

# A simulation framework for the modeling of adsorption related noise phenomena in electrochemical sensors

K. J. Krause<sup>a</sup>, E. Kätelhön<sup>a</sup>, P. S. Singh<sup>b</sup>, K. Mathwig<sup>b</sup>, S. G. Lemay<sup>b</sup> and B. Wolfrum<sup>ac</sup>

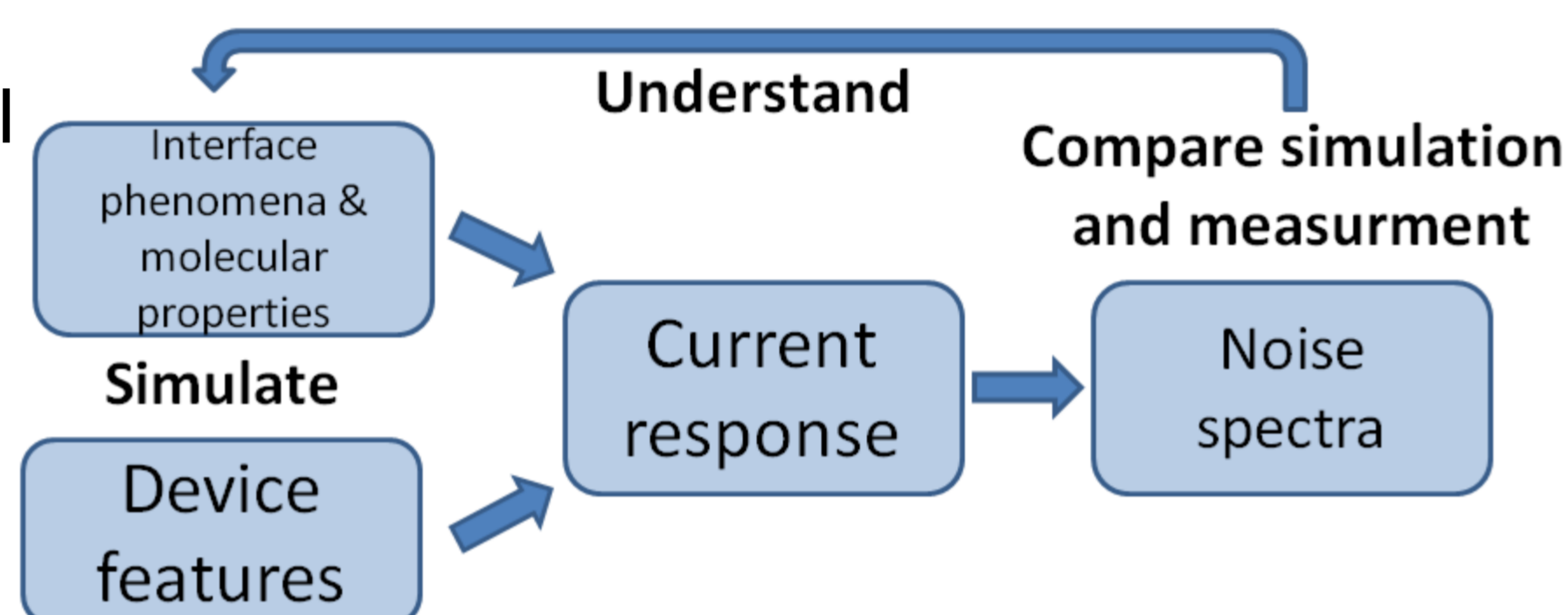
a) Institute of Bioelectronics (PGI-8/ICS-8) and JARA—Fundamentals of Future Information Technology, Forschungszentrum Jülich, D-52425 Jülich, Germany  
b) Institute for Nanotechnology, University of Twente, PO Box 217, 7500 AE Enschede, The Netherlands  
c) IV. Institute of Physics, RWTH Aachen University, 52074 Aachen, Germany

## Introduction

Nanocavity redox cycling devices are a promising tool for sensing applications. In most studies, the temporal current response of these devices is investigated. However, the spectrum of the recorded traces also offers unique information on the system allowing fluctuation effects to be investigated in the frequency regime<sup>[1]</sup>. Here we use a random walk model to investigate the impact of adsorption on the power spectra of redox cycling devices.

## Motivation

Simulate electrochemical devices at the molecular level for a deeper understanding of the device response.



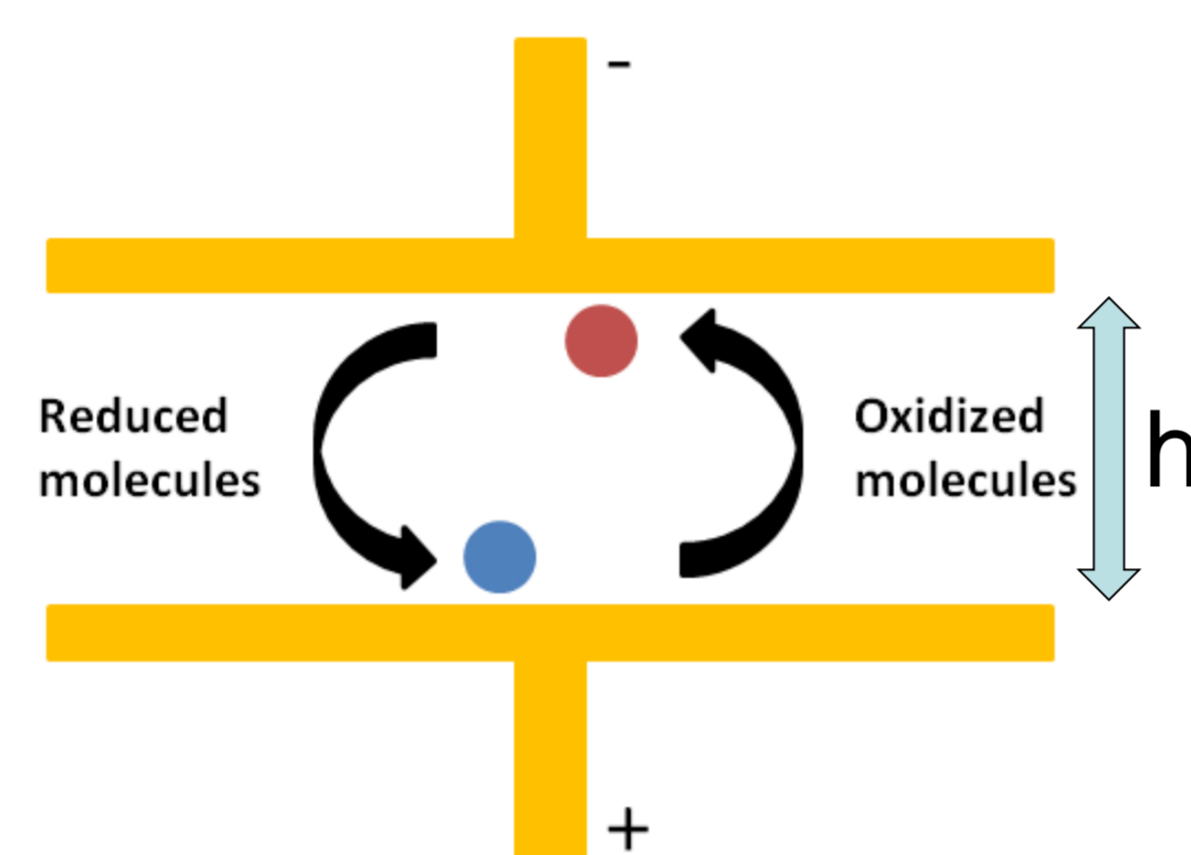
## Electrochemical detection and redox cycling

- Diffusive molecular movement in between two individually biased electrodes enables repetitive redox reactions at the two electrodes.

- Amplified net current across the gap:

$$I = \frac{e_0 z D n}{h^2}$$

$e_0 z$  = transferred charge per reaction  
 $n$  = number of molecules in between the electrodes  
 $D$  = diffusion constant



Redox cycling principle

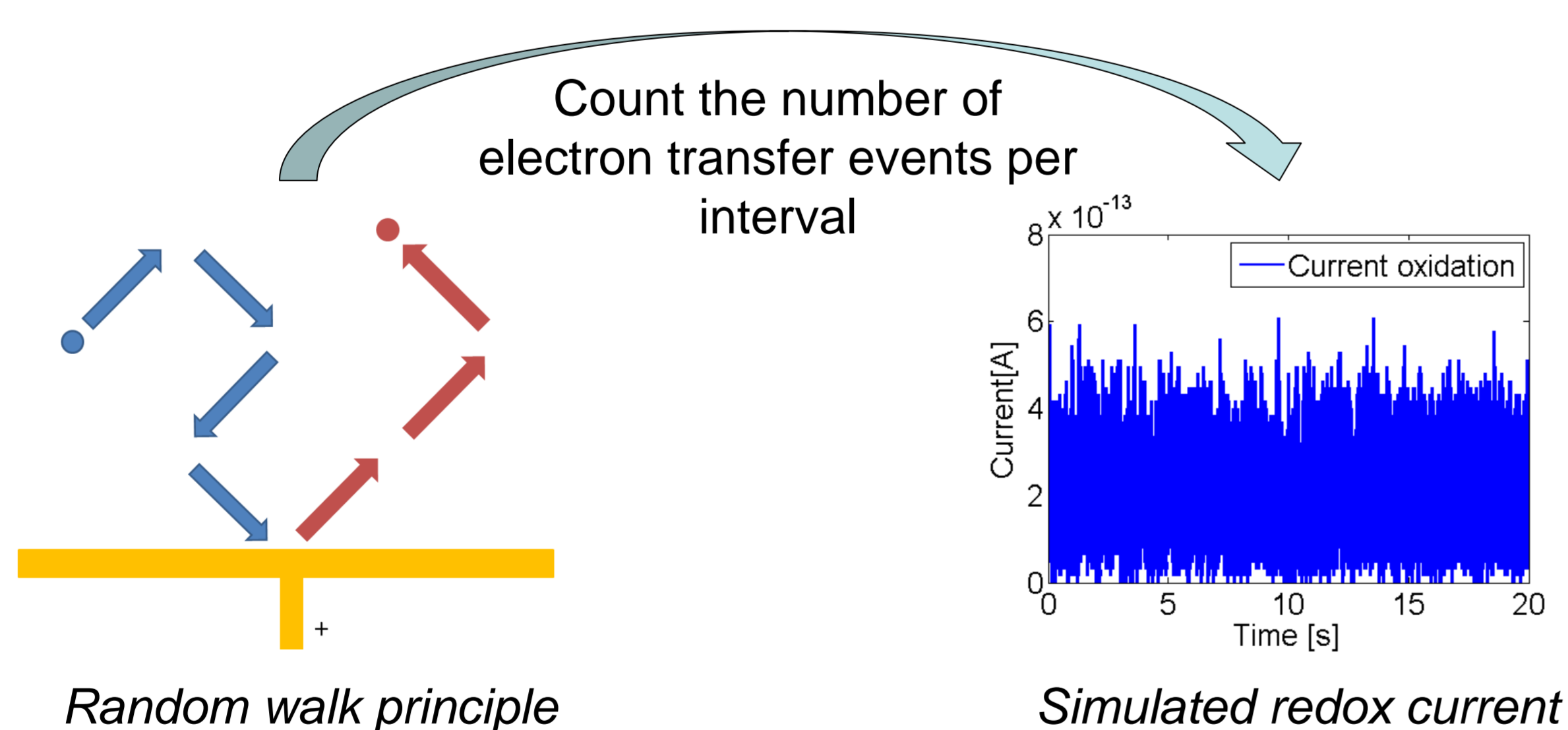
## Random walk – Simulation of molecule movement

- Written in C/C++, supports parallel computing, freely available

- Independent random walks for each molecule

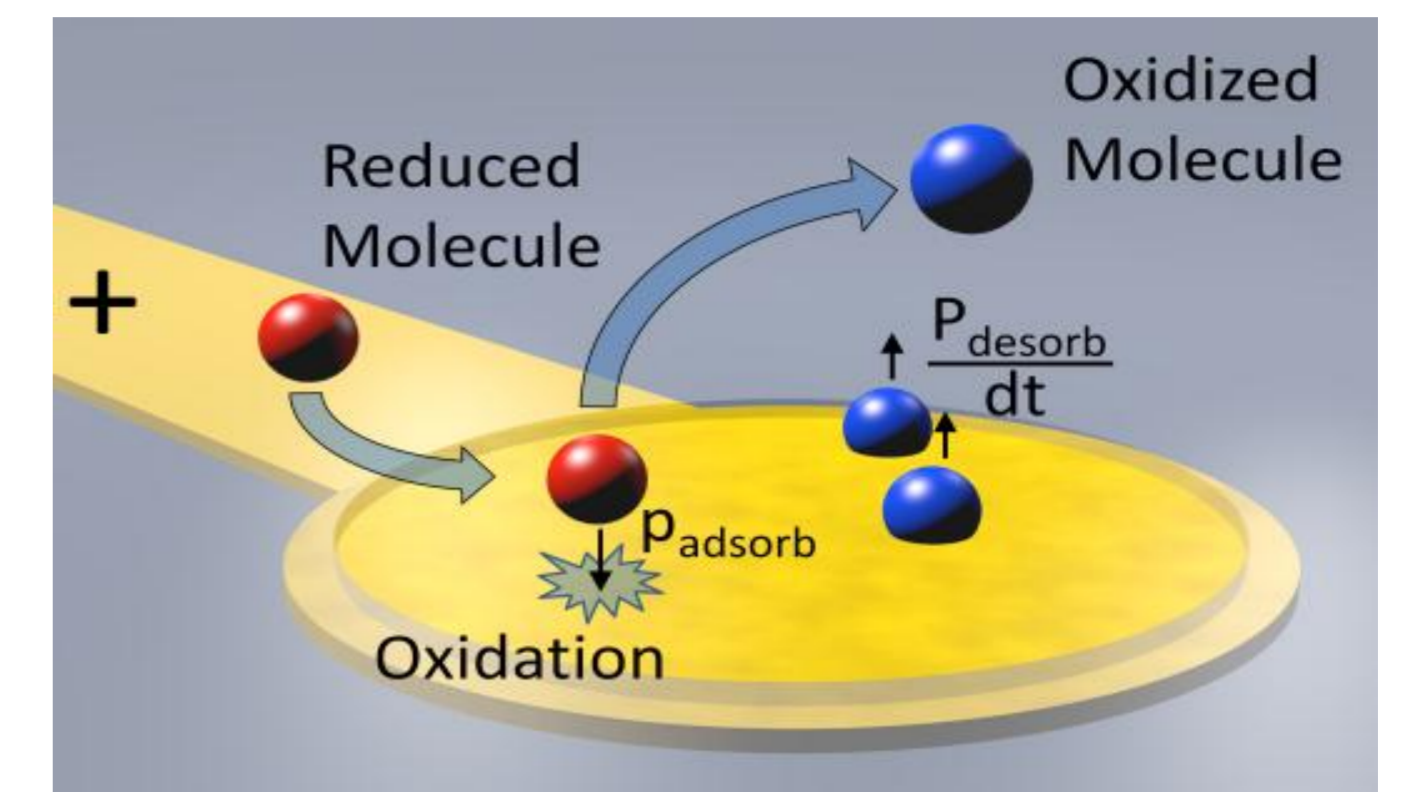
- Temporal and spatial step width match the diffusion equation

- Molecules can adopt two oxidation states



## Reversible adsorption

- Adsorption probability  $p_{\text{adsorb}}$
- Desorption probability  $p_{\text{desorb}}$  per random walk iteration  $dt$

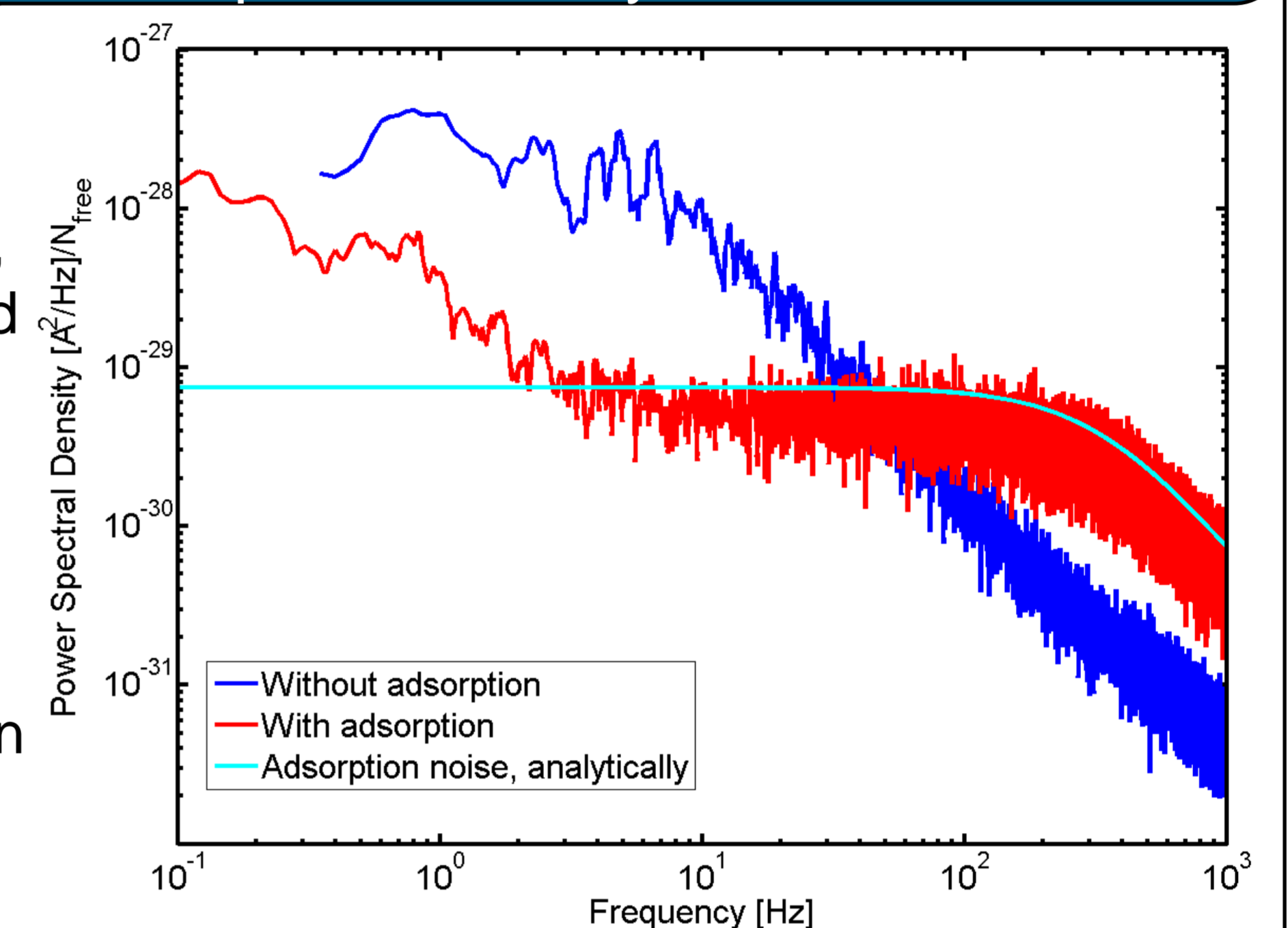


## Power spectral density

- Due to the slower effective diffusion caused by adsorption, the spectrum is shifted to lower frequencies

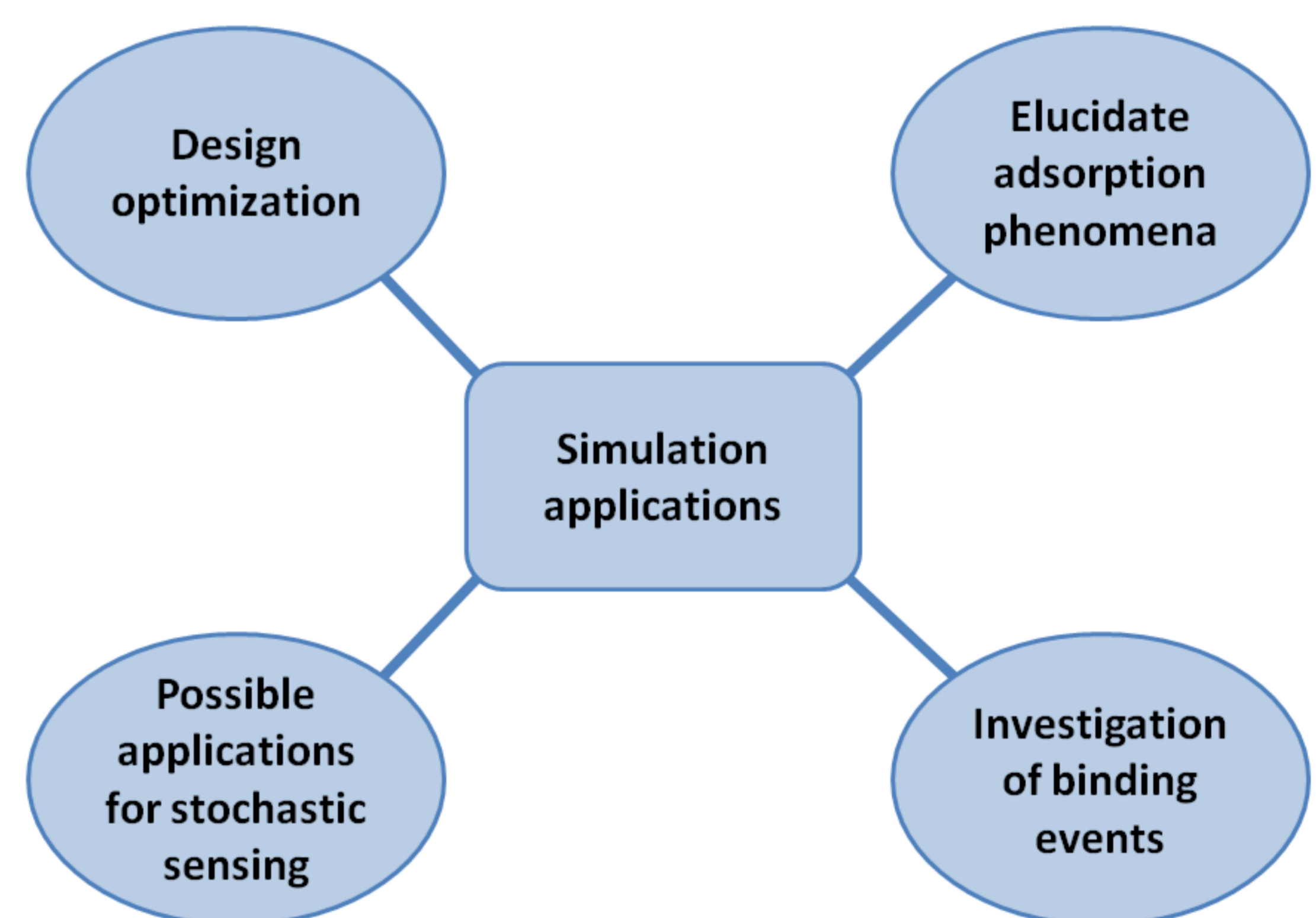
- In addition to the fluctuation noise plateau at low frequencies there is an adsorption noise plateau at higher frequencies

- The frequency regime of this plateau depends on the mean adsorption time



Power spectral density of the redox current with and without adsorption

## Simulation applications



## Conclusion

We present a simulation framework for the investigation of reversible adsorption on electrochemical redox cycling sensors. The code is freely available for further use and expansion<sup>[3]</sup>.

## Acknowledgements

We gratefully acknowledge funding by the Helmholtz Young Investigators Program.

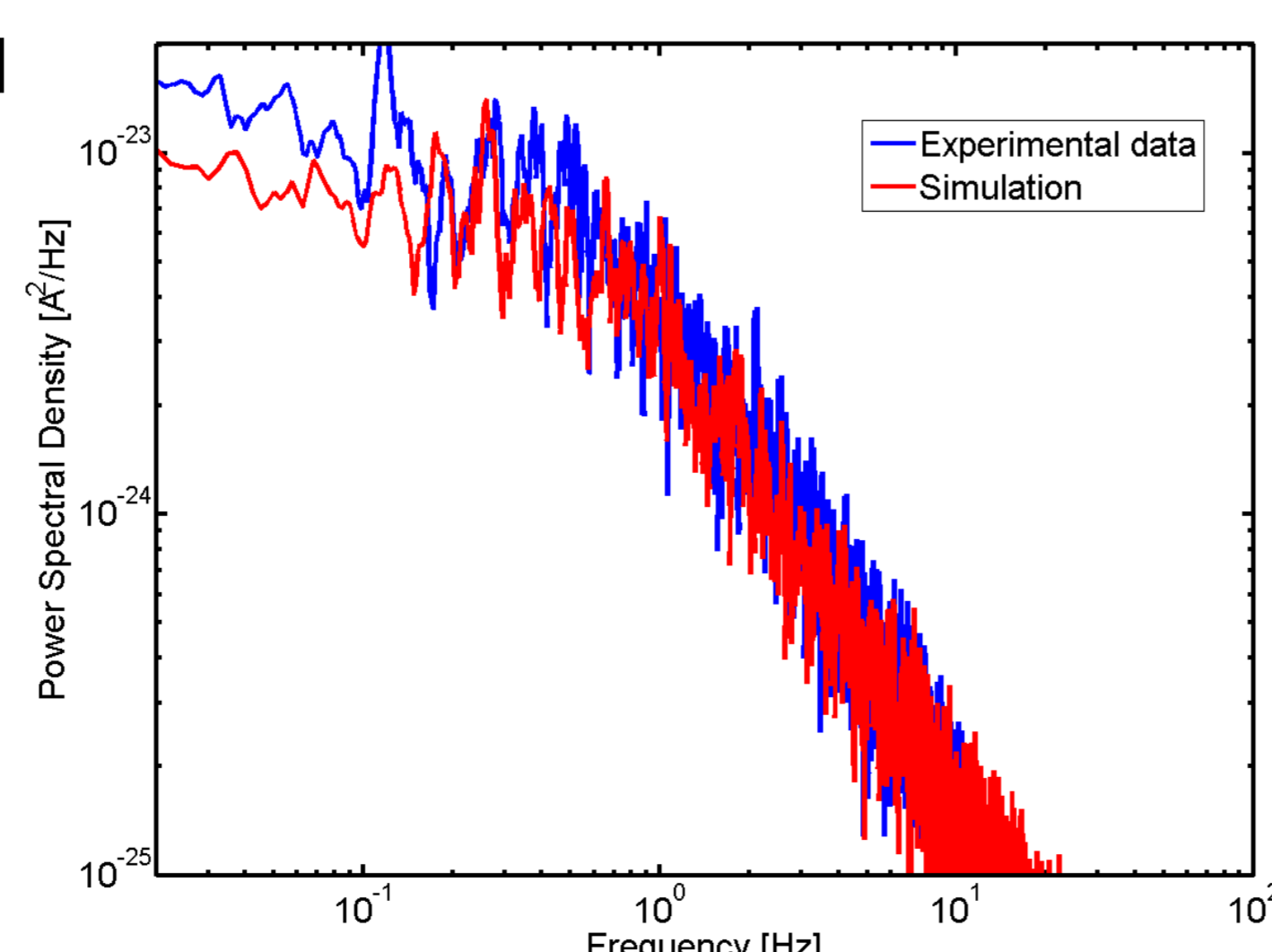
## References

- M. A. G. Zevenbergen, P. S. Singh, E. D. Goluch, B. Wolfrum, S. G. Lemay, *J. Am. Chem. Soc.*, **81**, (2009) 8203-8212
- E. Kätelhön, B. Wolfrum, *Phys. Status Solidi A*, 209, No. 5, 881–884 (2012)
- E. Kätelhön, K. J. Krause, P. S. Singh, S. G. Lemay, B. Wolfrum, *J. Am. Chem. Soc.*, **135**, (2013) 8874-8881

## Simulation and experiment

- Blue: Experimental data, 50  $\mu\text{M}$   $\text{Fc}(\text{MeOH})_2$  solution.

- Red: Simulation, analytically corrected for adsorption<sup>[3]</sup>.



Comparison of the power spectral density of the redox current of experimental data and simulation results.

