

1 **Evaluation of Anthropogenic Influences on Atmospheric**
2 **Oxygenated Organic Molecules in Both Gas and Particle**
3 **Phases over the Southeastern Tibetan Plateau**

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18 **Abstract:**

19 The Tibetan Plateau (TP), as the region with the highest altitude in the world, plays
20 an important role in regulating climate change, and is highly sensitive to
21 anthropogenic pollutants. To assess the impact of anthropogenic emissions on
22 atmospheric oxygenated organic molecules (OOMs) and regional air quality in the
23 southeastern TP, we conducted intensive field observations using iodide Chemical
24 Ionization Mass Spectrometry combined with a Filter Inlet for Gases and AEROsols
25 (FIGAERO-CIMS) during the @Tibet field campaigns. We detected 653 distinct
26 OOMs in both gas and particle phases, with most falling within the intermediate
27 volatility range. Supervised machine learning was used to classify OOMs based on
28 precursor origin, including biogenic (isoprene and monoterpenes) and anthropogenic
29 (aromatics and aliphatics) sources. Notably, OOMs derived from anthropogenic
30 emissions dominated the measured compounds, representing 32.5% in the gas phase
31 and 45.6% in the particle phase, substantially exceeding the contributions from
32 biogenic sources (27.5% and 21.8% in the gas and particle phases, respectively).
33 These results reveal the significant influence of anthropogenic emissions on
34 atmospheric species even in remote areas, highlighting the urgent need to consider
35 these impacts in future air quality assessments and pollution mitigation strategies in
36 the TP region.

37

38 **Keywords:**

39 Tibetan Plateau (TP); Chemical Ionization Mass Spectrometry with a Filter Inlet for

40 Gases and AEROSols (FIGAERO-CIMS); Anthropogenic Emissions; Oxygenated
41 Organic Molecules (OOMs); Machine Learning Approach

42

43 **Synopsis:**

44 Our work enhances our understanding of atmospheric pollutants and highlights the
45 importance of anthropogenic sources to OOMs formation in the southeastern TP.

46

47

48 **1. INTRODUCTION**

49 Organic aerosols (OA) constitute a significant fraction of fine particulate matter
50 globally,¹ adversely affecting air quality,² climate,³ and human health.⁴ The
51 complexity of OA, especially secondary organic aerosols (SOA), complicates our
52 understanding of their chemical compositions, sources, and formation,⁵⁻⁷ resulting in
53 substantial discrepancies between modeled and measured SOA concentrations.^{8, 9}

54 Oxygenated organic molecules (OOMs) are critical intermediates during SOA
55 formation, produced by the oxidation of gaseous precursors such as volatility organic
56 compounds (VOCs).^{5, 10} Therefore, the chemical characterization of OOMs is
57 essential to enhance our understanding of the sources, formation pathways, and
58 properties of SOA.^{11, 12} However, the identification of atmospheric OOMs at
59 molecular level remains limited, hindering the development of more accurate
60 atmospheric models.

61 The Tibetan Plateau (TP), often referred to as the “Third Pole”, plays a pivotal
62 role in atmospheric circulation and global climate,¹³ while providing substantial water
63 resources to downstream areas.¹⁴ Due to its sparse population and minimal industrial
64 activities, the atmosphere of the TP largely reflects global background conditions.¹⁵
65 Therefore, climate and cryosphere changes in the TP are particularly sensitive to
66 atmospheric pollutants.¹³ As such, it is imperative to assess the extent to which
67 anthropogenic sources impact the TP’s atmosphere. Previous studies using aerosol
68 mass spectrometer (AMS) measurements have demonstrated that secondary formation
69 is a major contributor to OA in the TP.¹⁶⁻¹⁸ However, the specific influence of

anthropogenic sources remains poorly constrained. Shen et al.¹⁹ detected SOA tracers originating from different precursors through offline measurements and found that aromatics may play an important role in SOA formation during certain seasons in the TP. Using a similar approach, Wan et al.²⁰ identified OA tracers from biomass burning and biogenic sources, while a large unexplained fraction of OA was attributed to other sources, such as anthropogenic emissions. Although offline measurement techniques are effective in identifying SOA tracers from different precursors, the number of detectable compounds is limited, leading to substantial uncertainty into the assessment of anthropogenic contributions. Nitrate chemical ionization mass spectrometry (CIMS) has been employed to measure gas-phase OOMs. Bianchi et al.²¹ and Liu et al.²² revealed the substantial role of biogenic precursors, especially monoterpenes, in SOA formation in the TP. However, nitrate CIMS is more sensitive to highly oxygenated compounds, and molecules with lower oxygen content are often underestimated or even entirely missed.²³ Furthermore, few studies have simultaneously investigated OOMs in both gas and particle phases, which is essential for capturing the full scope of atmospheric oxidation processes and sources contributions. Therefore, a comprehensive chemical characterization of OOMs in both phases is necessary to better quantify the role of anthropogenic sources in shaping the OOMs formation in the TP.

To gain insight into atmospheric gaseous and particulate OOMs in the southeastern TP, we deployed an iodide Chemical Ionization Mass Spectrometry with a Filter Inlet for Gases and AEROsols (FIGAERO-CIMS) at the Lulang observation site, as part of

92 the @Tibet field campaigns 2021. Iodide CIMS is a powerful tool for measuring
93 OOMs in the atmosphere,²⁵ exhibiting high selectivity towards highly polarizable
94 compounds.²⁶ And FIGAERO enables simultaneous gas and particle measurements.²⁷
95 Subsequently, we applied a machine learning approach to identify the precursors of
96 the measured OOMs, and elucidated the impacts of anthropogenic sources on gaseous
97 and particulate OOMs in the southeastern TP.

98

99 **2. MATERIALS AND METHODS**

100 **2.1 Measurement Site.**

101 The measurements were conducted at the Lulang observation site (94°44'E,
102 29°46'N, 3326 m a.s.l.) from April 7 to 24, 2021. Situated in a mountainous valley in
103 the southeastern TP, on the northern bank of the Yarlung Zangbo River,²² Lulang is
104 characterized by diverse vegetation, including high-altitude forests.²⁸ A village
105 (Zhaxigang) and a town (Lulang) are located approximately 1.7 km and 4.5 km to the
106 south, respectively. National Road 318 runs east of the site. The site was considered
107 not strongly affected by high-intensity on-road emissions due to few tourist vehicles
108 during the observation period.²⁹ Overall, it is an ideal location for investigating the
109 impact of anthropogenic sources on the background environments of the TP. Detailed
110 descriptions of this site can be found in previous studies.^{22, 28}

111 **2.2 Instrumentation.**

112 An iodide FIGAERO-CIMS (Aerodyne Research Inc.) was deployed to
113 simultaneously measure gaseous and particulate OOMs. The FIGAERO has three

114 inlet ports that operates in two modes.²⁷ During the gas mode, ambient air was drawn
115 into the ion molecule reactor (IMR) at a rate of 2 L min^{-1} , where the molecules reacted
116 with iodide ions to form product ions. The pressure and temperature in the IMR were
117 maintained at 200 mbar and 60 °C, respectively. Concurrently, ambient particles were
118 collected on a PTFE filter (1.0 μm pore size Zefon®, Zefon International) on the
119 FIGAERO at a rate of 6 L min^{-1} to ensure sufficient particle loading (up to 3.6 μg ,
120 based on $\text{PM}_{2.5}$ measurements).³⁰ After 20 min in the gas mode, the FIGAERO
121 switched to the particle mode, where particles collected on the filter were thermally
122 desorbed using heated ultrahigh-purity (UHP) N_2 at a rate of 2 L min^{-1} . Then the
123 desorbed vapors were immediately drawn into the IMR for detection. The filter
124 temperature followed a three-step process: ramping from ambient temperature to 180 °C
125 over 20 min, soaking at 180 °C for 8 min, and cooling back to ambient temperature
126 over 12 min. A complete cycle, including both gas and particle modes, took 1 h.

127 To quantify the measured OOMs, we calibrated the FIGAERO-CIMS using 9
128 organic standards. The sensitivity of levoglucosan was applied to all species as the
129 maximum sensitivity of CIMS,²⁶ thus the reported concentrations represented the
130 lower limits in this study. Considering the lowest sensitivity (Table S1), the
131 concentration of certain OOMs may be underestimated by up to two orders of
132 magnitude. The potential impact of relative humidity on sensitivity was not
133 considered in this study. As the increase of water content may variably affect the
134 sensitivity of OOMs,^{25, 31} this simplification could introduce systematic uncertainty
135 into the results. The backgrounds of gas and particle were conducted throughout the

136 measurement period. Peak fitting of the data was processed using the Tofware
137 software (v3.2.2; Tofwerk AG, Switzerland). More details about the FIGAERO-CIMS
138 can be found in Text S1.

139 **2.3 Assigning Precursors to Oxygenated Organic Molecules Using a Machine**
140 **Learning Approach**

141 A detailed workflow to identify the precursors of the measured OOMs is illustrated
142 in Figure S1. OOMs measured by FIGAERO-CIMS originate from both direct
143 emissions and the oxidation of precursors.³¹ To assign possible precursors to the
144 measured OOMs, it is necessary to exclude primary OOMs firstly. Biomass burning
145 has been identified as a significant source in the TP.^{19, 28} We excluded OOMs likely
146 from biomass burning (BB-OOMs) based on their correlations with C₆H₁₀O₅, an
147 important marker for biomass burning.³² Except for BB-OOMs, most measured OOMs
148 were probably produced through photochemical oxidation, indicated by their higher
149 concentrations during the period from 11:00 to 16:00 (Figure S2), when short-wave
150 radiation was highest.²² Gas-phase OOMs displayed higher concentration
151 enhancement than particle-phase OOMs, probably due to the mass transfer limitations
152 of gas-particle partitioning.³³ Subsequently, we omitted OOMs with a carbon number
153 ≤ 3 (Small OOMs), as these small molecules could derive from the oxidation of
154 multiple precursors.^{34, 35}

155 A supervised machine learning approach was used to assign possible precursors to
156 the remaining OOMs based on the methodologies of Qiao et al.³⁶ and Wang et al.³⁴ A
157 random forest model was established for classification, requiring a pre-labeled

158 training dataset. Wang et al.³⁴ has created a dataset comprising known oxidation
159 products from four precursor classes. Since the dataset is concluded from a series of
160 laboratory and field studies using I-CIMS to measure the products of different
161 precursors, it can also be applied in our study. The dataset consists of 82, 126, 215,
162 and 118 products from isoprene, monoterpane, aliphatic, and aromatic, respectively,
163 effectively capturing the characteristics of different precursor classes.³⁴ The direct
164 application of a multi-class classification model may introduce considerable
165 uncertainty due to the presence of OOMs that originate from multiple precursors. To
166 overcome this issue, we reformulated the original four-class classification task into
167 four independent binary classification tasks. In each task, a separate random forest
168 model was trained to predict whether a given OOM belongs to the corresponding
169 precursor class or not. This approach allows for overlapping classifications and
170 improves the robustness of precursor attribution. Model input features included the
171 number of carbon, hydrogen, oxygen, and nitrogen atoms (n_C , n_H , n_O , and n_N ,
172 respectively), as well as calculated parameters such as double bond equivalent (DBE),
173 hydrogen-to-carbon ratio (H/C), oxygen-to-carbon ratio (O/C), and average carbon
174 oxidation state (\overline{OS}_C). The random forest model was implemented using the sklearn
175 package (v1.3.0) in Python, producing the occurrence probability for each OOM
176 across the four precursor classes. We replaced the probability ≤ 0.6 with 0, proposing
177 that the OOM unlikely came from the corresponding precursor. Then the signal of
178 each OOM was assigned to four precursor classes based on the ratio of occurrence
179 probabilities. If no probability was > 0.6 , the OOM was labeled as “Other OOMs”.

180 More details about the probability calculations can be found in Text S2.

181 To train and evaluate the random forest model, the dataset was divided into two
182 subsets randomly: 70% for training and 30% for testing. We employed a grid search
183 method to determine optimal model parameters. Three parameters were examined in
184 our study, i.e. the number of trees (from 1 to 150), the maximum depth of each tree
185 (from 1 to 14), and the max features for the best split at each node of a decision tree
186 (from 1 to 8). Model performance was assessed using three metrics, i.e. *Precision*,
187 *Recall*, and *F1-score* (the calculation detailed in Text S3). The variation of the metrics
188 with three selected parameters was explored, as shown in Figure S3. The optimal
189 value was 21, 8, and 2 for the number of trees, the maximum depth of each tree, and
190 the max features, respectively. *Precision*, *Recall*, and *F1-score* of final random forest
191 model was calculated using these optimal values, as listed in Table S2. All *F1-scores*
192 were ≥ 0.75 , indicating high accuracy in predicting the precursors for OOMs. In
193 addition, a 5-fold cross-validation was conducted to prevent overfitting (Text S4 and
194 Figure S4), with the average *F1-score* of four precursor classes at each iteration $>$
195 0.75, thus demonstrating model reliability.

196

197 **3. RESULTS AND DISCUSSION**

198 **3.1 Characteristics of Oxygenated Organic Molecules in the Ambient Air of Tibet**

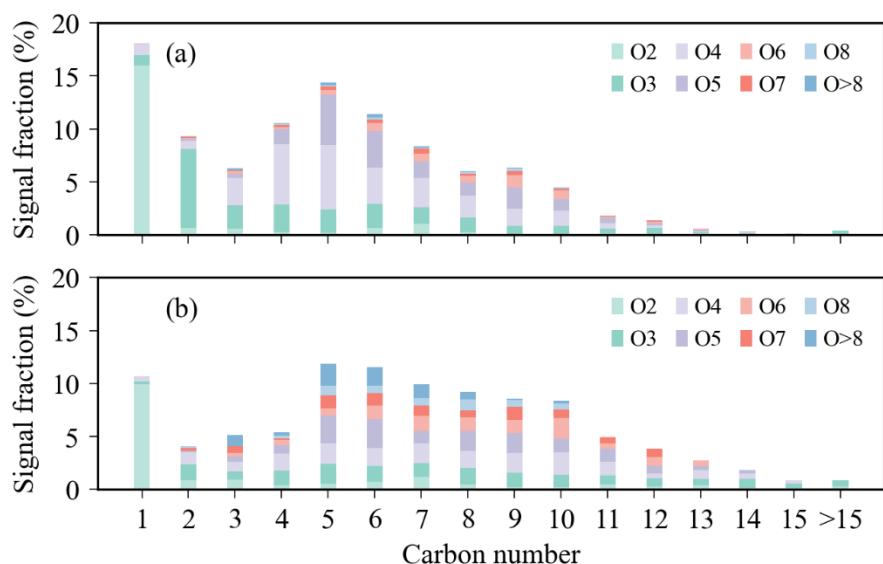
199 A total of 653 OOMs were identified within the m/z 150-400 Th in this study,
200 including 347 organic carbon species ($OC = C_xH_yO_z$, where $1 \leq x \leq 20$, y is an even
201 number not exceeding $2x+2$, and $z \geq 2$) and 306 organic nitrate species ($ON =$

202 $C_xH_yO_zN_{1\text{ or }2}$, when N_1 , $1 \leq x \leq 20$, y is an odd number not exceeding $2x+1$, and $z \geq 2$;
203 when N_2 , $1 \leq x \leq 20$, y is an even number not exceeding $2x$, and $z \geq 4$). The
204 campaign-averaged mass spectra of the measured gaseous and particulate OOMs are
205 shown in Figure S5. In general, larger OOMs exhibited a higher proportion in the
206 particle phase compared to the gas phase. The signal fractions decreased with the
207 increase of m/z when m/z was > 290 Th in the gas phase. The signal-weighted
208 average formulas of gaseous and particulate OOMs were $C_{5.1}H_{7.7}O_{3.8}N_{0.2}$ and
209 $C_{6.8}H_{9.3}O_{4.7}N_{0.3}$, corresponding to molecular weights of 132 and 171 g mol^{-1} ,
210 respectively. The oxygen number was higher, while the O/C was lower in the particle
211 phase than those in the gas phase, attributed to the larger fractions of more oxidized
212 small OOMs in the gas phase. It should be noted that several small OOMs exhibited
213 high abundances in the particle phase, likely resulting from thermal decomposition of
214 larger OOMs during thermal desorption on FIGAERO.^{27, 37} The presence of small
215 OOMs could lead to an underestimation of effective molecular weight and an
216 overestimation of bulk volatility in the particle phase. Organic nitrates were clearly
217 observed in both gas and particle phases, indicating the significant role of NO_x ($3.7 \pm$
218 2.4 ppbv during the observation period) in local atmospheric chemistry. It should be
219 noted that in addition to anthropogenic sources, NO_x may also come from strong
220 natural sources in the TP, such as soil microbial activity, lakes, and lightning-induced
221 NO_x .^{38, 39}

222 The numbers of carbon and oxygen of the measured gaseous and particulate OOMs
223 were analyzed, as shown in Figure 1. C_{1-2} OOMs constituted a significant proportion

(27.4%) in the gas phase, especially CH_2O_2 (16.0%). CH_2O_2 has been assigned as formic acid, commonly found in various atmospheres and produced by various precursors.⁴⁰ Apart from C_{1-2} OOMs, C_{4-7} OOMs were the main components in the gas phase, accounting for 44.8% of the total measured OOMs. In contrast, besides C_{5-7} OOMs, C_{8-10} OOMs also exhibited high abundances in the particle phase, with the total proportion of 26.2%. The major oxygen number was 4 in both gas and particle phases, comprising 28.5% and 18.8%, respectively. Notably, the fraction of highly oxygenated organic molecules ($\text{no} \geq 6$) was significantly higher in the particle phase than that in the gas phase.

233



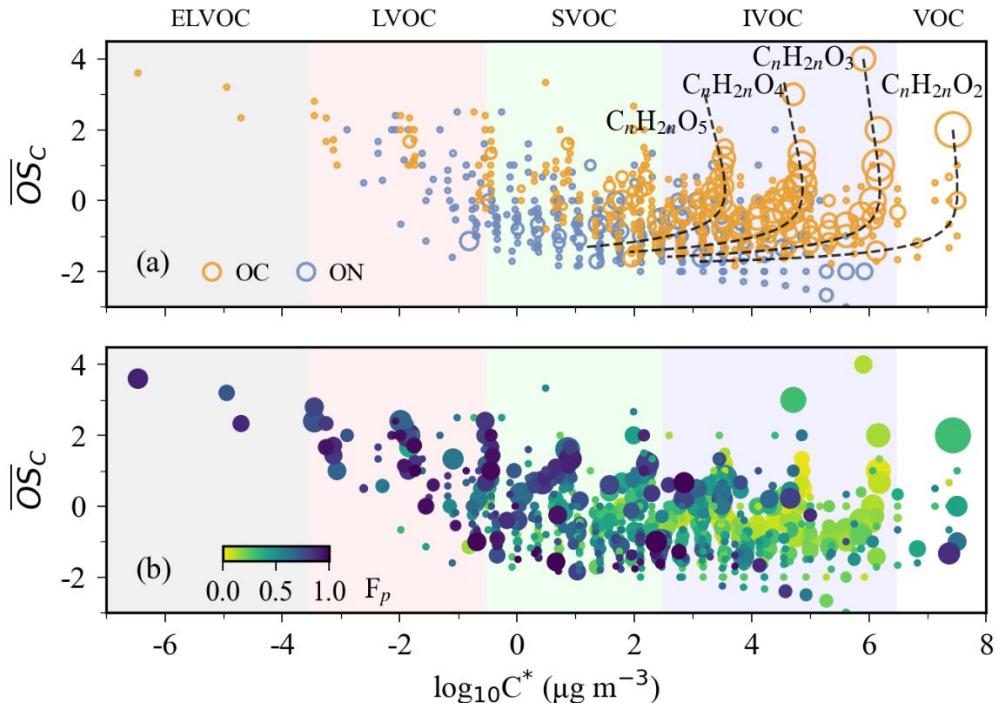
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235 **Figure 1. The distributions of carbon and oxygen numbers of the measured**
236 **OOMs. (a) Gas phase; (b) Particle phase.**

237 The different distribution characteristics between gaseous and particulate OOMs
238 depend on their different partition coefficients, which are affected by volatilities and
239 functional groups.⁴¹ Therefore, this study investigated the relationship between $\overline{\text{OS}}_C$

240 and saturation concentration (C^*) of measured OOMs, representing oxidation degree
241 and volatility, respectively, as shown in Figure 2. The calculations of \overline{OS}_C and C^* can
242 be found in Text S5. Several OOMs appeared with high volatility and oxidation
243 degree in both gas and particle phases, probably resulting from fragmentation
244 reactions during sufficient oxidation.⁴¹ Except for these OOMs, \overline{OS}_C generally
245 increased with the decrease of $\log_{10}C^*$, indicating that the addition of
246 oxygen-containing functional groups reduced volatility.⁴¹ OOMs with lower
247 volatilities were more likely to partition into the particle phase, demonstrated by their
248 higher F_p (the ratio of concentration in the particle phase to the sum concentration in
249 both the gas and particle phases) (Figure S6). However, some compounds with high
250 volatility also exhibited high F_p , probably due to the decomposition on FIGAERO.²⁷
251 ³⁷ The relationship of F_p and $\log_{10}C^*$ had a systematic shift from the theoretical line
252 based on gas-particle partitioning, indicating the volatility might be overestimated
253 through the formula method in our study.³⁷ On the contrary, there is no obvious
254 correlation between F_p and \overline{OS}_C . Gaseous OOMs predominantly fell within the
255 intermediate volatility organic compounds (IVOCs) range, accounting for 67.2% of
256 the total measured OOMs. While particulate OOMs were mainly distributed across
257 both IVOCs and semi-volatile organic compounds (SVOCs) ranges, accounting for
258 43.5% and 33.5%, respectively.

259



260

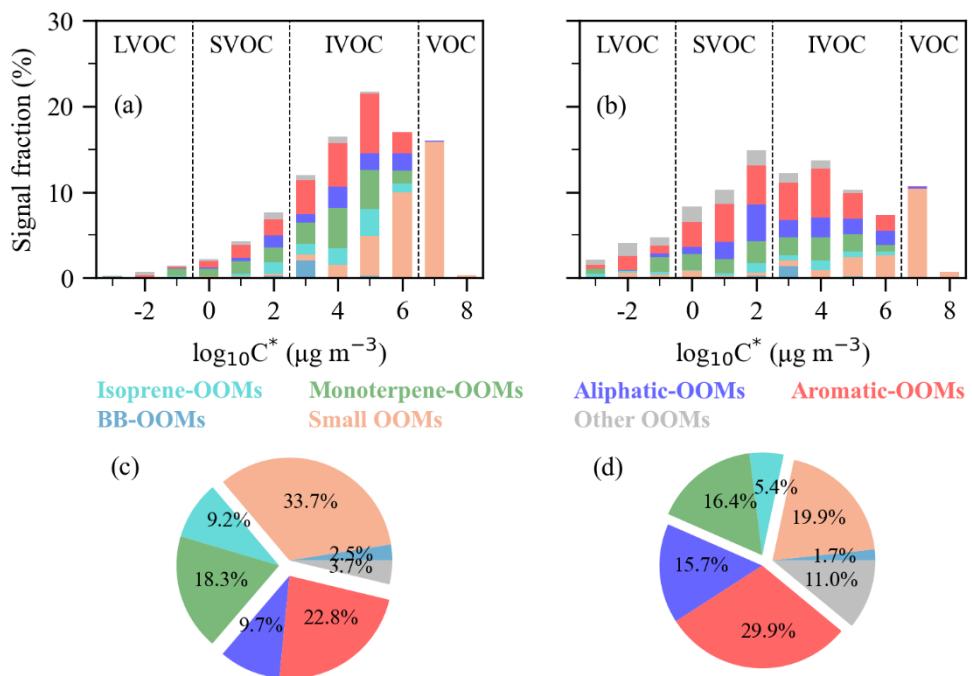
261 **Figure 2. Relationship between average carbon oxidation state (\overline{OS}_C) and**
 262 **logarithmic saturated vapor concentration ($\log_{10}C^*$) of the measured OOMs. (a)**
 263 **Gas phase; (b) Particle phase. The circle size is proportional to the logarithmic**
 264 **signal fraction. The black dotted lines in panel (a) are theoretical curves for these**
 265 **formulas. The circles in panel (b) are colored by F_p (the ratio of concentration in**
 266 **the particle phase to the sum concentration in the gas and particle phases). Both**
 267 **(a) and (b) include all measured OC and ON species.**

268 **3.2 Precursor Apportionment of Oxygenated Organic Molecules**

269 Biomass burning has been recognized as an important source in the TP.²⁸ $C_6H_{10}O_5$
 270 detected by FIGAERO-CIMS has been previously attributed to levoglucosan and its
 271 isomers,^{31, 42} regarded as important makers for biomass burning.³² The average mass
 272 concentration of particulate $C_6H_{10}O_5$ was $1.21 \pm 1.91 \text{ ng m}^{-3}$ in this study, slightly

273 lower than the result at another site in the TP,¹⁹ but significantly lower by 1-2 orders
 274 of magnitude compared to the results from typical megacities.^{11, 31} This may be
 275 attributed to the enhanced intensity of biomass burning activities and unfavorable
 276 meteorological conditions during certain seasons in megacities.^{11, 43} The diurnal
 277 profile of both gaseous and particulate C₆H₁₀O₅ exhibited a significant peak in the
 278 morning (Figure S7), suggesting enhanced biomass burning activity. In addition,
 279 C₆H₅NO₄, C₅H₁₀O₄, and C₈H₁₂O₆ displayed strong correlations with C₆H₁₀O₅ in both
 280 gas and particle phases ($R > 0.8$), indicating that they might also originate from
 281 biomass burning.

282



283

284 **Figure 3. Contributions of different precursor classes to the measured OOMs.**
 285 **Contributions of different OOMs at each volatility bin in the (a) gas phase, and**
 286 **(b) particle phase, respectively. Contributions of different OOMs to the total**

287 measured OOMs in the (c) gas phase, and (d) particle phase, respectively.

288 “BB-OOMs” represents the OOMs from biomass burning. “Small OOMs”

289 represents the OOMs with carbon number ≤ 3 .

290 After excluding 4 BB-OOMs and 61 small OOMs, we identified 50
291 isoprene-OOMs, 142 monoterpene-OOMs, 157 aliphatic-OOMs, 228 aromatic-OOMs,
292 and 89 aromatic-OOMs using the random forest model (Figure 3 and Figure S8). And
293 the remaining 89 OOMs were classified into “Other OOMs”. The diurnal patterns of
294 four types of OOMs are shown in Figure S9. As mentioned above, C₁₋₃ OOMs were
295 excluded as small OOMs from classification, due to the large overlap of them in the
296 oxidation products of various precursors.³⁴ Gaseous and particulate BB-OOMs
297 accounted for only 2.5% and 1.7% of the total measured OOMs, respectively. As
298 mentioned in the method, BB-OOMs only represent the compounds directly emitted
299 from biomass burning. Chen et al.²⁸ revealed the important contribution of biomass
300 burning to particles at the same site through the measurements of n-alkanes and
301 polycyclic aromatic hydrocarbons, which cannot be measured by FIGAERO-CIMS.

302 The measured oxidation products of n-alkanes and polycyclic aromatic hydrocarbons,
303 which may be emitted by biomass burning, will be categorized as aliphatic-OOMs
304 and aromatic-OOMs, respectively, in our study. As shown in Figure S10, over 75% of
305 isoprene-OOMs, aliphatic-OOMs, and aromatic-OOMs were distributed in the IVOCs
306 range in the gas phase. While 71.9% and 23.1% of gaseous monoterpene-OOMs were
307 distributed in the IVOCs and SVOCs ranges, respectively. In contrast, more
308 isoprene-OOMs, monoterpene-OOMs, aliphatic-OOMs, and aromatic-OOMs were

309 found in the SVOCs range in the particle phase, indicating that these OOMs were
310 more likely to partition into particle. Wang et al. (2024)³⁴ showed that the OOMs
311 derived from four precursors mainly distributed in the low volatility organic
312 compounds (LVOCs) range at multiple sites. This is because the volatility estimated
313 through formula method in our study would be several orders of magnitude higher
314 than that estimated through thermogram method in Wang et al. (2024).³⁷ To
315 investigate the sources of OOMs from different precursors, we simply assumed that
316 isoprene-OOMs and monoterpene-OOMs originated from biogenic sources, and
317 aliphatic-OOMs and aromatic-OOMs originated from anthropogenic sources.³⁵
318 Isoprene may be emitted from anthropogenic sources, but in background region the
319 anthropogenic fraction of isoprene is quite small.⁴⁵ Aromatics can also come from
320 biogenic sources, such as cyanobacteria in the lake.⁴⁶ Since there are no large lakes
321 around the sampling site, this impact can also be ignored. BB-OOMs were not
322 categorized as biogenic or anthropogenic sources. Due to the small fraction of
323 BB-OOMs, it will not significantly affect the conclusions. Monoterpene-OOMs and
324 isoprene-OOMs constituted only 27.5% and 21.8% of the total measured OOMs in the
325 gas and particle phases, respectively. The contribution of monoterpene-OOMs was
326 significantly higher than that of isoprene-OOMs, indicating that monoterpenes might
327 be more important in biogenic SOA formation in the southeastern TP.
328 Aliphatic-OOMs contributed 9.7% and 15.7% to the total measured OOMs in the gas
329 and particle phases, respectively, with alkanes regarded as the dominant precursors.³⁵
330 The proportions of aromatic-OOMs were 22.8% and 29.9%, respectively.

331 Aliphatic-OOMs and aromatic-OOMs contributed 32.5% and 45.6% to the total
332 measured OOMs in the gas and particle phases, respectively, emphasizing a more
333 critical role of anthropogenic sources in OOMs formation than biogenic sources in the
334 southeastern TP. If all other OOMs were classified as biogenic OOMs, the fraction of
335 total biogenic OOMs would be 31.2% and 32.8% in the gas and particle phases,
336 respectively, still lower than that of anthropogenic OOMs, indicating these
337 unclassified OOMs would not significantly impact the conclusions. Since small
338 OOMs account for only a minor fraction in the particle phase, including them in the
339 model did not significantly affect the overall conclusions (Figure S11). However, the
340 contribution of biogenic sources to OOMs would increase to a level comparable to
341 that of anthropogenic sources, as a larger number of small OOMs were classified as
342 biogenic OOMs. This highlights the need for accurate source apportionment of small
343 OOMs in future studies, for example through isotopic analysis.⁴⁴

344 To investigate the impact of regional transport to OOMs formation, a cluster
345 analysis of 72h backward air mass trajectories at 500m above ground level was
346 performed, as shown in Figure S12. This height was chosen to better represent the
347 lower free troposphere over the plateau's complex terrain, such as mountain and
348 valley.³⁸ The 72h duration was chosen based on the typical atmospheric lifetime of
349 OOMs. The air masses over the Lulang site primarily originated from the south, with
350 the shortest transport distance constituting 81% of all trajectories, indicating that local
351 emissions likely played a dominant role in shaping the observed OOMs composition.
352 There are no significant differences in OOMs compositions in both gas and particle

353 phases among different clusters, as shown in Figure S13. Although the air mass
354 clusters varied in transport distance, they all passed through the region associated with
355 the dominant short-range cluster. Consequently, characteristics of OOMs from
356 long-range transport may be masked by local emissions. Given the proximity of a
357 village, a town, and National Road 318, anthropogenic precursors and their oxidation
358 products may be transported to the Lulang site by these prevailing air masses from the
359 three nearby emission regions.

360 To the best of our knowledge, the present work is the first study to assign specific
361 precursors to OOMs measured by FIGAERO-CIMS in the TP. Previous studies have
362 resolved the contributions of primary and secondary sources to OA in other TP
363 regions, but the roles of different precursors have not been well characterized.^{16, 17, 47}
364 Through the measurement of SOA tracers, Shen et al.¹⁹ found that secondary organic
365 carbons (SOC) from aromatics accounted for approximately 25% of the total
366 estimated SOC at Nam Co lake. Guo et al.²⁹ demonstrated the significant
367 contributions of aromatics to SOA through the estimation of SOA potentials of VOC
368 components at the same site. However, Liu et al.²² reported that anthropogenic
369 precursors explained only 10% of extremely low-volatility organic compounds
370 measured by nitrate CIMS at the same site, lower than our results. The difference may
371 be attributed to the different measurement ranges of organic compounds,⁴⁸ as
372 FIGAERO-CIMS mainly measured S/IVOCs in our study. The accurate quantification
373 of both instruments may be important for their comparability in future study.⁴⁹ It
374 should be noted that the relative contributions of anthropogenic and biogenic sources

375 to OOMs in the TP can be influenced by multiple factors, such as emissions patterns,
376 atmospheric conditions, and sampling period. Bianchi et al.²¹ reported that biogenic
377 sources were an important contributor to particle at a site located at 5079 m a.s.l.,
378 possibly due to lower anthropogenic emissions in that region. Shen et al.¹⁹ found that
379 the contribution of anthropogenic sources to SOC increased significantly in April
380 compared to other summer months, which coincides with the sampling period of this
381 study. In addition, the impacts of anthropogenic sources on atmospheric species have
382 also been observed at other mountain sites.⁵⁰⁻⁵² For instance, Zhang et al.⁵³ discovered
383 that anthropogenic sources contributed 16%-35% to total gaseous organic compounds
384 at Shanghuang Mountain. Therefore, establishing the links between precursors and
385 OOMs in background sites should be given more consideration in future studies for
386 better understanding the impacts of anthropogenic sources on background
387 atmosphere.

388 Although our study provides valuable insight into the OOMs species in the TP,
389 there are several limitations that need to be considered. Firstly, the sampling duration
390 is limited to approximately two weeks in April 2021. As stated by previous studies,^{19,}
391 ²⁸ the emissions of biogenic and anthropogenic sources display significant seasonal
392 variation in the TP. For instance, biogenic emissions are typically enhanced during
393 summer,¹⁹ while anthropogenic emissions associated with tourism may peak at other
394 seasons.²⁸ The relative importance of biogenic and anthropogenic sources may vary in
395 different seasons. Nevertheless, our results provide a representative snapshot of
396 OOMs characteristics during the spring period in the TP and can serve as a baseline

397 for future seasonal comparisons. Secondly, there exist some uncertainties of
398 classifications based on the machine learning approach, especially for those OOMs
399 from multiple precursors. There are 18 species belonging to both isoprene-OOMs and
400 aliphatic-OOMs, and 29 species belonging to both monoterpene-OOMs and
401 aromatic-OOMs in the dataset.³⁴ We compared the occurrence probability of these
402 OOMs, as shown in Figure S14. The probability of aromatic was higher than that of
403 monoterpene for most OOMs derived from both aromatic and monoterpene,
404 indicating the aromatic-OOMs would be overestimated in the actual atmosphere. In
405 the real atmosphere, the contributions of different sources to these overlapping OOMs
406 may be influenced by meteorological conditions and the relative proportions of
407 emission sources. However, these factors are not considered in the current model. To
408 better understand this uncertainty, we performed a sensitivity analysis on the model.
409 We considered two scenarios: all overlapping OOMs were assumed to originate from
410 anthropogenic sources, and all overlapping OOMs were assumed to originate from
411 biogenic sources. The results (Figure S11) showed that anthropogenic OOMs
412 remained more abundant than biogenic OOMs, except for the gas phase under the
413 second scenario. This indicates that the classification of gas-phase OOMs may be
414 more susceptible to uncertainties associated with overlapping OOMs. In addition, the
415 value of threshold (0.6) may affect the probabilities of overlapping OOMs. To assess
416 the sensitivity of our results to this parameter, we also examined thresholds of 0.4 and
417 0.8, with the results shown in Figure S11. In both cases, contributions of
418 anthropogenic sources remained higher than those of biogenic sources, suggesting the

419 robustness of the results with respect to the choice of threshold. Thirdly, considering
420 the same sensitivity was assumed for all measured OOMs, the contribution of
421 different sources has large uncertainty. The transmission, which is a function of m/z,
422 is an important factor in affecting the sensitivity of different OOMs.⁵⁴ The
423 distributions of biogenic and anthropogenic OOMs on m/z were compared, as shown
424 in Figure S15. In the gas phase, their distributions were similar, indicating the
425 transmission has little impact on the contribution of two sources. In the particle phase,
426 biogenic OOMs distributed more in the large m/z range, indicating its contribution
427 would be slightly underestimated.

428

429 **3.3 Atmospheric Implications**

430 As the “Third Pole”, the TP significantly affects atmospheric circulation, global
431 climate, and cryosphere change.¹³ The TP’s atmospheric environment has been
432 considered representative of global background conditions due to limited human
433 activities.¹⁵ However, economic development has resulted in the inevitable emission
434 of anthropogenic pollutants into the TP’s environments.^{55, 56} Kang et al.¹³ noted that
435 these exogenous pollutants substantially impact regional climate and environmental
436 changes in the TP. In recent decades, the TP has experienced considerable climate
437 warming.⁵⁷ Anthropogenic pollutants can contribute to climate change through their
438 effects on solar radiation.⁵⁸ Our study highlights the significant impacts of
439 anthropogenic sources on OOMs formation in the southeastern TP, with
440 aromatic-OOMs being the dominant components of particulate OOMs. Laboratory

441 investigation by Nakayama et al.⁵⁹ demonstrated that secondary organic aerosols from
442 toluene photooxidation exhibit substantial light absorption properties, indicating
443 substantial brown carbon formation that contributes to positive radiative forcing and
444 atmospheric warming. These experimentally determined optical properties are
445 applicable to our observed anthropogenic OOMs, as aromatic precursors may undergo
446 comparable photooxidation.⁶⁰ Ji et al.⁵⁸ conducted comprehensive simulations of
447 carbonaceous aerosols over the TP and adjacent regions, revealing the importance of
448 aerosol longwave radiative forcing to climatic effect of aerosols. Our molecular-level
449 characterization of anthropogenic OOMs provides crucial chemical speciation data
450 that bridges the gap between emission sources and the formation of climatically active
451 light-absorbing secondary organic compounds, thereby enabling more accurate
452 parameterization of aerosol optical properties in regional climate models. These
453 findings underscore the necessity of considering anthropogenic influences in future
454 research. Overall, our study provides valuable insights into atmospheric species, and
455 enhances our understanding of pollutant sources and atmospheric chemical processes
456 in the TP.

457

458 **Supporting information**

459 Details about the FIGAERO-CIMS (Text S1, and Table S1). Details about the random
460 forest approach (Text S2-S4, Table S2, Figure S1, S3, S4, and S14). Calculations of
461 C^* and \overline{OS}_C (Text S5). Ratio of average concentrations between two different periods
462 (Figure S2). Campaign-averaged mass spectra (Figure S5). Relationship of F_p with

463 $\log_{10}C^*$ and \overline{OS}_C (Figure S6). Diurnal variations of BB-OOMs (Figure S7).
464 Classification of the measured OOMs (Figure S8, S9, and S10). Sensitivity analysis
465 (Figure S11). Cluster results of backward trajectories (Figure S12 and S13).
466 Distribution of different OOMs on m/z (Figure S15).

467

468 **Conflict of Interest Disclosure**

469 The authors declare no competing financial interest.

470

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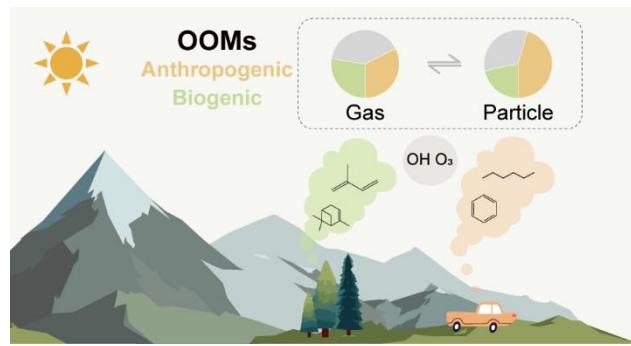
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