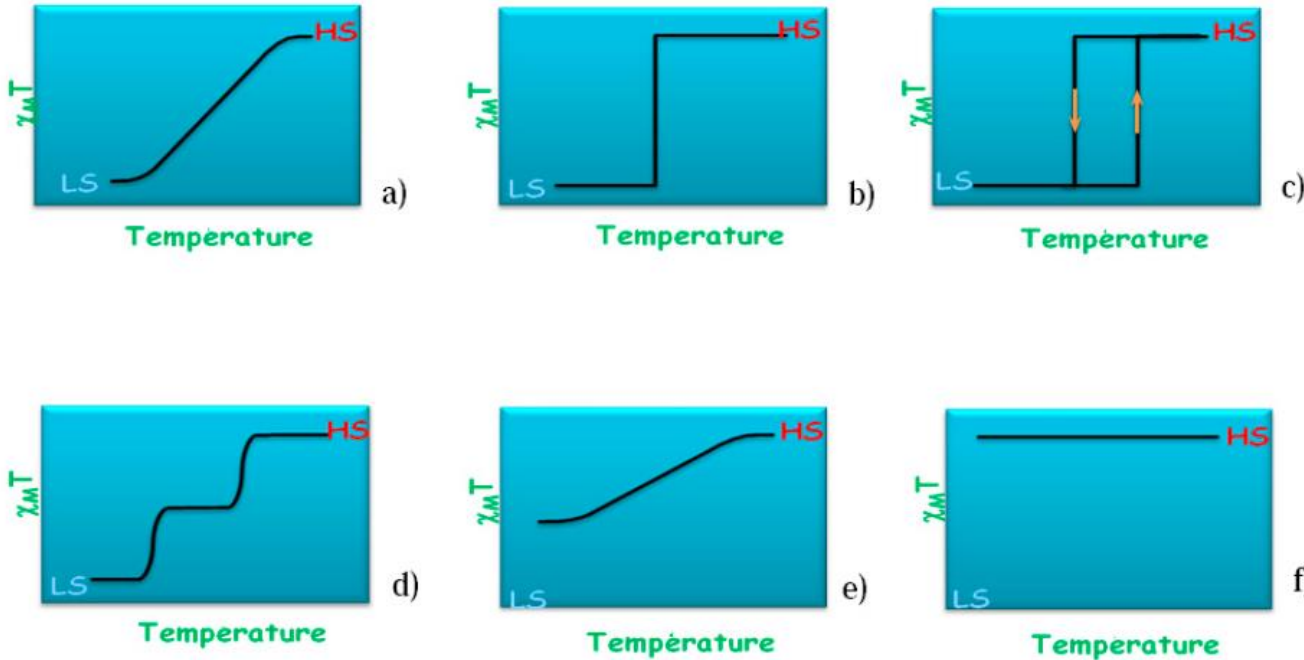
February 14th, 2024

Spin Crossover (SCO)

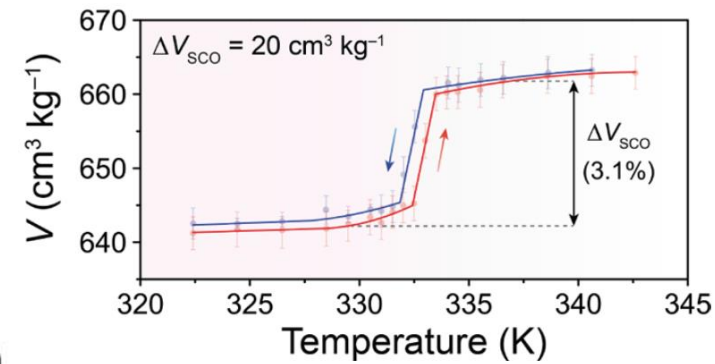
$3d^4$ $3d^5$ $3d^6$ $3d^7$
 $\text{Cr}^{2+}, \text{Mn}^{3+}, \text{Mn}^{2+}, \text{Fe}^{3+}, \text{Fe}^{2+}, \text{Co}^{3+}, \text{Co}^{2+}, \text{Ni}^{3+}$



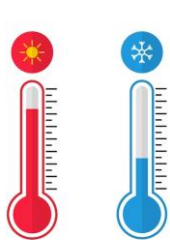
Properties



- Large thermal effect
- Color change
- Large Volume Change
- ...



Available external stimuli



temperature

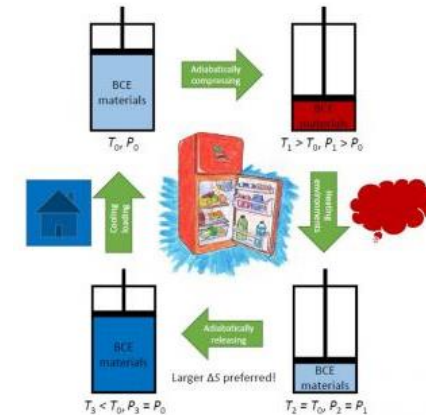
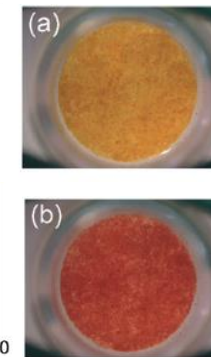
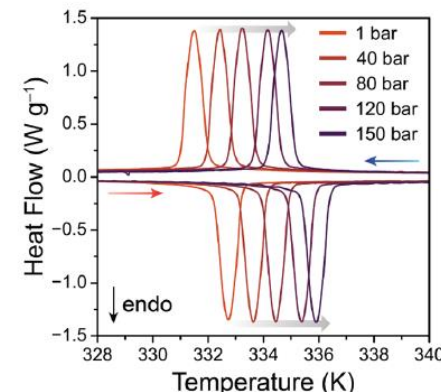


light



pressure

Others...

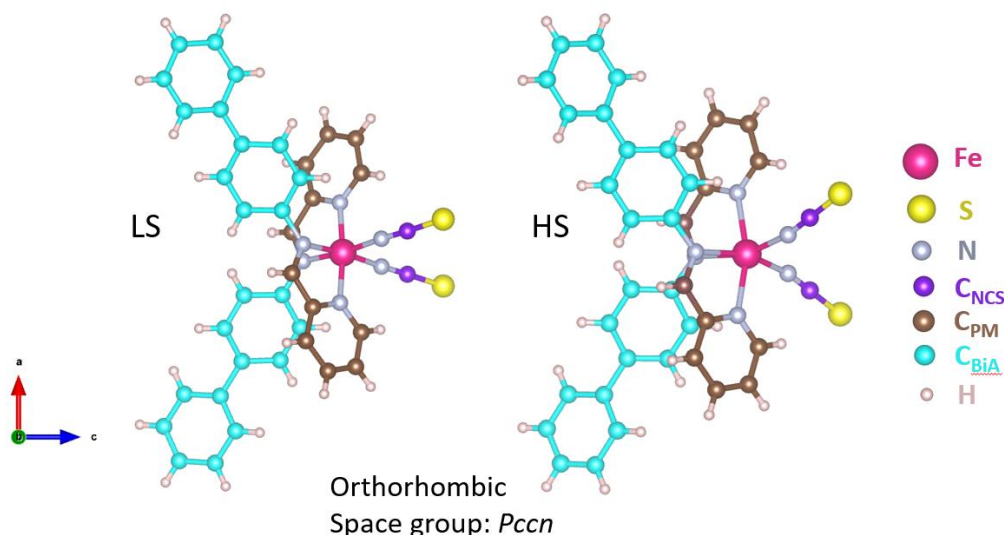


Jinyoung Seo et al. *J. Am. Chem. Soc.* **144**, 6493 (2022).

Samir F. Matar et al. *Int. J. Mol. Sci.* **16**, 4007 (2015).

Fe(PM-BiA)₂(NCS)₂

Compound	T_{SCO} (K)	$ \Delta S $ (J/K mol)	$ \Delta S $ (J/K kg)	$\partial T_{SCO} / \partial p$ (K/GPa)	Maximum ΔT_{ad}	Pressure required (GPa)
[Fe(PM-BiA) ₂ (NCS) ₂]	170	58	84	66	8.4	0.12
[Fe(phen) ₂ (NCS) ₂]	180	49	92	220	8.3	0.04
{Fe[H ₂ B(pz) ₂] ₂ (bipy)}	160	48	95	188	7.6	0.04
{Fe(pmd) ₂ [Cu(CN) ₂] ₂ }	140	36	80.5	380 (avg.)	5.6	0.013
[Fe(pmea)(NCS) ₂]	184	60 (calc.)	44	146	4	0.03



Monoclinic

$P2_1/c$

$T_{1/2} = 208$ K

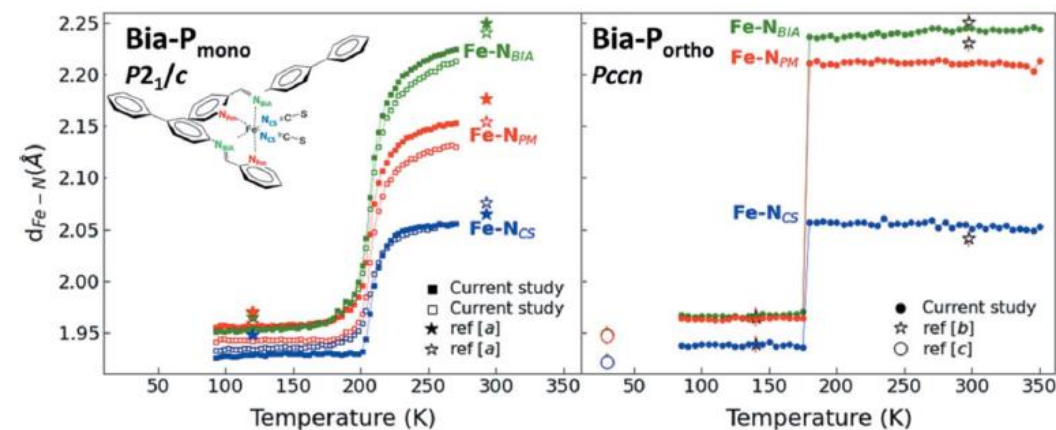
$T_{hys} = 1$ K

Orthorhombic

Pccn

$T_{1/2} = 177$ K

$T_{hys} = 5$ K



Entropy

$$\Delta S = \Delta S_{\text{el}} + \Delta S_{\text{vib}} + \Delta S_{\text{conf}} + \Delta S_{\text{rot}}$$

$$\Delta S_{\text{el}} = R \ln \left(\frac{2S_{\text{HS}}+1}{2S_{\text{LS}}+1} \right)$$

$$= 13.5 \text{ J mol}^{-1} \text{ K}^{-1}$$

$$\Delta S_{\text{vib}} = R \sum_{\lambda} \ln \left[\frac{v_{\lambda}^{\text{LS}}}{v_{\lambda}^{\text{HS}}} \right] =$$

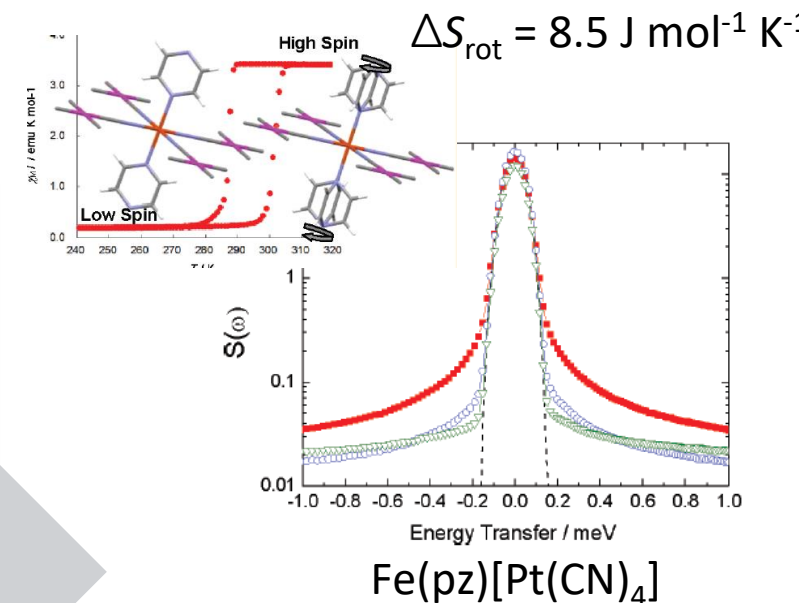
$$30\% \sim 70\% \Delta S_{\text{tot}}$$

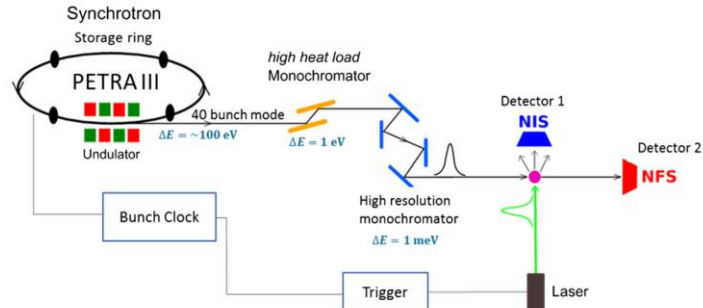
ΔS_{el}

ΔS_{rot}

ΔS_{vib}

ΔS_{conf}





Partial Density of States (P01@Petra III)

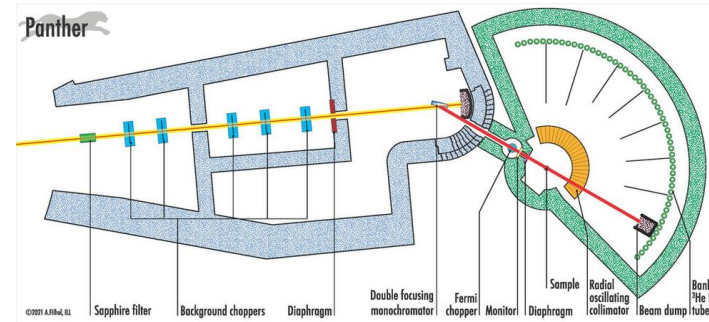
- Determine the partial density of states of Fe^{2+} by using NIS.
- Study the pressure effect on the partial DOS of Fe^{2+} .

Required sample

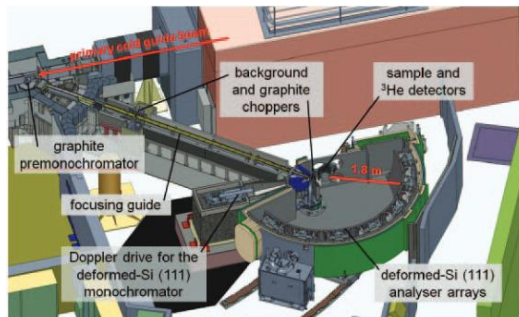
^{57}Fe SCO sample

Density of states (Panther@ ILL) (scheduled)

- Explore the lattice dynamics beyond the entire Brillouin zone.
- Deuterated sample will be used to avoid the incoherent contribution from H.



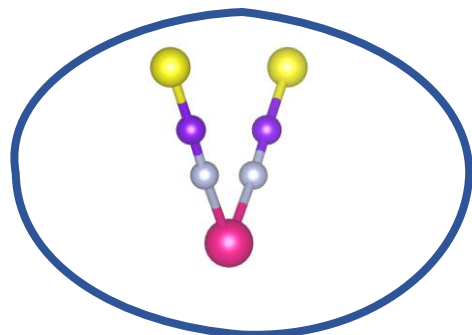
Protonated and deuterated complexes



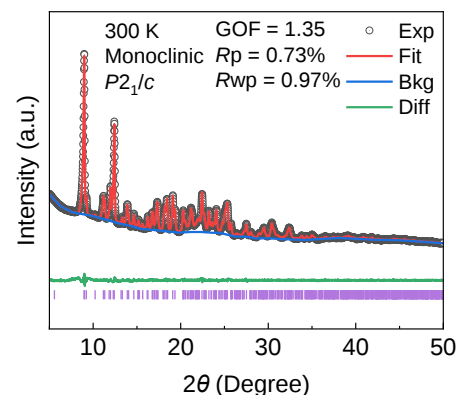
Local Dynamics (EMU and Pelican@ANSTO)

- Study the local dynamics at different timescale;
- Determine the geometry of the local motions;
- Extract the entropy contribution from local motions.

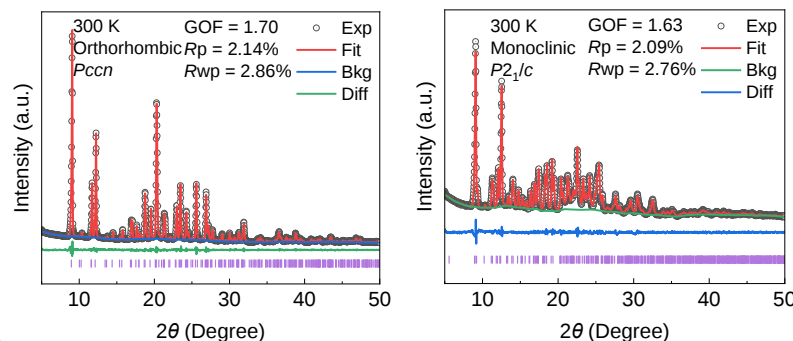
Protonated complexes



Substitute the FeSO_4 by $^{57}\text{FeSO}_4$

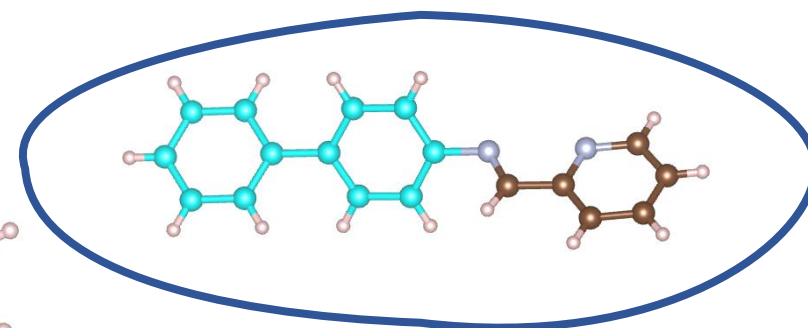
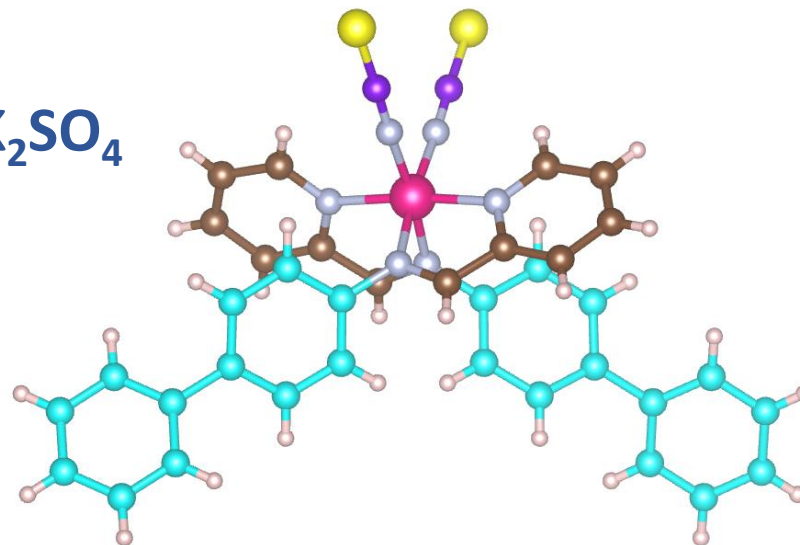
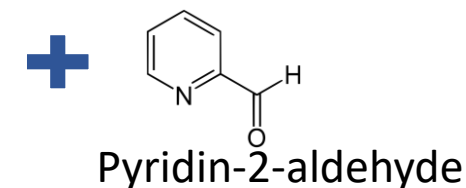
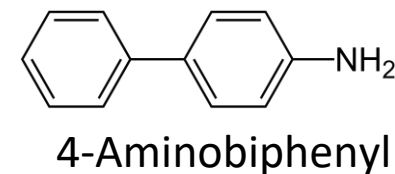


Protonated sample



Deuterated Sample

- ☐ More than **95%** deuteration level has been achieved on **4-Aminobiphenyl**.
- ☐ The deuteration for Pyridin-2-aldehyde is still ongoing

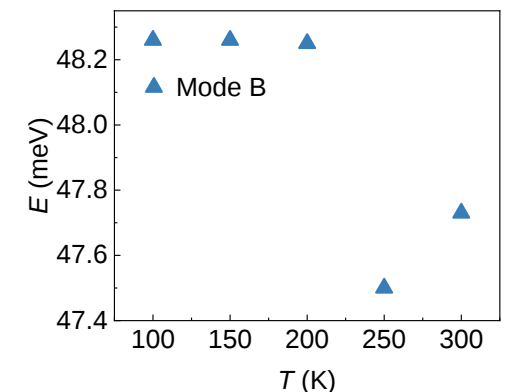
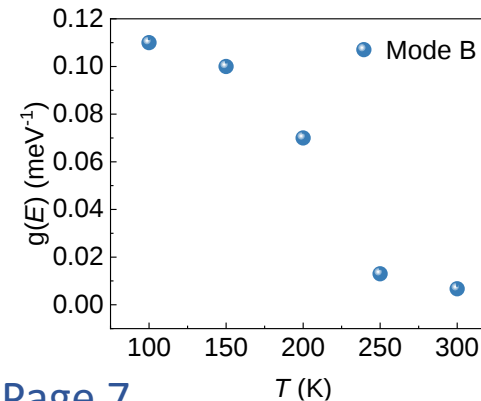
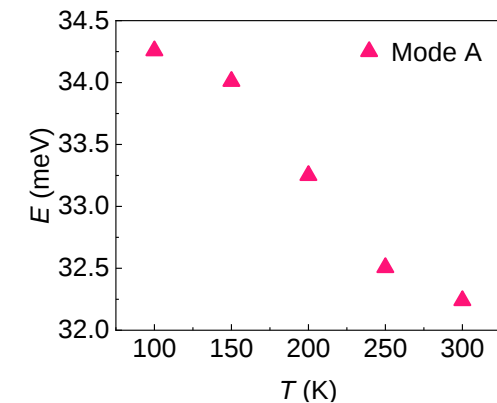
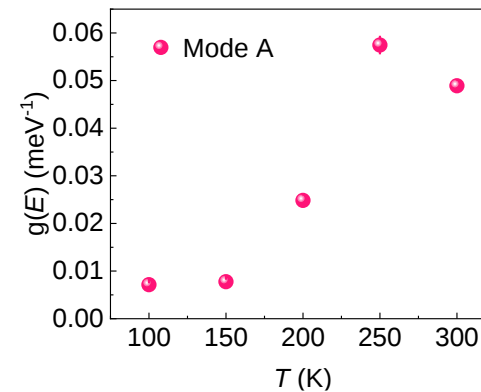
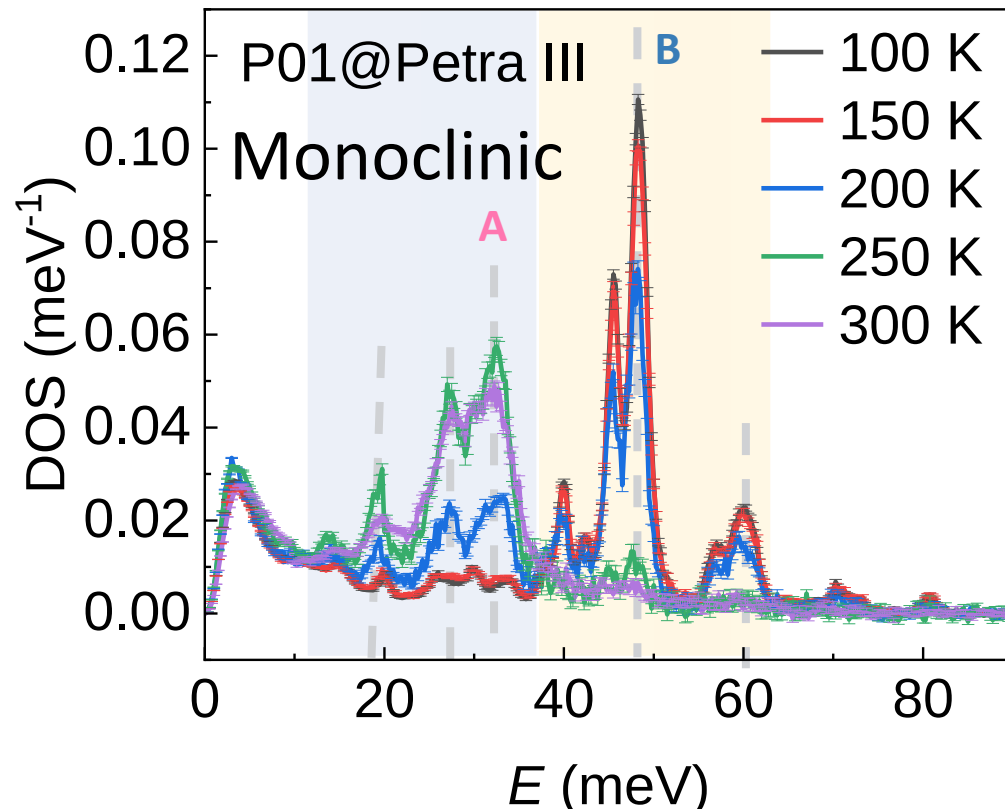


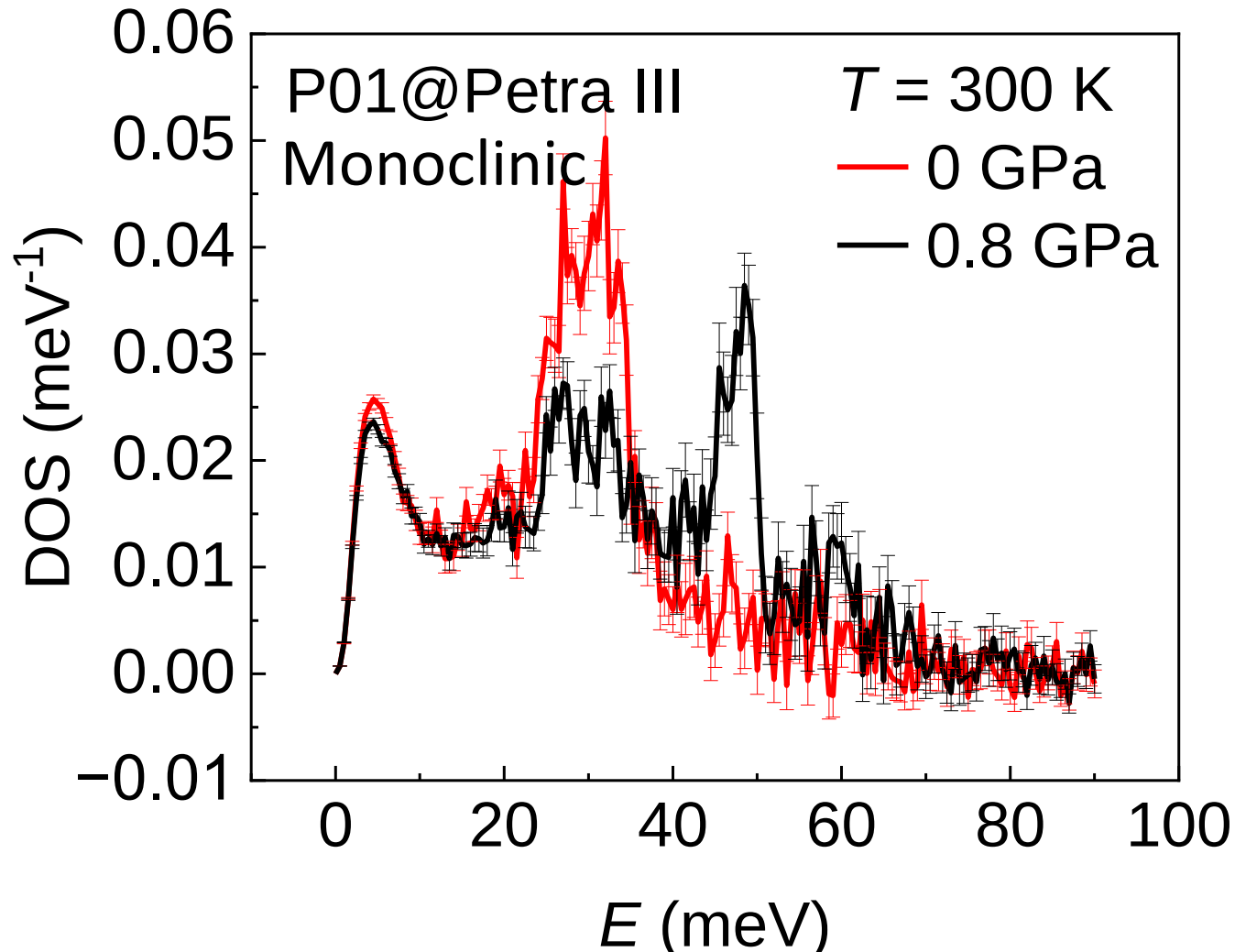
- Temperature induced **Fe-contributed phonon modes** correspond to the **gradual SCO transition** of the monoclinic sample.

$$S_1 - S_2 = 3N_A k_B \int_0^\infty (g_1(E) - g_2(E)) \ln(E) dE.$$

$$\Delta S_{T_0 \rightarrow T} = 12.72 \text{ J mol}^{-1} \text{ K}^{-1}$$

30.2% of ΔS_{tot}

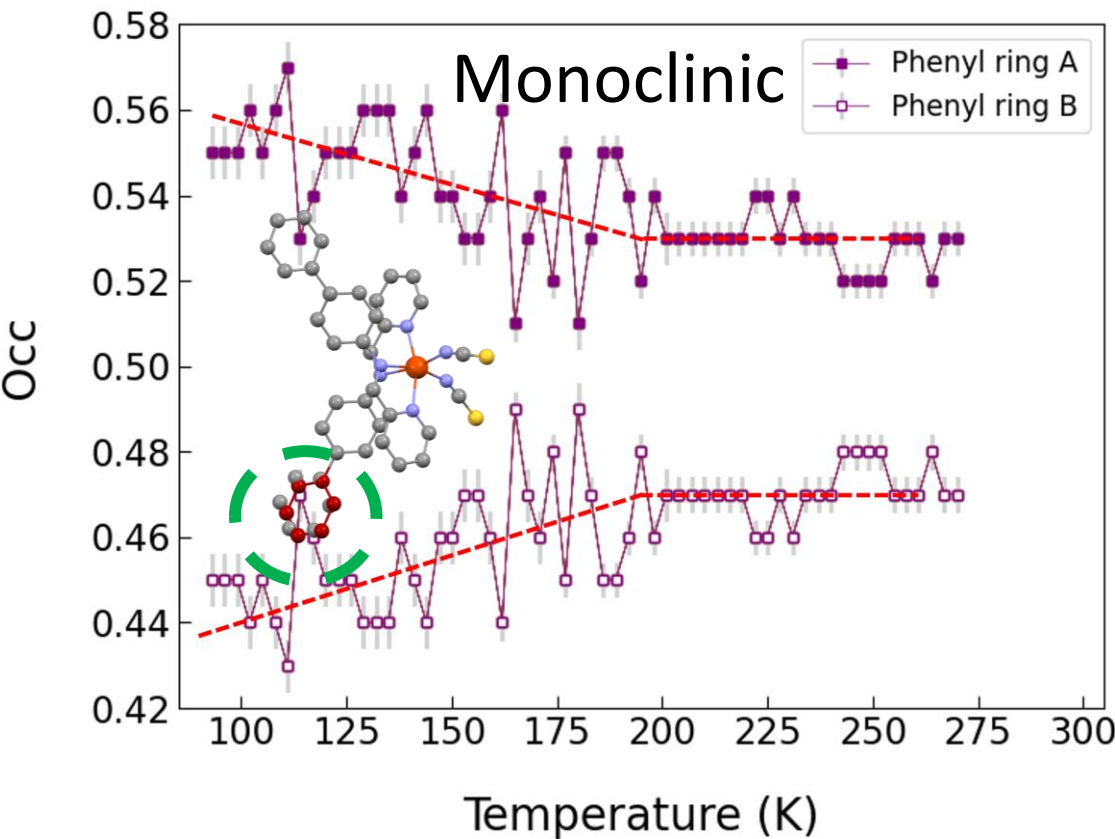




- The pressure triggered SCO transition can be observed under 0.8 GPa at 300 K.
- 0.8 GPa is not sufficient to completely suppress the HS at 300 K

$$\Delta S_{P_0 \rightarrow P} = 8.48 \text{ J mol}^{-1} \text{ K}^{-1}$$

- The temperature dependent occupation of the phenyl ring indicates the dynamical disorder in the system.



Quasi-elastic neutron scattering is the solution.

$$S(Q, \omega) = S_{\text{coh}}(Q, \omega) + S_{\text{inc}}(Q, \omega)$$

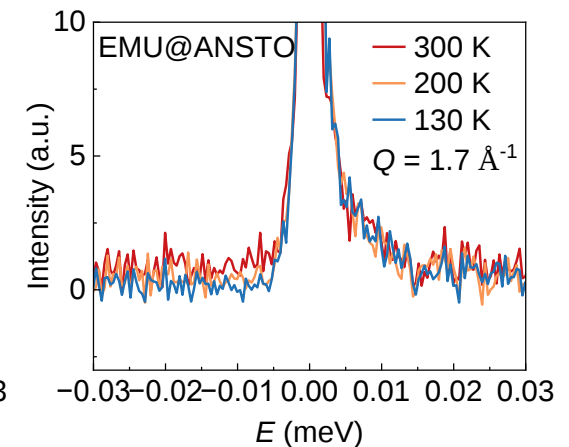
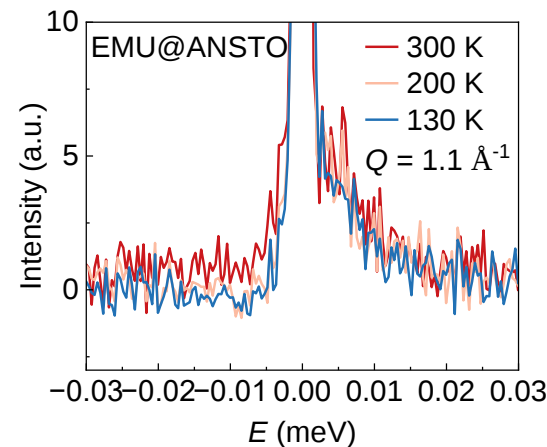
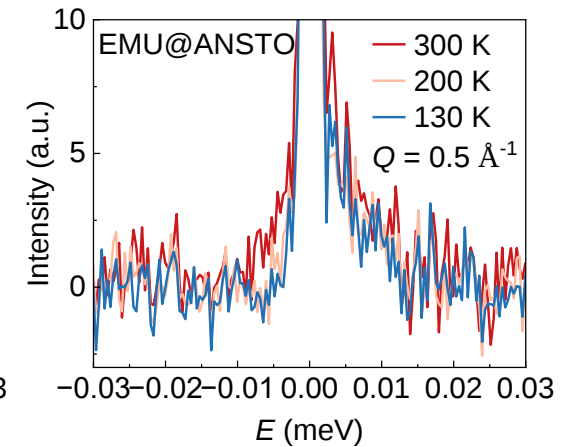
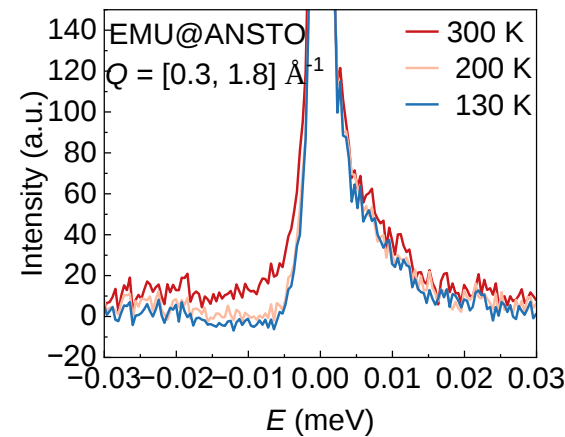
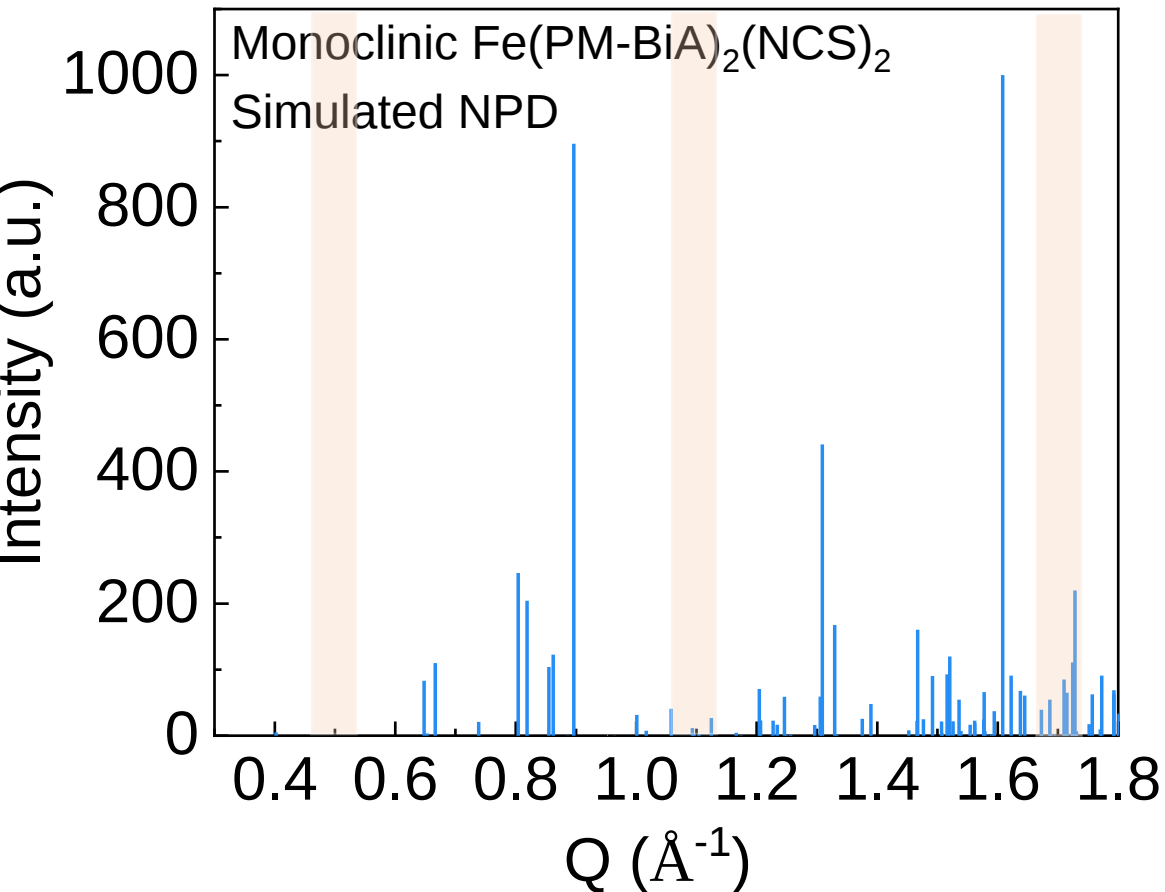
- Geometry of motion
- Time scale of diffusion

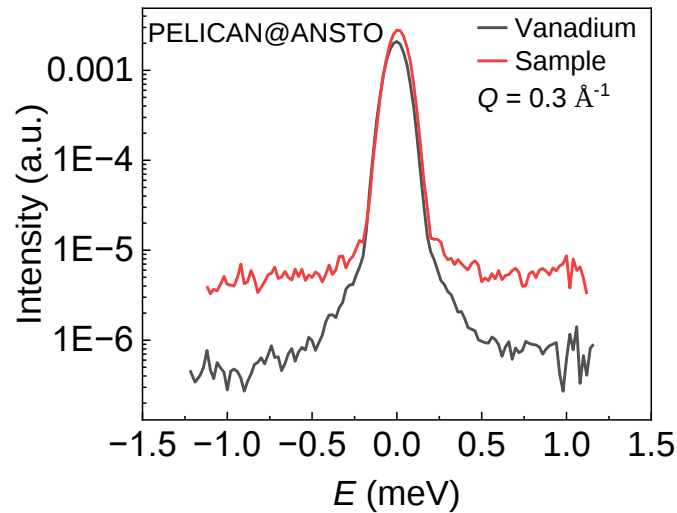
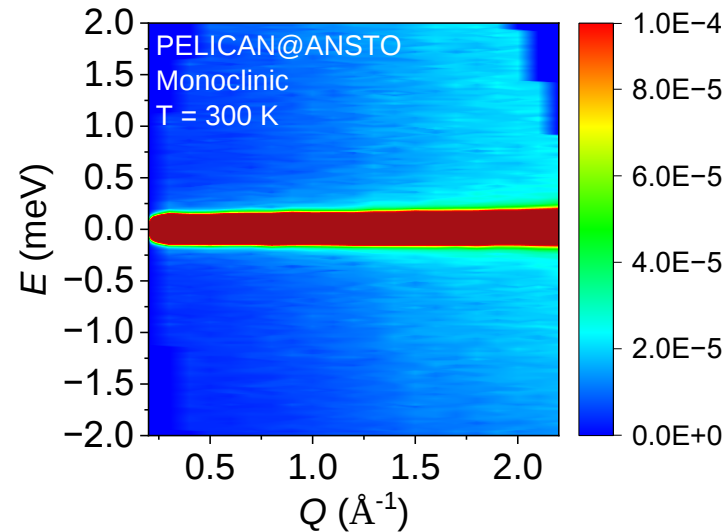
$$S(Q, \omega) = S_{\text{inc}}(Q, \omega) \otimes R(Q, \omega) \rightarrow \text{Longest time scale}$$

Elastic incoherent scattering factor $\rightarrow EISF = \frac{I_{\text{elastic}}}{I_{\text{elastic}} + I_{\text{inelastic}}}$

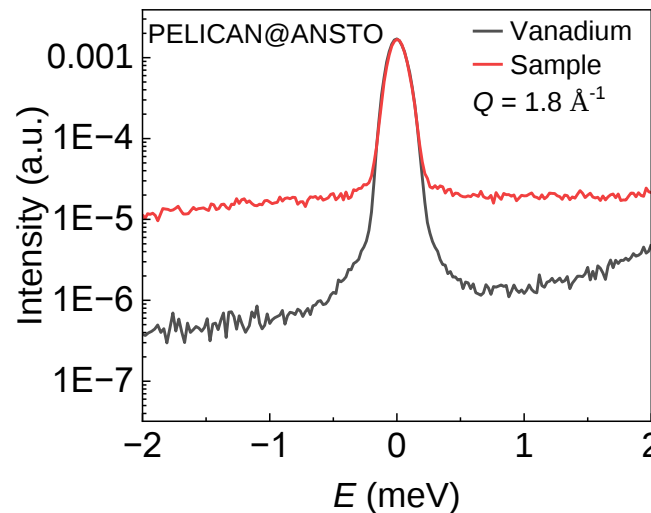
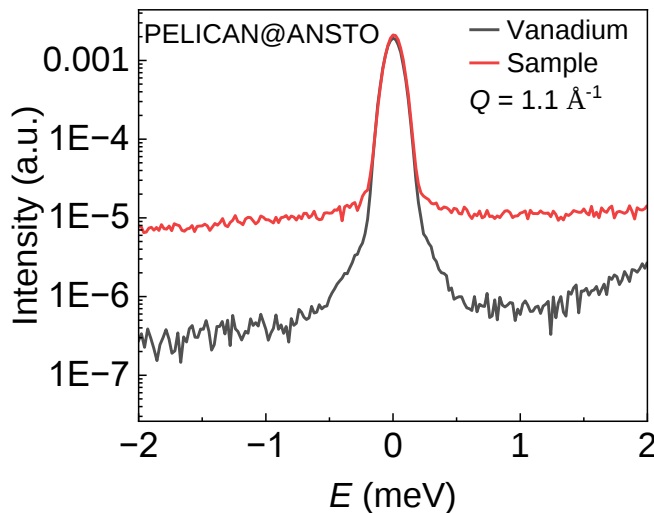
Location	Instrument	Energy resolution (μeV)	Energy range (meV)	Q-range (Å ⁻¹)
ANSTO	EMU	1.1	-0.028 to 0.02	0.4-1.8
ISIS	IRIS	17.5	-0.55 to 0.57	0.6-1.6
ANSTO	Pelican	65	-19.99 to 2.59	0.2-1.8

- The neutron backscattering indicates that there is **no local mode at nanosecond scale**.





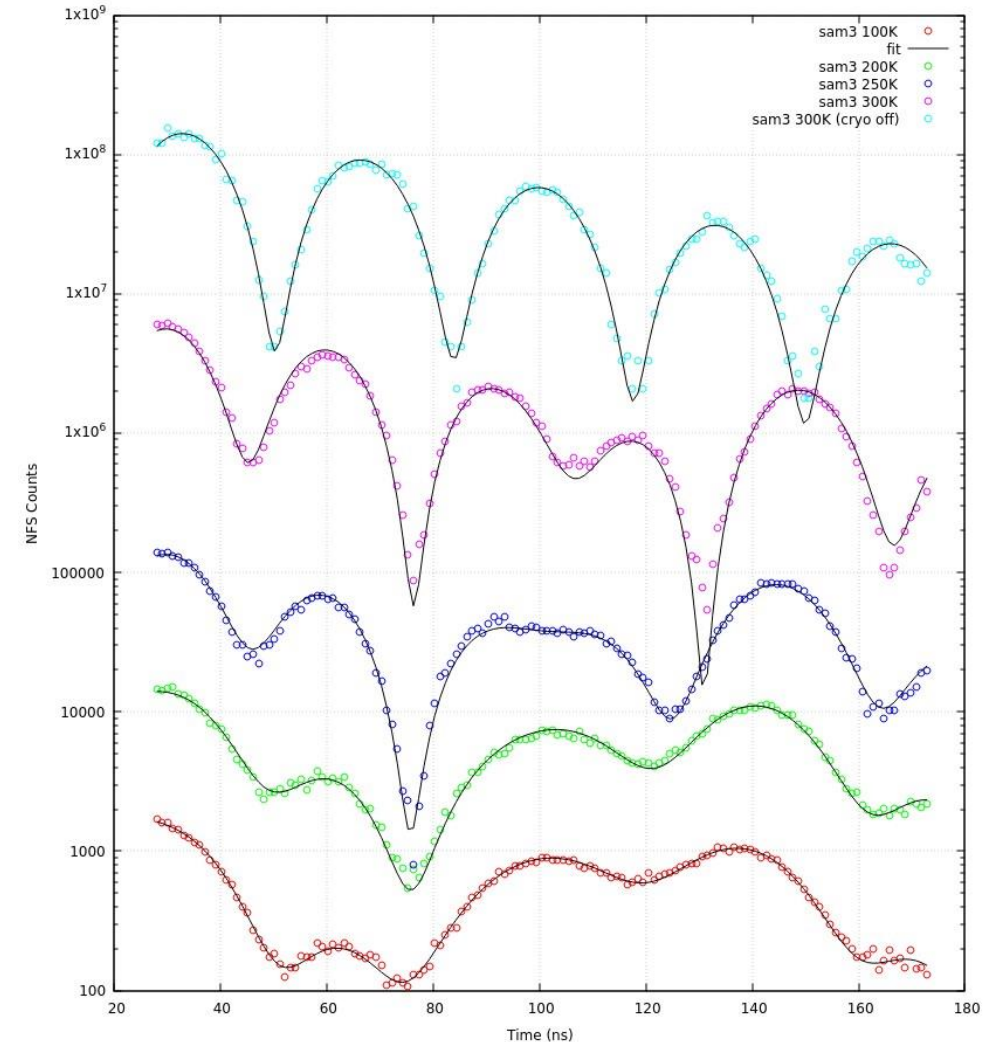
□ An apparent Q dependence feature of the $S(Q, E)$ indicates the localized modes in this system.



□ The wide QENS signal indicates an extremely fast local mode at picosecond scale.

1. Protonated sample and ^{57}Fe SCO compounds are ready for further experiments.
2. The **Fe-contributed entropy change** ($12.72 \text{ J mol}^{-1} \text{ K}^{-1}$) is extracted from the DOS, which is **30.2% of the total entropy change** ($42 \text{ J mol}^{-1} \text{ K}^{-1}$).
3. The INS experiment at Pelican confirmed the existence of fast localized modes in the monoclinic compound.

- ☐ Deuteration.
- ☐ INS at Panther.
- ☐ Detailed local motion.
- ☐ Polarized neutron.



Acknowledgement



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Thanks for your attention!