



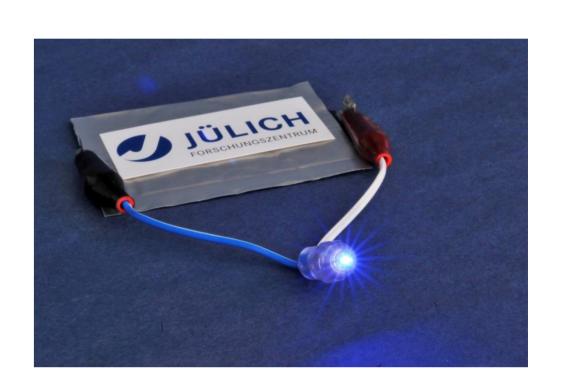
## $\text{Li}_{7}\text{La}_{3}\text{Zr}_{7}\text{O}_{17}$ electrolyte for all-solid-state batteries

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A major drawback of conventional Li-ion batteries is the use of organic liquid electrolytes. As **CONCLUSION & OUTLOOK** an alternative, all-solid-state batteries with one of the most promising oxide materials,  $\text{Li}_7 \text{La}_3 \text{Zr}_7 \mathbb{O}_{17}$  (LLZ) are investigated. LLZ is an ionic conductor with a good thermal and electrochemical stability (up to 1250°C and 8V vs. Li/Li+) and a chemical compatibility to metallic lithium. The ion conductivity can be further improved by partial substitution of Al, Ta or Y into the LLZ.

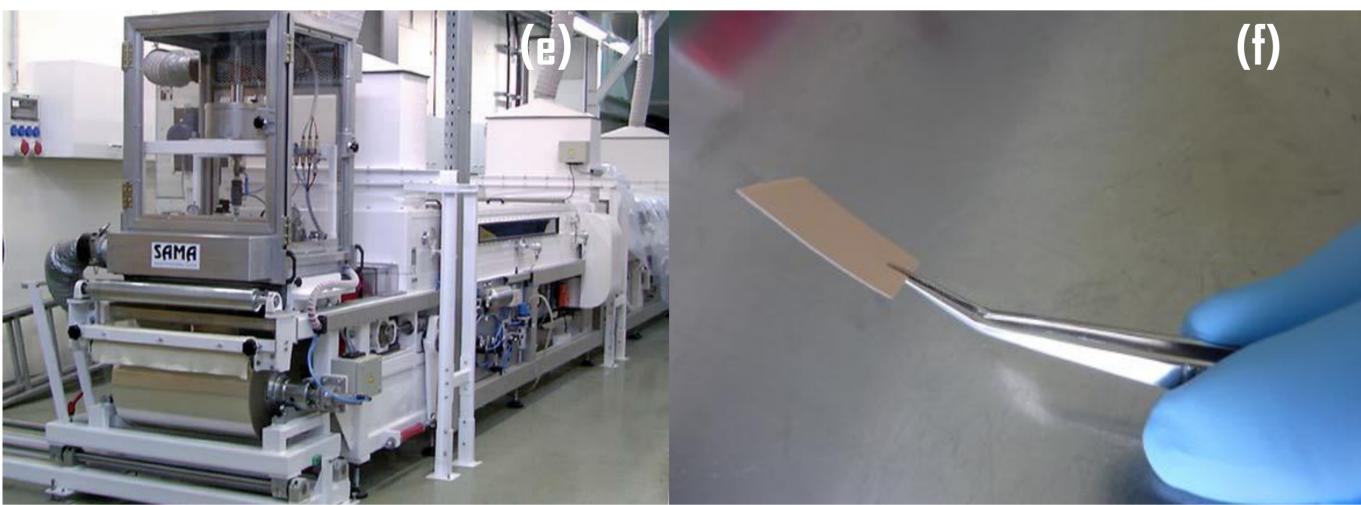
At IEK-1, we investigate two main approaches for all-solid-state battery fabrication. Very thin layers are processed by PVD aiming to achieve a thin film battery in the range of a few micrometers. For large scale fabrication of functional layers, tape casting of LLZ is investigated.

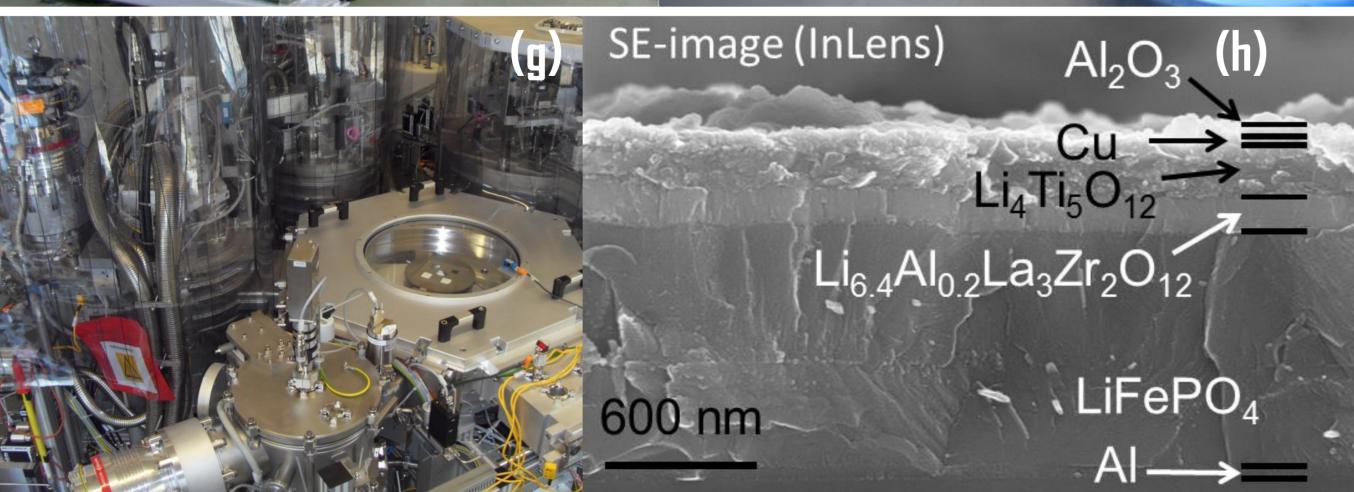
- Al-, Ta- and Y-substituted LLZ were investigated for different Li concentrations with respect to the materials conductivity. Ta:LLZ shows the highest total conductivity ( $\sigma_{inn} RT \approx 10^{-3}$  S/cm) and almost no dependence on Li concentration.
- LCO|Ta:LLZ|Li all-solid-state cell was fabricated from bulk sample and drove a LED at 22°C (ASB) works!)
- LLZ can be fabricated by tape casting and PVD processing.
- → By combining tape casting and PVD processing a thin allsolid-state battery with higher energy density should be developed.



## MATERIAL RESEARCH

- Inductively Coupled Plasma (ICP) measurement of Li concentration in (a) Alsubstituted LLZ, **(b)** Ta-substituted LLZ and **(c)** Y-substituted LLZ. There is a high correlation between Li concentration in Al- and Y-substituted LLZ and the total conductivity but not for Ta-substituted LLZ. The molar numbers were normalized to Zr concentration.
- (d) Temperature dependence of the ionic conductivity of Al-, Ta- and Y-substituted LLZ. Ta-substituted LLZ shows the highest total ionic conductivity among the three materials. The total conductivity was derived from impedance spectra (1MHz-1Hz, amplitude: 20 mV/mm, temperature range:  $22-300^{\circ}\text{C}$ ).
- **(a)** Total conductivity (S/cm) Ta-LLZ **Dwell time (hours) Dwell time (hours)** Temperature (°C) **(C)** uctivity (S/cm) Ta-LLZ Y-LLZ 2,25 2,50 2,75 3,00 3,25 3,50 1000/T (1/K) **Dwell time (hours)**





- Synthesis of several kilograms of Al-substituted LLZ by spray pyrolysis at once.
- Fabrication of LLZ by easily up-scalable **(e)** tape casting at IEK-1 to **(f)** 90 µm thick green-tape substrates. These tapes are used for sintering studies at different temperatures and atmospheres.
- Thin film electrodes or solid electrolytes like LLZ can be successfully deposited by (g) physical vapor deposition (PVD) at IEK-1. The thickness and adhesion of these thin layers can be visualized by **(h)** SEM. Between an aluminum (AI) and copper (Cu) conductor LFP (LiFePO<sub>4</sub>) was used as cathode and LTO ( $Li_4T_5D_{17}$ ) as anode material. The electrodes were separated by Alsubstituted LLZ ( $\text{Li}_{6.4}\text{Al}_{0.7}\text{La}_{3}\text{Zr}_{7}\text{O}_{17}$ ). On top an  $\text{Al}_{7}\text{O}_{3}$ -protection layer is deposited.

## Acknowledgements

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