## INTERACTIONS BETWEEN MAGNESIUM SILICATE SUPPORT SUBSTRATE AND STATE-OF-THE-ART CATHODES DURING CO-SINTERING OF AN ALL-CERAMIC SOFC

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3 JARA-ENERGY, AACHEN, GERMANY



### **PROJECT: KERSOLIFE 100**

#### Funded by:



#### Goal:

Understanding and modelling of material interactions and degradation mechanisms in all-ceramic SOFCs

#### **Project coordination**

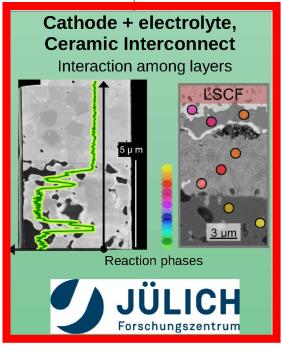


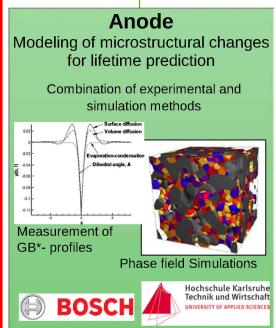
Long-term tests

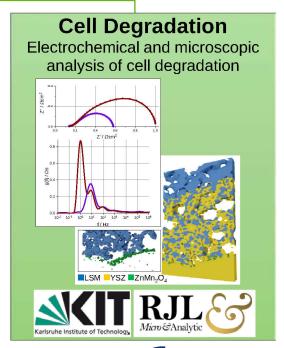
Models for degradation analysis

Implementation of findings into industrial field

# Ceramic support Thermo-mechanical stability and reliability of ceramic support Measuring of temperature-dependent K<sub>IC</sub> Fractographical analysis Hochschule Aalen BOSCH



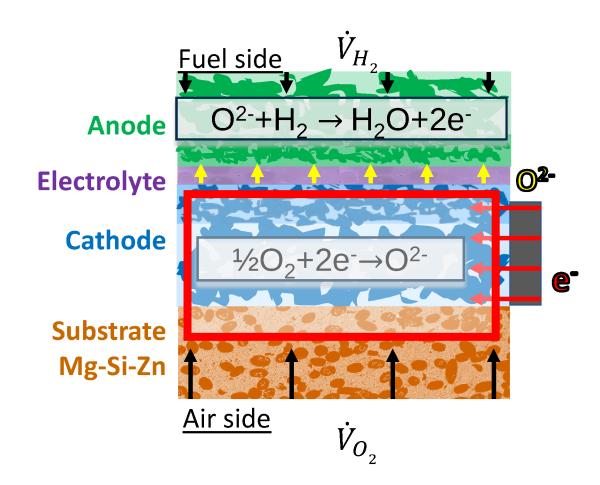






### **INERT SUPPORTED SOLID OXIDE FUEL CELL (SOFC)**

- All ceramic SOFC concept [1]
- Support material Zn-doped magnesium silicate (Mg-Si-Zn)
- Simplified manufacturing route: co-sintering
- Compromised sintering temperature
   <1300°C for all layers</li>
- Materials of each layer adjusted to new processing route
- Here, focus on the cathode







### **CONTENT OF PRESENTATION**

### **Evaluation of cathodes within design:**

- 1. Chemical interactions
- 2. Electrochemical performance
- 1. Microstructure
- 2. Co-sinterability
- 3. Summary

Index of te	ested cathodes		
LSM	La <sub>x</sub> Sr <sub>y</sub> MnO <sub>3-d</sub> (baseline)	LSF	$La_{0,58}Sr_{0,4}FeO_3$
PSCF	$Pr_{0,58}Sr_{0,4}Co_{0,2}Fe_{0,8}O_3$	LCCF	$La_{0.58}Ca_{0.4}Co_{0.2}Fe_{0.8}O_3$
LSCF	$La_{0,58}Sr_{0,4}Co_{0,2}Fe_{0,8}O_3$	LSFM_95S1M3	$(La_{0.9}Sr_{0.1})_{0.95}Fe_{0.7}Mn_{0.3}O_3$
LSC	$La_{0,58}Sr_{0,4}CoO_3$	LSFM_95S2M8	$(La_{0.8}Sr_{0.2})_{0.95}Fe_{0.2}Mn_{0.8}O_{3}$
Mitglied der Helmhol	tz-Gemeinschaft		Forschungsze

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### CHEMICAL INTERACTIONS

Introduction Chemical interactions

Electrochemical performance

Microstructure

Co-sintering

Summary

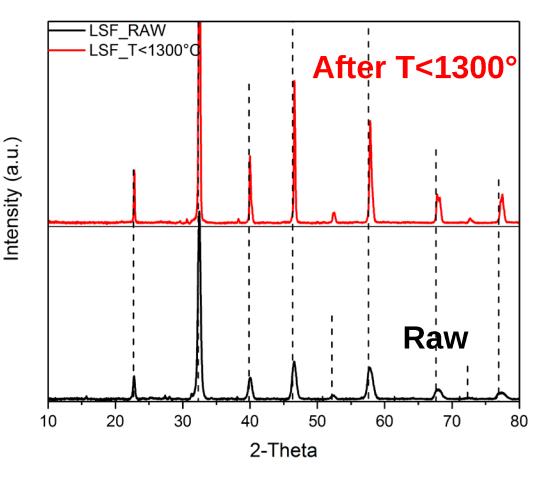


### STABILITY OF PEROVSKITES AFTER THERMAL

#### TREATMENT

- XRD measurements
- Perovskite phase stability tested before and after sintering
- For all cathodes: LSC, LSCF, PSCF, LSM, LSFM, LSF:
  - perovskite appears stable after <1300°C, 5h
- Example given: LSF

LSF





#### REACTIVITY SUPPORT SUBSTRATE AND CATHODES

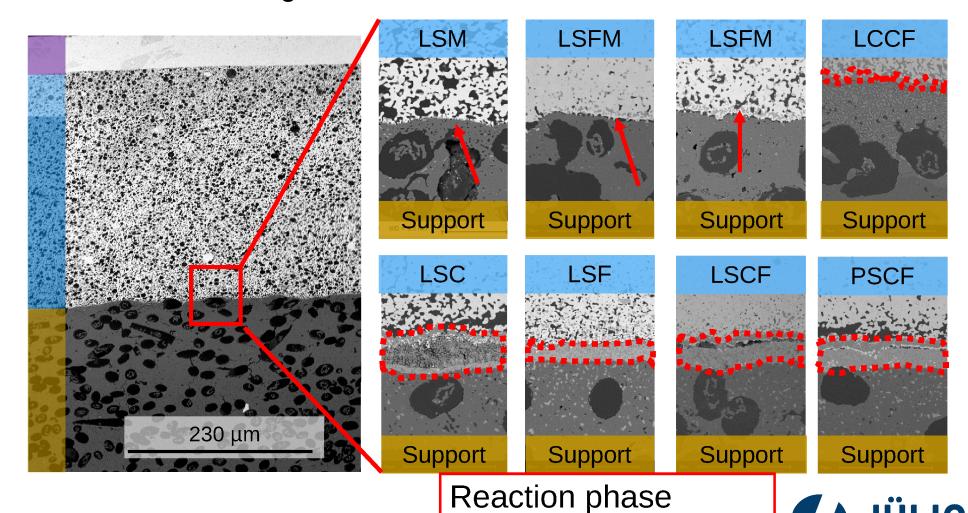
Co-sintered at <1300°C 5h, SEM image

Electrolyte 8YSZ dense

Cathode
La<sub>1-x</sub>Sr<sub>x</sub>MnO<sub>3</sub>
+8YSZ
porous

Current collector La<sub>1-x</sub>Sr<sub>x</sub>MnO<sub>3</sub> porous

> Support Mg, Si, Zn Porous



segregations / layers

Forschungszentrum

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### REACTIVITY SUPPORT SUBSTRATE AND CATHODES

Co-sintered at <1300°C 5h, SEM image

Zn diffuses "far" into LSM cathode >200µm Forms Mn-Zn spinel [1]

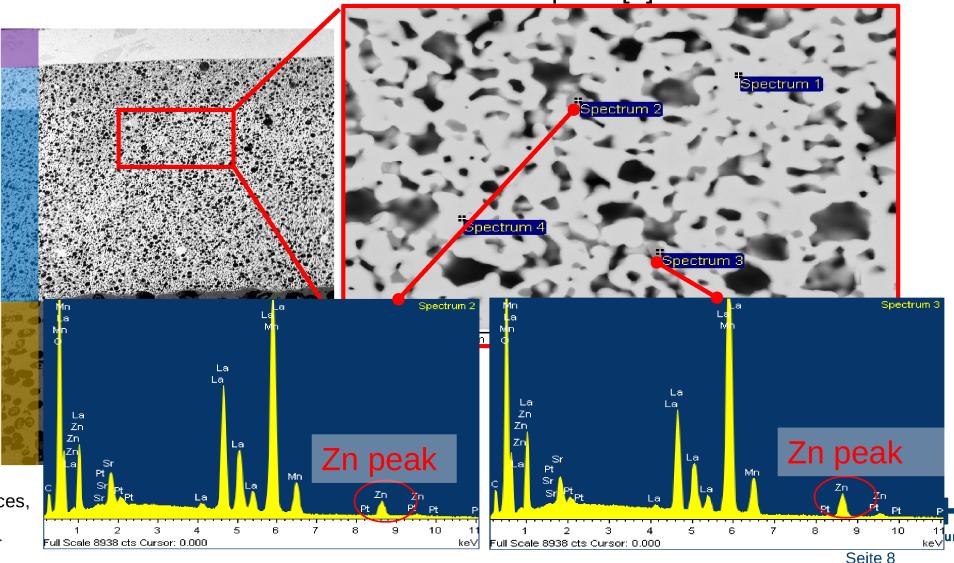
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> Support Mg, Si, Zn Porous

[1] Matté et al. J. Power Sources, 2018



### GAS DIFFUSION TEST: X-RAY PHOTOELECTRON SPECTROSCOPY ANALYSIS

- Detected Zn on LSM/8YSZ:
  - Gas diffusion of Zn

Cathode LSM20

+8YSZ Electrolyte 8YSZ

Air gap

Mg-Si-Zn substrate

Annealing <1300°C, 5h

Relative CPS ×10<sup>-3</sup>

Relative CPS ×10<sup>-3</sup>

Sr3d

C1s

C1s

ZrO<sup>2</sup>

Zr3p3

Zr3p3

Zr3p3

LSM/8YSZ side

Post annealing <1300°C, 5h

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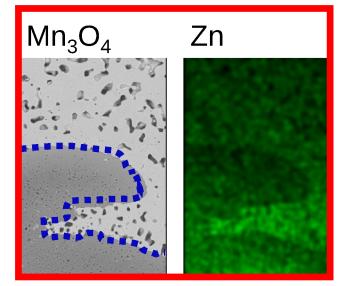
.2 0.4 0.6 0.8 1.9 Binding Energy [keV]

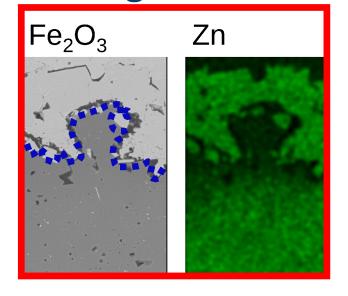


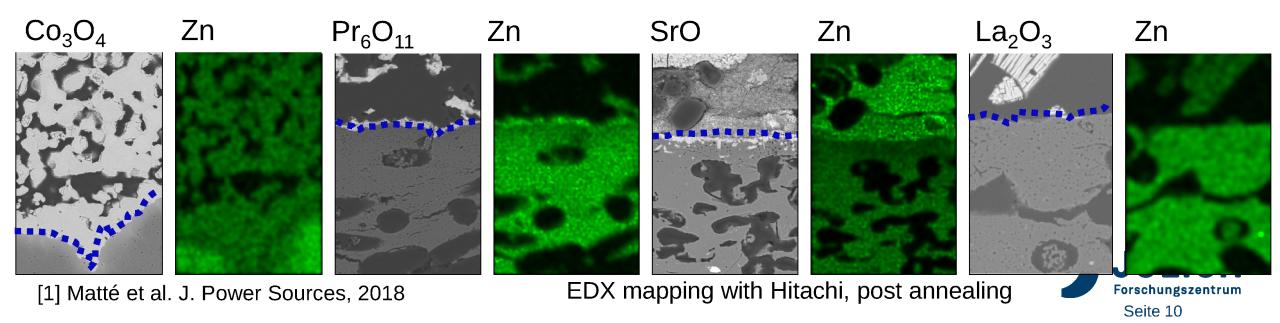
0.2

### REACTIVITY OF SINGLE OXIDES WITH Zn IN Mg-Si-Zn

- Possible getters for Zn identified
- Apparent high reactivity / solubility with Zn of single oxides
- Zn and Mn forms spinel phase [1]
  - ➤ Mn<sub>3</sub>O<sub>4</sub> and Fe<sub>2</sub>O<sub>3</sub>chosen for tests in LSM





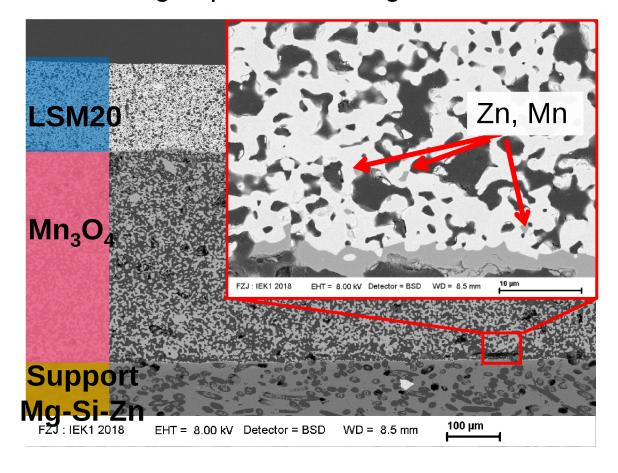


### **LSM: TEST WITH GETTER LAYERS**

#### Zn diffusion depths:

- In Mn<sub>3</sub>O<sub>4</sub> ~100μm,
  - no Zn found in LSM >400μm
- In  $Fe_2O_3 \sim 40 \mu m$ ,
  - Zn found in LSM >400μm
- $\rightarrow$  Mn<sub>3</sub>O<sub>4</sub> is suitable Zn getter in LSM

SEM images post annealing 1200°C, 5h





# ELECTROCHEMICAL PERFORMANCE

Introduction

**Chemical** interactions

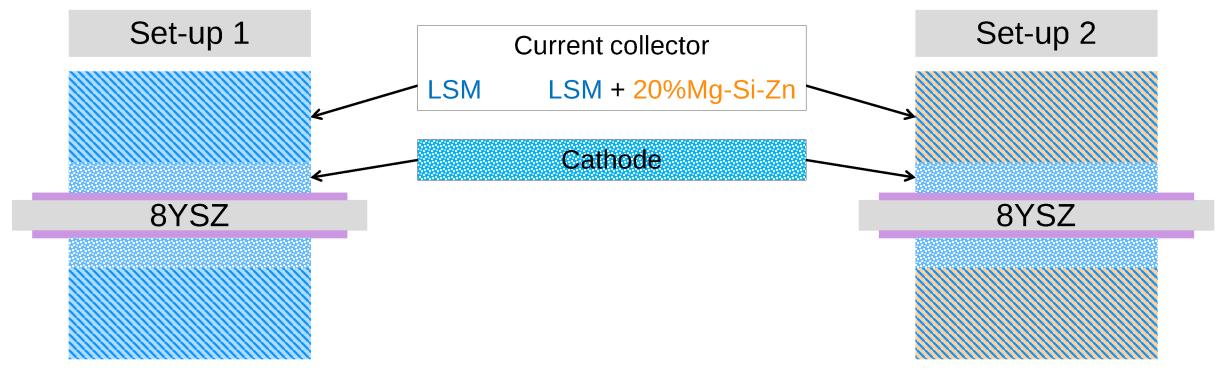
Electrochemical performance

Microstructure

Co-sintering

Summary





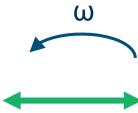
- Symmetric cells co-sintered at <1300°C</li>
- Maximized Mg-Si-Zn content → maximized impact
- Baseline particle size of all cathode powders  $d_{50} \sim 0.8 \mu m$



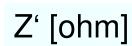
- Electrochemical impedance spectroscopy (EIS)
- Measured 700-800 °C, ω=10<sup>6</sup>-10<sup>-1</sup>Hz
- "Simple benchmarking" value used:
  - Radius of Z' in the Nyquist plot @750 ℃

Nyquist plot

**R**(**Z**') [ohm]

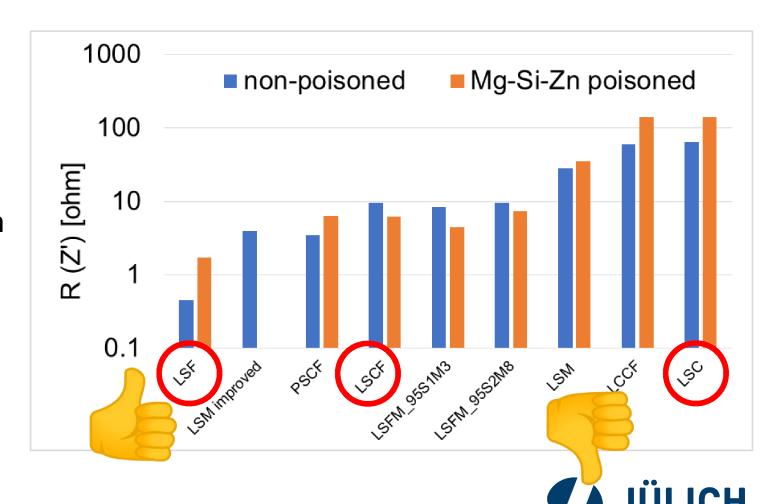


[ohm]





- LSF lowest polarization resistance, LSC the highest
- Unexpected poisoning effects
- Usually LSC and LSCF are high performance cathodes [1]
- Effect of microstructure?



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### MICROSTRUCTURE

Introduction

Chemical interactions

Electrochemical performance

**Microstructure** 

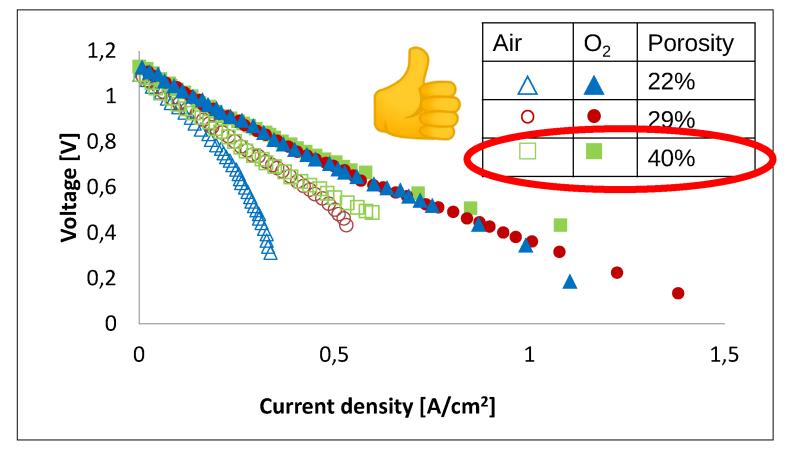
Co-sintering

Summary



### REQUESTED MICROSTRUCTURAL POROSITY

For La<sub>0.8</sub>Sr<sub>0.2</sub>MnO<sub>3</sub>+8YSZ cathode



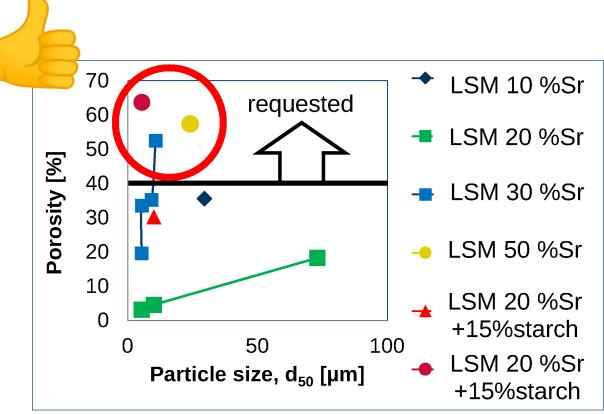
Best cathode performance found at 40% porosity [1]



### **EVALUATION OF MICROSTRUCTURAL PARAMETERS**

### For $(La_{1-x}Sr_x)_yMnO_3$

- Porosity of ≥40% at co-sintering <1300°C with:</li>
  - various combinations of:
    - Sr-content
    - particle size
    - pore formers
- Sintered porosity increases with Sr-content (x≤0.5) [1, 2]
  - Sintering mass transport: La/Sr- lattice vacancy diffusion controlled [3, 4]
  - La/Sr- lattice vacancies decrease with increasing Sr-content [2, 5]



Co-sintered at <1300°C 5h

Porosity: image analysis

Particle size d<sub>50</sub>: LA-950 V2 Retsch Horiba



<sup>[1]</sup> Roosmalen et al., Solid State Ionics, 1993. [2] Wolfenstine et al., Solid State Ionics, 1996.

<sup>[3]</sup> Wolfenstine et al., Journal of Materials Research, 1996. [4] Palcut, Journal of Physical Chemistry C, 2007.

<sup>[5]</sup> Takeda et al., Material Research Bulletin, 1991.

### CO-SINTERING

Introduction Chemical Electrochemical performance Microstructure Co-sintering Summary



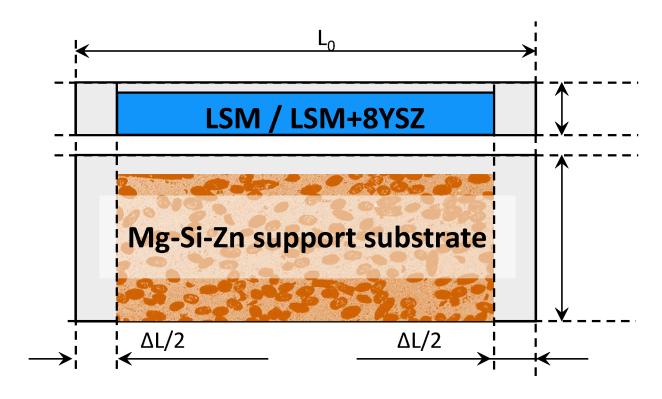
### FREE SINTERING: LSM+8YSZ CATHODE, SUPPORT SUBSTRATE

Free sintering strain behavior depends on:

- Powder (paste) properties
- Temperature

#### Material optimization:

- Matching of time and temperature dependant strain rate  $\dot{\epsilon}_{\rm T}$  or
- Adaptation of rheological properties



True strain rate: 
$$\dot{\epsilon}_T = \frac{\partial}{\partial t} \log(1 + \Delta L/L_0)$$



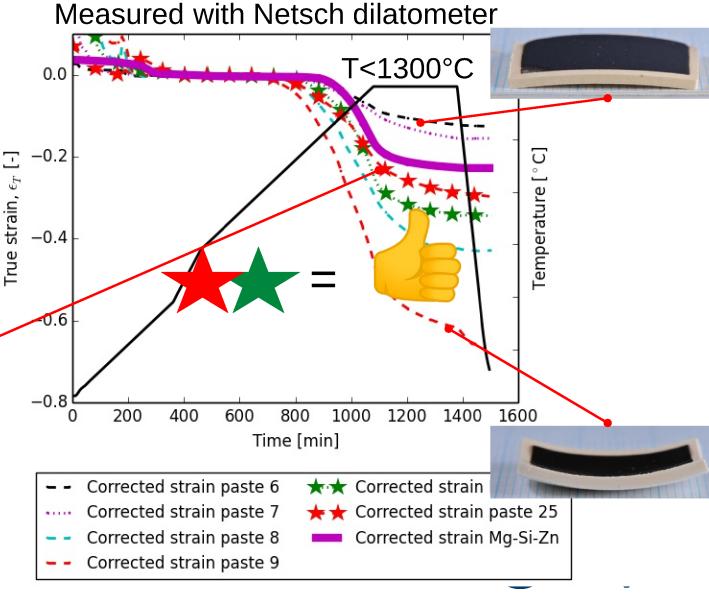
### **CO-SINTERABILITY PREREQUISITE**

LSM/8YSZ cathodes and current collectors, sintering behavior adjusted:

- Particle size, Sr-content
- Pore formers added

Matching sintering curves → Successful co-sintering

Cathode Mg-Si-Zn substrate



### SUMMARY

Introduction Chemical Electrochemical Microstructure Co-sintering Summary



### **SUMMARY**

- Novel all ceramic SOFC design; evaluation of cathodes within new design and processing conditions:
- Chemical interactions: All cathodes react with elements in support
  - But, reactions may be decreased / suppressed by getters
- Electrochemical performance: so far LSF shows the highest performance in EIS
- Microstructure: at new processing conditions needs optimization
- Co-sinterability: must be considered during material optimization
- OUTLOOK: Improve microstructure of LSF / LSCF
- Most promising cathodes → single cell measurements



Succesful

with LSM

### **ACKNOWLEDGMENT**

A special thanks to MSc Fabian Grimm for conducting measurements

BMWi
for the funding

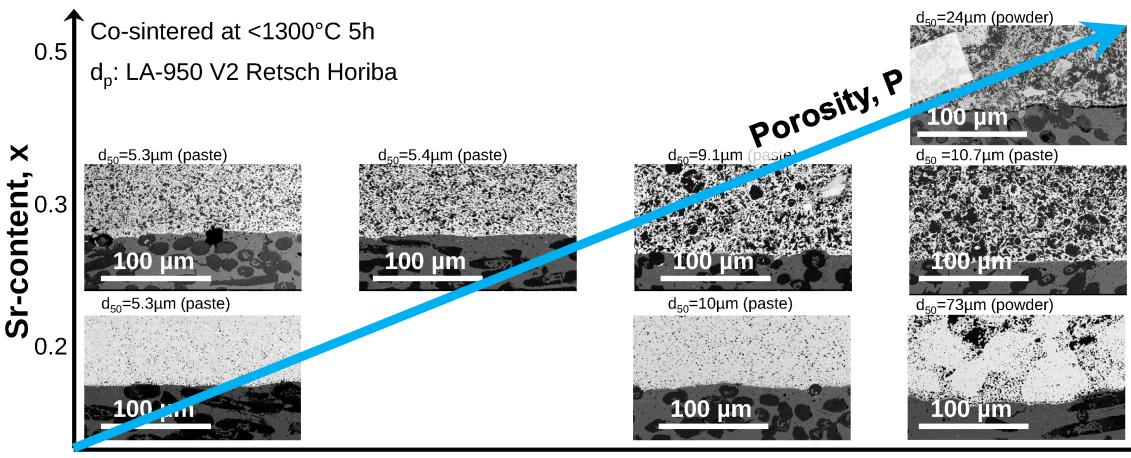
Federal Ministry
for Economic Affairs
and Energy

Project partners for their cooperation **BOSCH** Hochschule Karlsruhe Technik und Wirtschaft **JÜLICH** UNIVERSITY OF APPLIED SCIENCES Forschungszentrum chschule Aalen

Colleagues at
Forschungszentrum Jülich
for support in investigations

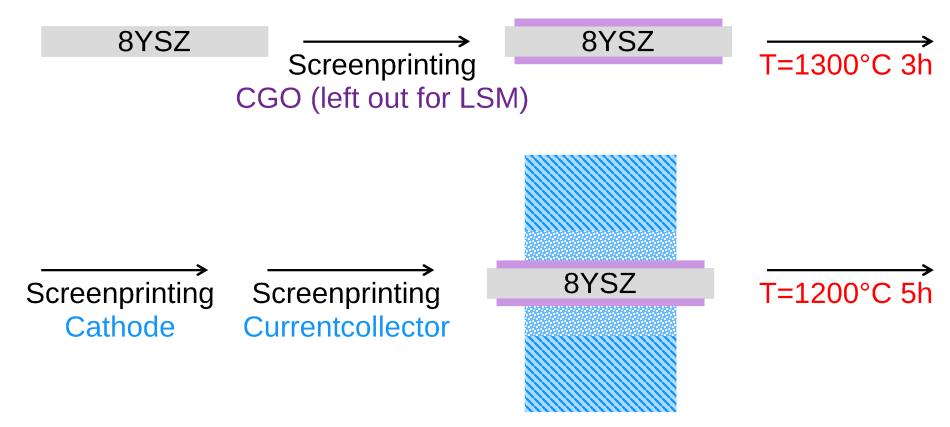


### IMPACT OF PARTICLE SIZE AND Sr-CONTENT ON SINTERED MICROSTRUCTURE (La<sub>1-x</sub>Sr<sub>x</sub>)<sub>y</sub>MnO<sub>3</sub>



**Powder particle size** 





• Impedance-measurements samples: symmetric cells

