



Dissolved organic matter characteristics of deciduous and coniferous forests with variable management: different at the source, aligned in the soil

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Abstract. Dissolved organic matter (DOM) is part of the biogeochemical cycles of carbon and nutrients, carries pollutants and drives soil formation. The DOM concentration and properties along the water flow path through forest ecosystems depend on its sampling location and transformation processes. To improve our understanding of the effects of forest management, especially tree species selection and management intensity, on DOM concentrations and properties of samples from different ecosystem fluxes, we studied throughfall, stemflow, litter leachate and mineral soil solution at 26 forest sites in the three regions of the German Biodiversity Exploratories. We covered forest stands with three management categories (coniferous, deciduous age class and unmanaged beech forests). In water samples from these forests, we monitored DOC concentrations over 4 years and characterized the quality of DOM with UV-vis absorption, fluorescence spectroscopy combined with parallel factor analysis (PARAFAC) and Fourier transform ion cy-

clotron resonance mass spectrometry (FT-ICR-MS). Additionally, we performed incubation-based biodegradation assays. Multivariate statistics revealed strong significant effects of ecosystem fluxes and smaller effects of main tree species on DOM quality. Coniferous forests differed from deciduous forests by showing larger DOC concentrations, more lignin- and protein-like molecules, and fewer tannin-like molecules in throughfall, stemflow, and litter leachate. Cluster analysis of FT-ICR-MS data indicated that DOM compositions, which varied in aboveground samples depending on tree species, become aligned in mineral soil. This alignment of DOM composition along the water flow path in mineral soil is likely caused by microbial production and consumption of DOM in combination with its interaction with the solid phase, producing a characteristic pattern of organic compounds in forest mineral soils. We found similarly pronounced effects of ecosystem fluxes on the biodegradability of DOM, but surprisingly no differences between decid-

uous and coniferous forests. Forest management intensity, mainly determined by biomass extraction, contribution of species, which are not site-adapted, and deadwood mass, did not influence DOC concentrations, DOM composition and properties significantly.

1 Introduction

Dissolved organic matter (DOM) processing and transport is highly dynamic in forest ecosystems (Kaiser and Kalbitz, 2012) and plays an important role in the biogeochemical cycles of carbon and nutrients (Bolan et al., 2011; Kaiser and Kalbitz, 2012). The chemical composition of DOM strongly affects its role in the carbon and nutrient cycles of ecosystems (Bolan et al., 2011). The chemical composition, in turn, depends on the DOM source and its processing along the water flow path through ecosystems.

Following the water path through a forest ecosystem, there are numerous sources and sinks of DOM: rainwater moves through the atmosphere, washes through forest canopy and understory vegetation, infiltrates and percolates the forest litter layer and the organic-matter-rich topsoil, and passes further downward through the deeper mineral soil reaching groundwater tables and entering the aquifer. Precipitation incorporates atmospheric aerosol ingredients like dust and gases, containing organic carbon (Aitkenhead-Peterson et al., 2003). Typical dissolved organic carbon (DOC) concentrations of precipitation range from 0.6 to 7.6 mg L⁻¹ in Europe (Aitkenhead-Peterson et al., 2003). The below-canopy fluxes consist of throughfall and stemflow, both containing DOM of different quality (Moore, 2003; Inamdar et al., 2012; Levia et al., 2012; Levia and Germer, 2015; Michalzik et al., 2016; van Stan and Stubbins, 2018). Organic compounds are released from leaves (Wickland et al., 2007), twigs and tree stems (Levia and Germer, 2015), but also from insects (Michalzik et al., 2016) and bacteria (Lindow and Brandl, 2003; Müller et al., 2006) inhabiting the canopy and leaf surfaces. Important sources of DOM, especially at the soil surface, are decomposition products of leaf litter (Cleveland et al., 2004; Klotzbücher et al., 2013) and deadwood or coarse woody debris (Kahl et al., 2012; Bantle et al., 2014; Magnússon et al., 2016). Major belowground sources of DOM are root exudates (Yano et al., 2000; Baetz and Martinoia, 2014; Tückmantel et al., 2017), microbial primary and secondary metabolites (Aitkenhead-Peterson et al., 2003), and degradation products of soil organic matter (DOM as leftovers of soil organic matter degradation; e.g., Gödde et al., 1996; Hagedorn et al., 2004).

The ecosystem fluxes (throughfall, stemflow, litter leachate and soil solution) in turn are influenced by forest management practices. Thus, the source of DOM is affected by changing the tree composition through partial or complete removal and/or replacement of specific tree species, by ex-

porting biomass and by modifying the proportion of deadwood (Goldmann et al., 2015; Augusto et al., 2002). Various studies under laboratory and field conditions showed differences in litter leachate DOC concentrations, DOM biodegradability and compositions for different tree species (Cleveland et al., 2004; Don and Kalbitz, 2005; Klotzbücher et al., 2013; Cuss and Guéguen, 2013). In the mineral soil, the chemical composition of root exudates appears to be species-specific and hence the microbial rhizosphere community associated with each plant species is different (Van Dam and Bouwmeester, 2016). Changing the amount and species of deadwood influences fungal community composition, wood decomposition and release of DOM quantity and quality (Arnstadt et al., 2016).

Both, sources and processing, affect DOM chemical composition (Stubbins et al., 2017). Biological DOM production and mineralization are important mechanisms regulating DOM dynamics in the environment (Benner, 2002; Bolan et al., 2011). Biodegradability of DOM is, in addition to other controls, driven by intrinsic characteristics like molecular structure, functional group content or size of the molecules (Marschner and Kalbitz, 2003). During DOM transformation and mineralization by microorganisms, several classes of chemical compounds are preferentially oxidized to CO₂ (e.g., carbohydrates), while others passively accumulate as leftovers, e.g., lignin, lipids and waxes (Kalbitz et al., 2003). Similarly, some fractions of DOM are sorbed more strongly by components of the solid soil (minerals, organic matter), systematically changing the DOM quality (e.g., Kaiser et al., 1996).

We hypothesized (i) that the composition and the biodegradability of DOM changes from a dominance of nonaromatic nitrogen-rich compounds of high bioavailability to highly aromatic, increasingly oxidized, nitrogen-poor compounds with decreased bioavailability along the water flow path through forest ecosystems, from throughfall (TF), stemflow (SF) and litter layer leachate (LL) to mineral topsoil (TOP) and subsoil (SUB) solution. We postulated (ii) that aboveground changes in DOC concentrations and DOM composition are mainly controlled by selective biological degradation, while changes in mineral soil are governed by sorption to mineral surfaces. Moreover, we hypothesized (iii) that the dominant tree species as well as forest management intensity affect the DOM composition as well as the direction and magnitude of its changes along the water flow path: the former because of the presence of species-specific compounds in DOM, like phenolic secondary metabolites in beech forests, and the latter, measured as the Forest Management Index (ForMI), in addition to others because of its influence on the biomass production and C input into the soil (Kahl and Bauhus, 2014).

To test these hypotheses, we assessed the quality, structural composition and bioavailability of DOM and its concentration measured as DOC in 26 differently managed forests in three regions in Germany. We characterized DOM

quality using a combination of indices derived from UV-vis absorbance, fluorescence components derived from parallel factor analysis (PARAFAC) of fluorescence–excitation–emission matrices (EEMs), and molecular formulae obtained with high-resolution Fourier transform ion cyclotron resonance mass spectrometry (FT-ICR-MS). Additionally, we assessed DOM biodegradability in an incubation experiment.

2 Material and methods

2.1 Study sites

We conducted the study on experimental plots at the Schwäbische Alb (Alb), the Hainich-Dün (Hainich, HAI) and the Schorfheide-Chorin (Schorfheide, SCH) sites of the German Biodiversity Exploratories, which were established as a platform for large-scale and long-term functional biodiversity research (DFG Schwerpunktprogramm 1374, <http://www.biodiversity-exploratories.de>, last access: 23 March 2019). For sample collection, we selected nine forest plots at each of the Alb (AEW1–AEW9) and Hainich (HEW1–HEW6 and HEW10–HEW12) sites and eight forest plots at the Schorfheide sites (SEW1–SEW3 and SEW5–SEW9). The forests comprise three management categories: (i) unmanaged beech-dominated forests (*Fagus sylvatica* L., for at least 60 years), (ii) beech-dominated (deciduous) age-class forests, and (iii) coniferous age-class forests (spruce-dominated, *Picea abies* L. for Alb and Hainich and pine-dominated, *Pinus sylvestris* L. for Schorfheide). As a measure for forest management intensity, we used the forest management intensity indicator (ForMI) proposed by Kahl and Bauhus (2014). The ForMI is the sum of three management-related factors: the ratio of the harvested to total tree volume, the contribution of not-site-adapted tree species, and the contribution of deadwood volume with saw cuts to the total deadwood mass. Higher ForMI values indicate a higher intensity of forest management. Important climatic and geological information of the three sites is given in Fischer et al. (2010). A summary as well as essential property of the investigated forest plots are given in the Supplement (Table S1). Chemical soil properties for Hainich and Schorfheide plots are given in Table 1.

2.2 Sampling

We collected solution samples every 2 weeks for 2 days from above- and belowground ecosystem fluxes during the vegetation periods from April 2011 to November 2015. We sampled throughfall (TF), stemflow (SF), litter leachate (LL), mineral topsoil (TOP) and subsoil solutions (SUB) as volume-weighted composite samples of multiple individual samplers for each ecosystem flux.

We sampled TF with 20 funnel-type collectors (diameter 0.12 m, polyethylene, PE) per forest ecosystem, which we placed 0.3 m above the soil surface, arranged in two lines of 10 samplers in a cross-shaped form. To minimize alterations of the samples, e.g., by evaporation, photochemical reactions or growth of algae, we wrapped the sampling bottles with aluminium foil and covered the opening of the collection bottle with a 1.6 mm polyester mesh and a table-tennis ball. We sampled SF with sliced polyurethane hoses (diameter: 0.04 m) fixed around tree stems and sealed with a polyurethane-based glue to the bark of three trees per site at approximately 1.5 m in height. The polyurethane hose was connected with a polypropylene or high-density (HD) PE barrel via a PE tube. We collected forest floor litter leachate with three zero-tension lysimeters per site (280 cm² sampling area each) consisting of polyvinyl chloride plates covered with a PE net (mesh width 0.5 mm) connected via PE hoses to 2 L HDPE bottles (Nalgene®) stored in a box belowground. We sampled soil solution with nylon membrane (0.45 µm) and borosilicate glass suction cups (ecoTech, Germany). Three suction cups were installed beneath the A horizon (TOP) at approximately 10 cm in depth. Another three suction cups were installed in the B horizon (SUB) at approximately 50 cm in depth. Because of shallow soils, the installation of suction cups in subsoil was not possible in the Alb plots. Suction cups were connected to 2 L HD-PE bottles (Nalgene®) in an insulated aluminium box placed into a soil pit. We extracted soil water by applying a vacuum of 50 kPa to the HDPE bottles with an electric pump after each sampling.

After recording sample volumes with graded cylinders and merging samples from individual samplers to volume-weighted composite samples per flux and plot in the field, we transported the samples on ice to the laboratory and stored them overnight at 5 °C. In the laboratory, all samples were filtered through cellulose filters (Sartorius, Germany, grade 292) on the following day. We washed the filters with 100 mL of deionized water and 10 mL of sample prior to filtration of the remaining sample and froze all filtered samples at –18 °C until further analysis. Preliminary tests showed that freezing the samples decreased the measured DOC concentration by 5 % on average and also affected DOM fluorescence (Thieme et al., 2016). However, since the samples of all ecosystem fluxes (TF, SF, LL, TOP, SUB) were affected at a similar magnitude, freezing did not hamper the comparison of samples regarding changes in DOM quality and DOC concentration.

An overview of sampling time and sample composition per analysis is given in Table 2. Pictures of sampling installations are given in the Supplement (Fig. S1). Detailed information of selected plots per site, number of measured samples per ecosystem flux and composition of pooled samples for all measurements is provided in the Supplement (Table S2: DOM characterization: fluorescence; Table S3: DOM characterization: FT-ICR-MS; Table S4: DOM biodegradability).

Table 1. Chemical soil properties and mean dissolved organic carbon (DOC) concentrations of plots in the Hainich-Dün (HEW) and Schorfheide-Chorin (SEW) sites. LL: litter leachate; TOP: topsoil solution; SUB: subsoil solution; reduction cDOC (%): reduction of DOC concentration as a percentage between LL and TOP or TOP and SUB; C_{org} : organic carbon content of soil; Al_0 : aluminum content extracted with ammonium oxalate; Fe_0 : iron content extracted with ammonium oxalate.

Plot	Ecosystem flux/ soil layer	Management category	DOC (mg L^{-1})	Reduction cDOC (%)	C_{org} (g kg^{-1})	Al_0 (g kg^{-1})	Fe_0 (g kg^{-1})	Clay (g kg^{-1})	Texture (KA5*)	pH soil (CaCl ₂)
HEW1	LL	coniferous age class	39.23							
HEW1	TOP	coniferous age class	11.26	71.31	69.14	3.28	3.50	326	Lu	7.0
HEW1	SUB	coniferous age class	15.46	−37.36	28.99	4.38	3.89	239	Uls/Tl	7.5
HEW2	LL	coniferous age class	41.54							
HEW2	TOP	coniferous age class	24.72	40.49	50.60	1.43	4.92	241	Lu/Ut4	4.6
HEW2	SUB	coniferous age class	7.70	68.84	6.95	1.73	2.98	589	Tu2	7.0
HEW3	LL	coniferous age class	66.08							
HEW3	TOP	coniferous age class	16.76	74.64	47.74	2.33	3.18	359	Ut3/Ut2	3.9
HEW3	SUB	coniferous age class	14.04	16.22	10.33	2.38	2.37	634	Tl	6.7
HEW4	LL	deciduous age class	22.76							
HEW5	LL	deciduous age class	18.26							
HEW5	TOP	deciduous age class	7.50	58.92	61.77	3.79	3.19	457	Lu	5.2
HEW5	SUB	deciduous age class	5.12	31.78						7.2
HEW6	LL	deciduous age class	17.57							
HEW6	TOP	deciduous age class	11.30	35.70	34.40	2.19	3.73	214	Lu	4.3
HEW6	SUB	deciduous age class	5.20	54.02	5.15	2.45	3.62	442	Tu2/Tl	5.4
HEW10	LL	unmanaged	24.18							
HEW10	TOP	unmanaged	7.95	67.15	67.59	3.49	4.74	485	Ut4	4.1
HEW11	LL	unmanaged	29.77							
HEW11	TOP	unmanaged	10.96	63.20	58.52	3.31	4.72	404	Ut4	4.9
HEW11	SUB	unmanaged	12.10	−10.41	19.78	3.46	4.32	517	Tu3	4.9
HEW12	LL	unmanaged	24.02							
HEW12	TOP	unmanaged	7.42	69.09	31.13	1.72	2.64	164	Ut4	3.9
HEW12	SUB	unmanaged	5.60	24.52	5.58	2.43	3.19	424	Tu2	5.9
SEW1	LL	coniferous age class	67.07							
SEW1	TOP	coniferous age class	58.63	12.59	18.34	1.82	2.02	5	SI2	3.6
SEW1	SUB	coniferous age class	16.19	72.39	2.06	2.05	2.03	1	SI2	3.9
SEW2	LL	coniferous age class	58.50							
SEW2	TOP	coniferous age class	26.73	54.31	16.99	1.78	1.94	32	SI2	3.5
SEW2	SUB	coniferous age class	11.40	57.34	2.26	2.68	2.50	33	SI2	4.2
SEW3	LL	coniferous age class	57.20							
SEW3	TOP	coniferous age class	37.09	35.15	20.95	1.61	1.62	17	SI2	3.3
SEW3	SUB	coniferous age class	15.06	59.39	4.05	2.09	1.38	3	SI2	4.0
SEW5	LL	deciduous age class	32.84							
SEW5	TOP	deciduous age class	91.81	−179.59	29.56	1.20	1.04	1	SI2	3.1
SEW5	SUB	deciduous age class	27.86	69.65	2.50	2.21	1.29	1	SI2/Su2	3.4
SEW6	LL	deciduous age class	37.84							
SEW6	TOP	deciduous age class	12.84	66.05	31.05	2.39	2.52	23	SI2	3.4
SEW6	SUB	deciduous age class	8.48	34.00	1.45	1.77	1.60	17	SI2	3.9
SEW7	LL	unmanaged	26.20							
SEW7	TOP	unmanaged	46.86	−78.84	24.30	1.74	1.78	1	SI2	3.2
SEW7	SUB	unmanaged	16.89	63.96	6.37	1.38	1.55		SI2	3.7
SEW8	LL	unmanaged	41.33							
SEW8	TOP	unmanaged	29.03	29.76	29.20	1.86	1.58	20	SI2	3.1
SEW8	SUB	unmanaged	13.07	54.97	10.28	1.52	1.48	1	SI2	3.2
SEW9	LL	unmanaged	42.50							
SEW9	TOP	unmanaged	39.94	6.01	22.96	0.95	1.01	18	SI2	3.0
SEW9	SUB	unmanaged	14.92	62.65	4.81	1.43	1.09	1	SI2	3.7

* KA5: Ad-Hoc-Arbeitsgruppe Boden (2005).

Table 2. Overview of the samples used for the various analytical procedures. TF: throughfall; SF: stemflow; LL: litter leachate; TOP: topsoil solution; SUB: subsoil solution.

Part of study	Period of sampling	Sites	Management category (number of investigated plots per site)	Ecosystem fluxes	Number of analyzed samples
DOM characterization					
DOC + fluorescence	April 2011–November 2013	Hainich, Schorfheide	unmanaged (three) deciduous age class (two) coniferous age class (three)	TF, SF, LL, TOP, SUB	466 (79 ^b)
FTICR-MS	April/May 2015	Schorfheide	unmanaged (two) coniferous age class (two)	TF, SF, LL, SUB	eight ^c
DOM biodegradability	October 2012	Alb, Hainich, Schorfheide	unmanaged (three) deciduous age class (three or two) ^a coniferous age class (three)	TF, SF, LL, TF, SF, LL, TOP, SUB	25 ^d

^a Three plots for the Alb site and two plots for Hainich and Schorfheide sites. ^b Mean DOC and fluorescence spectra per plot and ecosystem flux used for all statistical analyses.

^c Pooled solution samples per management category and ecosystem flux. ^d Pooled solution samples per site, management category and ecosystem flux.

2.3 Sample processing for optical and chemical characterization of DOM

We thawed the samples over night at 8 °C and conducted fluorescence and UV-absorption measurements without further preparations. In total we analyzed 466 Hainich and Schorfheide samples of all ecosystem fluxes and management categories taken between 2011 and 2013. To balance uneven sample numbers, we calculated mean EEMs per plot and ecosystem flux resulting in a dataset with 79 EEMs. For FT-ICR-MS analysis, we chose TF, SF, LL and SUB samples from unmanaged beech and coniferous age-class forests of the SCH sites in April and May 2015. To gain enough sample volume for the analysis, we pooled samples from two forest plots per management category gaining a total of eight samples. After re-filtration (0.45 µm, Whatman GF/C), samples were desalted and concentrated using solid phase extraction (SPE, C18 Hydra cartridges, Machery & Nagel, Düren, Germany) using methanol (≥ 99.98 %, ultra LC-MS grade; Carl Roth, Karlsruhe, Germany) as eluent. After SPE, the solution was dried at room temperature. Before FT-ICR-MS measurements, the samples were redissolved in methanol.

2.4 Sample processing to assess the biodegradability of DOM

We used TF, SF and LL samples from plots of all management categories collected in October 2012 to assess the biodegradability of DOM. In this study, we refer to biodegradable dissolved organic carbon (BDOC) as the DOC utilized by heterotrophic microbes via complete mineralization of C sources to obtain energy and by incorporation of carbon into microbial biomass. For each management category and ecosystem flux, we pooled samples from two to three forests per site, gaining a total number of 25 composite

samples. We filtered the samples through a 0.2 µm Vacuflow filter in a laminar flow box beside a Bunsen burner and transferred 40 mL of the filtrate to sterile 250 mL suspension culture flasks (Greiner Bio-One, Frickenhausen, Germany). After adding 2 mL of bacterial inoculum, we closed the flasks with semipermeable caps. We incubated each sample in triplicate for seven time intervals (0, 3, 6, 10, 14, 20 and 28 days) at 20 °C in the dark. Following the incubation, we filtered the samples through sterile 60 mL Soft-Ject single-use syringes (Henke-Sass, Wolf; Tuttlingen, Germany) equipped with nylon syringe filters with a pore size of 0.45 µm (Rotilabo, Carl Roth; Karlsruhe, Germany). The filtered samples were stored frozen until the measurement of DOC concentrations and the UV-vis and fluorescence spectra.

We prepared the bacteria inoculum by collecting and merging soil samples from forests of each site. We combined moist sieved field fresh soil from the first 10 cm after removing the litter layer of all three sites with unfiltered TF solution of the same sites with a soil : solution ratio of 1 : 10, subsequently shook the mixture for 30 min and centrifuged it for 10 min (Heraeus Megafuge 16, Thermo Scientific; Waltham, USA). We stored the supernatant at 8 °C prior to incubation.

2.5 Measurement of DOC concentrations, UV-vis absorption and fluorescence spectra

We measured DOC concentrations (routine limit of quantification: 3 mg L⁻¹) as non-purgeable organic carbon (NPOC) on a Shimadzu TOC-VCPH analyzer (Shimadzu, Duisburg, Germany). Absorption spectra of DOM were recorded for wavelengths ranging from 400 to 600 nm using a Lambda 20 UV-vis spectrometer (Perkin Elmer, USA) equipped with a 1 cm quartz cuvette. Measurements were baseline-corrected using ultrapure water and all sample spectra were blank

subtracted (ultrapure water, EVOQUA, Warrendale, USA). Fluorescence excitation emission matrices (EEMs) were recorded on a Hitachi F-4500 fluorescence spectrometer (Hitachi, Japan) directly after absorption measurement in the same cuvette. We used excitation wavelengths ranging from 240 to 450 nm (5 nm steps) and emission wavelengths ranging from 300 to 600 nm (2 nm steps) with a slit width of 5 nm and scan speed of 12 000 nm min⁻¹. We corrected our EEMs according to the protocol of Murphy et al. (2010) with the *fdomcorrect* function in the *drEEM* toolbox (Murphy et al., 2013) using MATLAB (2015). For the excitation and emission correction factors, we used the supplies provided by the manufacturer. We measured ultrapure water fluorescence spectra for blank correction and for converting EEMs to Raman units by normalizing them to the area under the Raman peak at a 350 nm excitation wavelength. We diluted the samples with ultrapure water to ensure an absorption < 0.3 at 254 nm (Ohno, 2002) and subsequently performed the inner-filter correction, again using the *fdomcorrect* function in the *drEEM* toolbox (Lakowicz, 2006).

Using the absorbance spectra, we calculated specific ultraviolet absorbance (SUVA₂₅₄) as the absorbance at 254 nm per meter of pathlength of light, divided by the concentration of DOC in milligrams per liter, reported in liters per milligram per meter. The SUVA₂₅₄ index reflects the bulk aromaticity of DOM (Weishaar et al., 2003).

2.6 DOM characterization using FT-ICR-MS and UV absorption

Ultra-high-resolution mass spectra were acquired using an ESI LTQ FT ultra-high-performance instrument (Thermo Fisher Scientific, San Jose, CA, USA) equipped with a 7 T supra-conducting magnet (Oxford Instruments, Abingdon, UK). The mass spectrometer was used in negative mode, tuned daily and calibrated following a standard optimization procedure for almost all settings. Hence, the settings of the ion optics typically varied slightly from day to day. Samples were analyzed within 3 days as pure methanol solution without any pH modification or water addition. Typical standard conditions were: spray voltage of 2.9 kV, capillary voltage of -50 V and tube lens of -93 V. The best performances for our sample set were received when sheath, auxiliary and sweep gas were turned off. The transfer capillary temperature was set to 275 °C. Samples were introduced into the ESI source with a syringe pump at a rate of 5 µL min⁻¹. Mass spectra in profile mode were recorded in full scan from 200 to 1000 Da, measured at resolving 400 000 at *m/z* 400 Da (for complete separation of CHONS- from ¹³C₁CHOS in even-numbered peaks). Each individual mass spectrum contained 50 transients. The automatic gain control target in the ICR cell was set to 5 × E5 (for nearly negligible interactions between the ions) to achieve deviations considerably below 1 ppm (supplier specification). Six spectra were averaged for improving the statistical robustness of the final spectra that were further

processed. The mean deviation of the raw spectra was approximately 0.4 ppm at *m/z* 400 Da; therefore all files were recalibrated before calculation (to prevent two possible assignments as CHO and CHOS₂ for the same peak, which would lead to excluding this mass from further consideration and therefore loss of information). Prior to and between some analyses, blanks were measured.

For quality control, all peaks of at least two randomly selected masses (odd and subsequent even numbered) were characterized by hand to control the exactness of the recalculated peaks and to set constraints in the calculation program as followed: C, H and O: unlimited; N and S: 0–3 (without the combination S_{>1}N₃); ¹³C: 0–1 and P=0. Molecular formulae were assigned using an in-house-developed post-processing Scilab routine (Scilab Enterprises, Rungis, France).

2.7 Analysis of fluorescence and FT-ICR-MS data

To identify the underlying fluorescence components of the DOM, we used parallel factor analysis (PARAFAC) to mathematically decompose the trilinear data of the EEMs (Stedmon et al., 2003). All further preprocessing steps of EEMs, like smoothing of Rayleigh and Raman scatter and normalization, as well as the PARAFAC analysis were conducted with the *drEEM* toolbox (Murphy et al., 2013) in MATLAB (2015). We chose a six-component PARAFAC model (referred to as C1 to C6), visually checked the randomness of residuals and the component spectral loadings, split-half validated the model, and generated the best fit by random initialization. For comparison in statistical analyses, we used the relative percentage distribution of the six PARAFAC components (percentage of the sum of fluorescence of all PARAFAC components) %C1 to %C6 instead of C1 to C6. Identified PARAFAC components were described by comparison with published PARAFAC models, either manually or by using the OpenFluor database (Murphy et al., 2014).

To analyze FT-ICR-MS data, we used van Krevelen plots (van Krevelen, 1950) to visualize and characterize the assigned molecular formulae gained from the raw MS spectra. Therefore, the elemental ratios of oxygen to carbon (O/C) and hydrogen to carbon (H/C) for each formula of CHO compounds were plotted. Depending on the position in the van Krevelen diagram, all assigned formulae can roughly be grouped according to major classes of biopolymers found in natural organic matter like tannin, lignin, lipids, proteins, amino sugars and hydrocarbons. We used the classification according to Sleighter and Hatcher (2007), applying these assignments: lipids (H/C = 1.7–2.25, O/C = 0–0.22), proteins (H/C = 1.5–2.0, O/C = 0.2–0.5), amino sugars, (H/C = 1.5–1.75, O/C = 0.55–0.7), carbohydrates, (H/C = 1.5–2.0, O/C = 0.7–1.0), lignin (H/C = 0.75–1.5, O/C = 0.2–0.6), tannins (H/C = 0.5–1.25, O/C = 0.6–0.95) and condensed hydrocarbons (H/C = 0.2–0.75, O/C = 0–0.7). The number of formulae in each class were then

summed and normalized by the total number of assignable formulae for all functional groups to produce a relative abundance (as percent) for the six classes of biopolymers (Tfaily et al., 2015). Additionally, we conducted a cluster analysis with the standardized peak intensities of assigned formulae using Jaccard's distances and Ward's method in R (R Core Team, 2015, vegan package; Oksanen et al., 2015) according to Ide et al. (2017) and Stubbins et al. (2017).

With a correlation analysis (Spearman's rank order correlation, stats package in R; R Core Team, 2018), we linked the modeled PARAFAC components with the biochemical information resulting from FT-ICR-MS measurements. Here, we used the relative abundances of PARAFAC components and the relative abundances of biopolymers extracted from van Krevelen plots.

2.8 Effect of ecosystem flux, tree species, and management on DOM composition and biodegradability: calculations and statistical analysis

We used permutational multivariate analyses of variance (vegan package in R; Oksanen et al., 2015; Euclidean distances) to assess the effect of ecosystem flux (TF, SF, LL, TOP, SUB), main tree species (deciduous or coniferous), management intensity (ForMI) and their interactions on DOM composition (PARAFAC components, SUVA₂₅₄). DOC concentration values were not included to separately investigate effects of the drivers of DOM composition and DOC quantity. To visualize the PERMANOVA results, we conducted a PCA (vegan package in R; Oksanen et al., 2015) with the same DOM composition variables. For the PCA, we scaled all DOM composition variables to reach unit standard deviation.

We used a type II ANOVA (car package in R; Fox and Weisberg, 2011) with interaction to test whether ecosystem flux, main tree species or management intensity affected DOC concentration (model Df = 19, residual Df = 59). Here, we log-transformed DOC concentrations to improve normal distribution and homoscedasticity of the residuals.

We conducted univariate pairwise tests to assess effects of ecosystem flux for each of the PARAFAC components, separately for deciduous and coniferous forests. Moreover, we tested separately for management categories (deciduous age-class, beech unmanaged and coniferous age-class forests) if the ecosystem flux had an effect on DOC concentration and SUVA₂₅₄. Finally, for DOC concentration and SUVA₂₅₄, we assessed pairwise differences of main tree species and management category for each of the ecosystem fluxes. If normal distribution of the residuals was given, we used pairwise *t* tests with the Holm–Bonferroni correction (stats package in R; R Core Team, 2018); otherwise, we applied Nemenyi–Damico–Wolfe–Dunn tests (Monte Carlo test variant with 50 000 iterations, coin package in R; Hothorn et al., 2006).

To describe the degradation kinetic of our DOM samples, we fitted a single exponential model. We quantified the rate of biodegradation by the mineralization constant (*k*) based on measurements of DOC concentrations measured during the entire incubation period. Changes in DOM composition during degradation were assessed by projecting the six components of the PARAFAC model on the EEMs from samples measured before and after 28 days of incubation.

With a paired PERMANOVA (vegan package in R; Oksanen et al., 2015) we tested the effect of incubation, ecosystem fluxes (TF, SF, LL, TOP, SUB) and management category (deciduous age class, beech unmanaged and coniferous age class) and their interactions on DOC concentration, SUVA₂₅₄ and %PARAFAC components of incubated samples. Subsequently, we used Wilcoxon rank sum tests as paired test (stats package in R; R Core Team, 2018) to evaluate the effect of incubation on DOC concentrations, SUVA₂₅₄ and %PARAFAC values. Finally, we used Spearman's rank order correlation (stats package in R; R Core Team, 2018) to assess the relationships between all variables (%BDOC, *k*, SUVA₂₅₄, %PARAFAC).

3 Results

3.1 Drivers of DOC concentrations and SUVA₂₅₄ in the solution samples from different ecosystem fluxes

Mean DOC concentrations varied among water samples collected from different ecosystem fluxes and depending on main tree species (Fig. 1). Following the water flow path, mean concentrations of solutions from unmanaged beech and deciduous age-class forests roughly increased from TF (9 and 8 mg L⁻¹) via SF (18 and 28 mg L⁻¹) to LL (31 and 26 mg L⁻¹). They remained similar in TOP (24 and 31 mg L⁻¹) and decreased to SUB (13 and 12 mg L⁻¹) samples. Mean DOC concentration in coniferous age-class forests reached a maximum in SF (90 mg L⁻¹) and decreased continuously via LL (55 mg L⁻¹) and TOP (30 mg L⁻¹) to SUB (13 mg L⁻¹). The ANOVA showed a significant effect of ecosystem fluxes and main tree species on DOC concentrations (ANOVA, *p* < 0.001), but management intensity (ForMI) had no significant effect. Comparing the DOC concentrations of all ecosystem fluxes within each management category (beech unmanaged, deciduous age class and coniferous age class), only few statistical significant differences were found (Fig. 1). We found no differences in DOC concentrations of all ecosystem fluxes between the differently managed beech forests. DOC concentrations in TF, SF and LL from beech forests were significantly smaller than those from coniferous forests.

Mean SUVA₂₅₄ values (indicative of the aromaticity of the DOM) were similar for all ecosystem fluxes except LL independent of management category. Mean values for TF, SF, TOP and SUB were 1.6–2.6 L mg⁻¹ m⁻¹ with coniferous SF

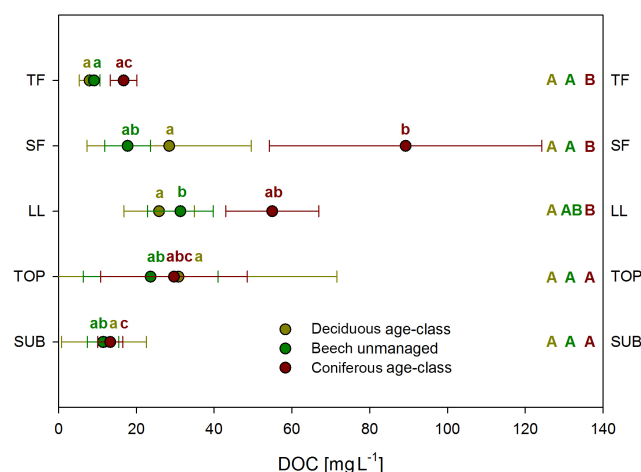


Figure 1. Mean DOC concentrations in ecosystem fluxes (TF, SF, LL, TOP, SUB) grouped according to management categories (deciduous age class, beech unmanaged, coniferous age class). Whiskers show standard deviations. TF: throughfall, $n = 201, 224, 244$; SF: stemflow, $n = 140, 207, 140$; LL: litter leachate, $n = 179, 199, 203$; TOP: topsoil solution, $n = 60, 87, 47$; SUB: subsoil solution, $n = 63, 56, 65$. Capital letters (reading horizontally): differences between management categories; lowercase (reading vertically): differences between water samples collected from different ecosystem fluxes within the same management category.

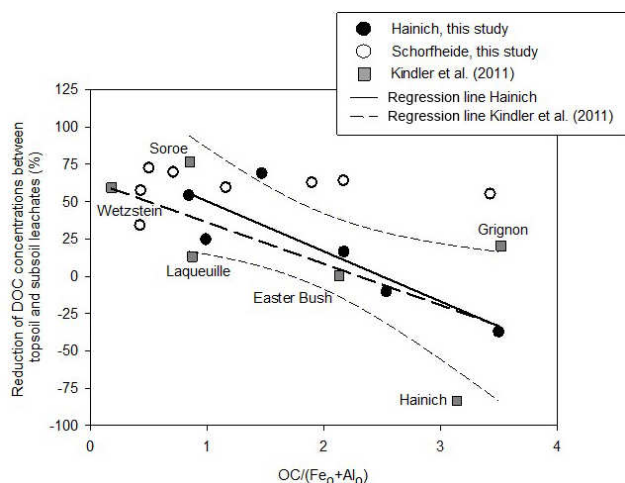


Figure 2. Percentage reduction of DOC concentrations between topsoil leachates (TOP) and subsoil leachates (SUB) in relation to the carbon saturation of pedogenic Fe and Al (hydr)oxides. For the Hainich sites of this study the reduction of DOC concentrations significantly decreases with increasing $OC/(Fe_0 + Al_0)$ ratio (reduction = $84\% - 34 \cdot OC/(Fe_0 + Al_0)$; $p = 0.027$, $R = 0.86$). The relative increase in DOC concentrations at high OC surface loadings was likely caused by a passive enrichment of remaining soil water with DOC due to water withdrawal by evapotranspiration. The names refer to the sites of the Kindler et al. (2011) study.

rising up to $2.9 \text{ L mg}^{-1} \text{ m}^{-1}$. Significantly higher $SUVA_{254}$ values ($p < 0.05$) for LL samples compared with all other sample types equalled $3.5\text{--}3.7 \text{ L mg}^{-1} \text{ m}^{-1}$.

3.2 FT-ICR-MS characterization of DOM composition

The FT-ICR-MS spectra revealed differences in the distribution and abundance of organic molecules of varying mass and composition between ecosystem fluxes and management categories (Figs. 3 and 4).

The numbers of assigned formulae in coniferous forest samples were similar for water samples of all different ecosystem fluxes, ranging between 8126 and 9522. In contrast, we found a slightly higher number of assigned formulae for LL (10 112) and SUB (13 447) samples from unmanaged beech forests compared to TF (9878) and SF (5435) samples.

Elemental formulae of CHO compounds for all samples plotted as van Krevelen diagrams revealed distinct differences between coniferous (pine) and unmanaged beech forests for solution samples collected from all aboveground ecosystem fluxes (Fig. 5). While van Krevelen plots for all pine forest samples exhibited a distinct share of formulae with a H/C ratio of 1.2–1.6 and a O/C ratio of 0.3–0.6, there was a lack of them in the aboveground beech forest samples. The space covered in the diagrams by DOM compositions of the different tree species became increasingly aligned following the water path (Fig. 5).

Depending on their position in the van Krevelen diagram, we assigned molecular formulae to seven major biomolecular classes according to Sleighter and Hatcher (2007). Comparing their relative abundances between water samples collected from varying ecosystem fluxes and main tree species (Table 3), we found distinct differences for both. While lignin-like formulae were the dominant molecules in all ecosystem fluxes of coniferous forest DOM (50 %–66 %), we found almost balanced shares of lignin- and tannin-like molecules for TF (20 %–35 %) and SF (39 %–40 %) of unmanaged beech forest DOM. The share of tannin-like molecules generally increased from TF via SF to LL samples and decreased again to SUB samples independent of main tree species. The share of tannin-like molecules reached up to 70 % in beech forests and only up to 27 % in pine forests. The other compounds like protein-like compounds, lipid-like compounds, amino-sugar-type compounds and carbohydrate-like compounds hardly contributed to the total molecular composition. Only condensed hydrocarbons had additional, noticeable shares of molecule composition for pine and beech TF samples (15 % and 36 %, Table 3).

Cluster analysis with the numbers of molecules assigned to major groups of biomolecules showed three distinct clusters. One included the subsoil solution samples of both, the pine and beech forest stands in the Schorfheide, the second all remaining solution samples from other ecosystem fluxes of pine forests and the third the same for beech forests (Fig. 6).

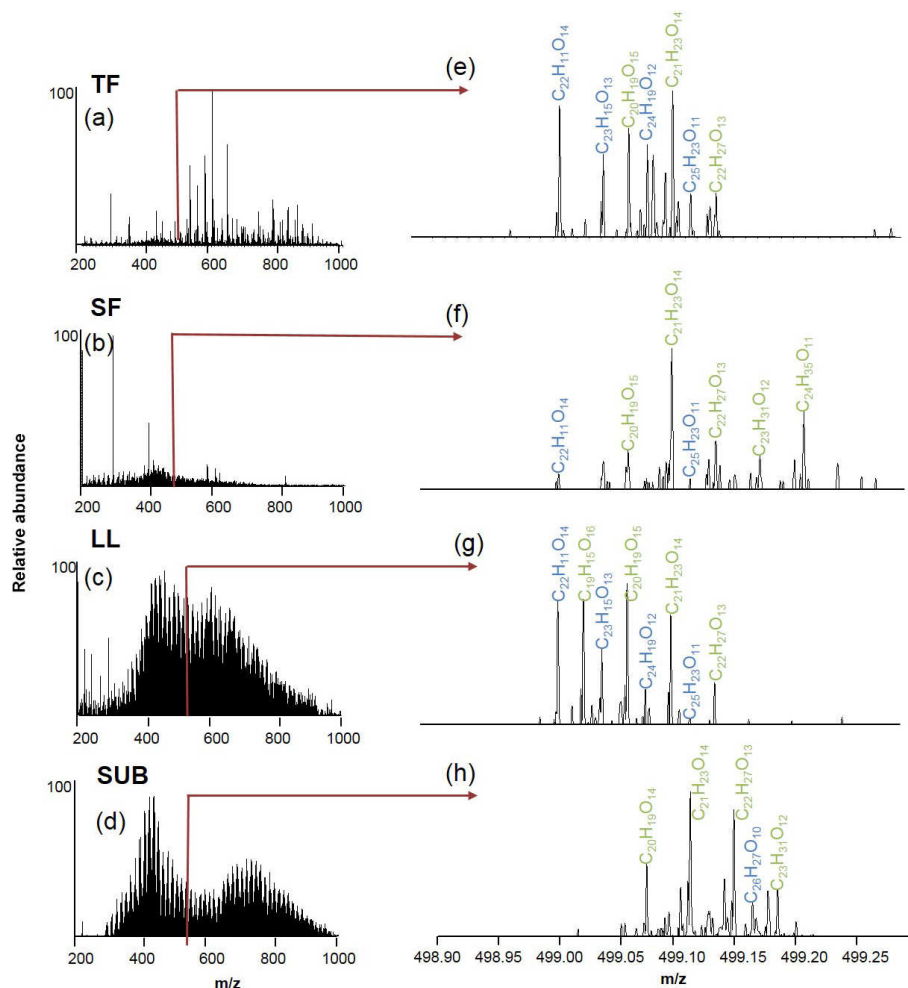


Figure 3. Electrospray ionization Fourier transformation ion cyclotron resonance mass spectra (ESI-FT-ICR-MS) of coniferous (pine) age-class forest samples from the Schorfheide (a–d) and details for m/z 499 (e–h). Assigned molecular formulae are in green and blue. TF: throughfall; SF: stemflow; LL: litter leachate; SUB: subsoil solution.

3.3 PARAFAC components – description and correlation with biochemical compounds

Analyzing the fluorescence samples collected from 2010 to 2013 (see Table 1), we validated a six-component PARAFAC model describing the variation in the fluorescence of DOM. The components were referred to as C1 to C6. Two fluorescence components (C1 and C6) had single excitation and emission maxima, whereas the other four components (C2 to C5) showed two local excitation maxima alongside one emission maximum. Component C1 was characterized by an excitation maximum < 250 nm and an emission maximum at 436 nm. C2 showed two peaks of local excitation maxima at 265 and 375 nm, having an emission maximum at 480 nm. C3 exhibited two local excitation maxima, one at wavelengths < 250 nm and the second at a wavelength of 315 nm, combined with an emission maximum at 404 nm. C4 showed two local excitation maxima at wavelengths < 250 nm and

at a wavelength of 325 nm, with an emission maximum at 446 nm. The fourth component with two local excitation maxima (< 250 and 350 nm) was C5, which showed an emission maximum at a wavelength of 428 nm. The fluorescence of component C6 was characterized by an excitation maximum at 280 nm and an emission maximum at 334 nm. For detailed spectra of all PARAFAC components see the Supplement (Fig. S2).

We applied the previously validated six-component PARAFAC model to the fluorescence spectra of the DOM samples that were also characterized using FT-ICR-MS spectra (see Table 1) to explore the molecular chemical background of the underlying fluorescence patterns. We found a significant positive correlation (Spearman's rho, $p < 0.05$) between the relative contribution of the fluorescence component C2 and the relative number of tannin-like molecules identified by mass spectrometry. Significant negative correlations occurred between %C2 and the fraction of identified

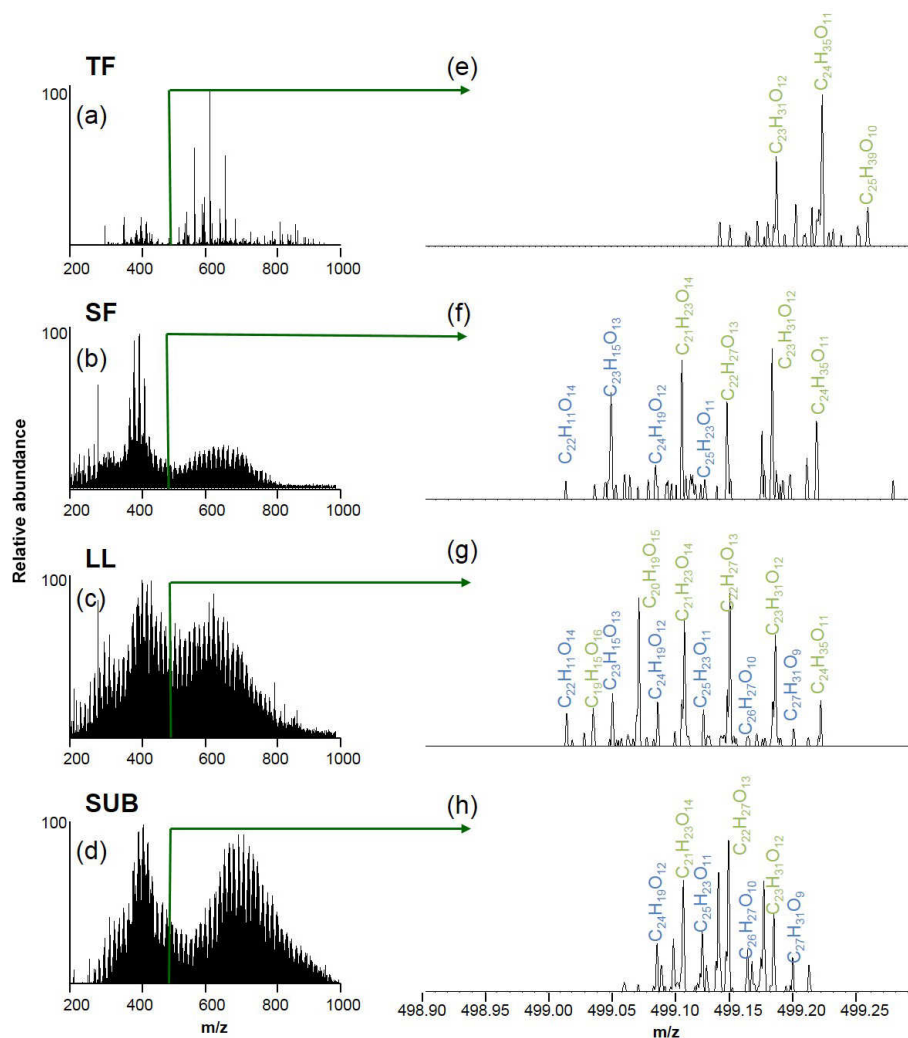


Figure 4. Raw electrospray ionization Fourier transformation ion cyclotron resonance mass spectra (ESI FT-ICR-MS) of unmanaged beech forest samples from the Schorheide (a–d) and details for m/z 499 (e–h). Assigned molecular formulae are in green and blue. TF: throughfall; SF: stemflow; LL: litter leachate; SUB: subsoil solution.

protein-like and amino-sugar-like molecules (Table 4). The relative contribution of fluorescence component C3 to overall fluorescence significantly and positively correlated with the fraction of molecules assigned to the class of lignin-like biopolymers, while a significant negative correlation (Spearman's rho, $p < 0.05$) was observed with the fraction of tannin-like molecules (Table 4). The relative contribution of PARAFAC component C6 to overall fluorescence positively correlated with the fraction of protein-like and amino-sugar-like molecules (Table 4).

While the relevance of fluorescence components C2 and C4 for the overall fluorescence intensity increased with increasing DOC concentrations of the undiluted original samples, the contribution of fluorescence component C1 decreased with increasing DOC concentrations (Table 4).

3.4 Drivers of the PARAFAC components

Considering the lack of significant differences between unmanaged and age-class beech-dominated forests for DOC concentrations and SUVA₂₄₅ values, we focused on comparing deciduous and coniferous forests. With a mean share of 32 %–39 %, component C1 dominated the overall fluorescence of DOM samples from both forests (Fig. 7). Comparing water samples from different ecosystem fluxes for shares of %C1 in between those two forest categories, we only found significant differences for LL samples (Wilcoxon test, $p < 0.05$). The mean contribution of tannin-like components C2 ranged from 12 % to 23 % of total fluorescence and differed significantly between samples of aboveground ecosystem fluxes of deciduous and coniferous forests, with samples from the former showing a larger share of C2 to total fluorescence than samples from coniferous forests (Wilcoxon

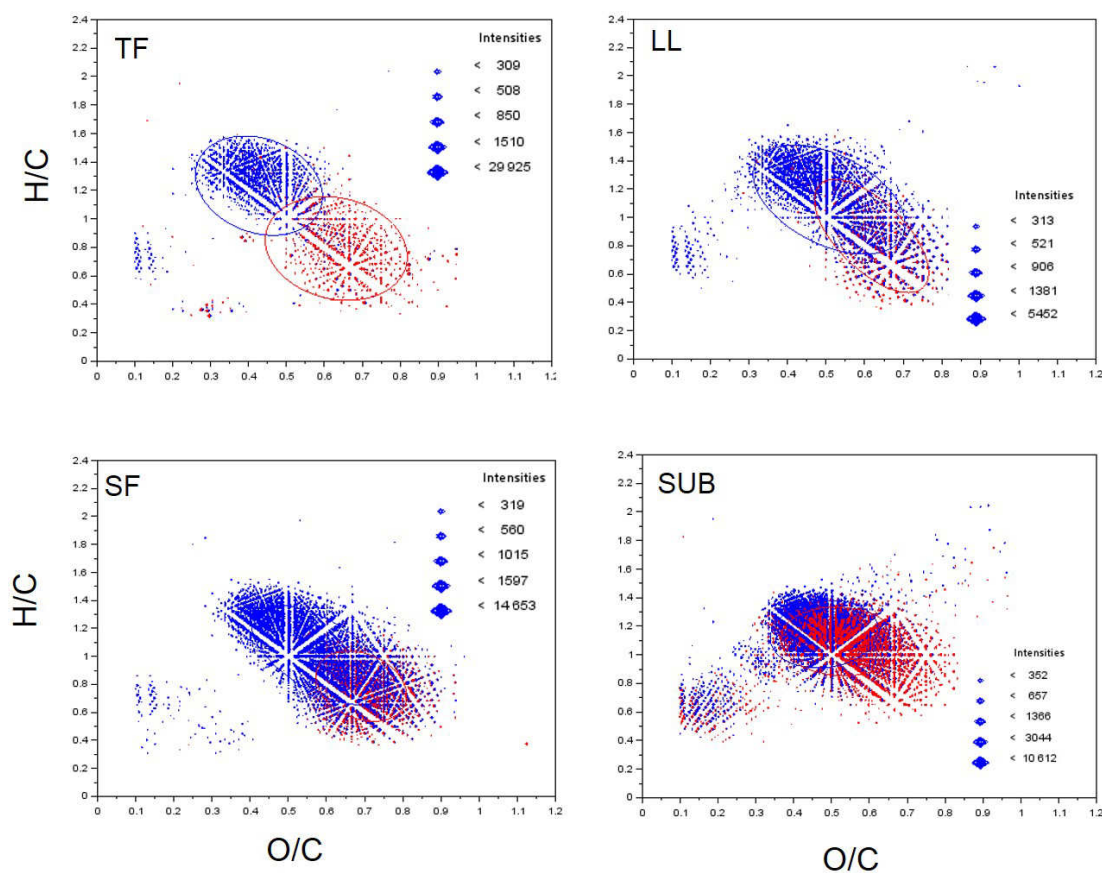


Figure 5. Van Krevelen plots of CHO compounds for unmanaged beech (red) and coniferous (pine, blue) forest DOM samples. Ellipsoids indicate space covered by DOM samples. TF: throughfall; SF: stemflow; LL: litter leachate; SUB: subsoil solution.

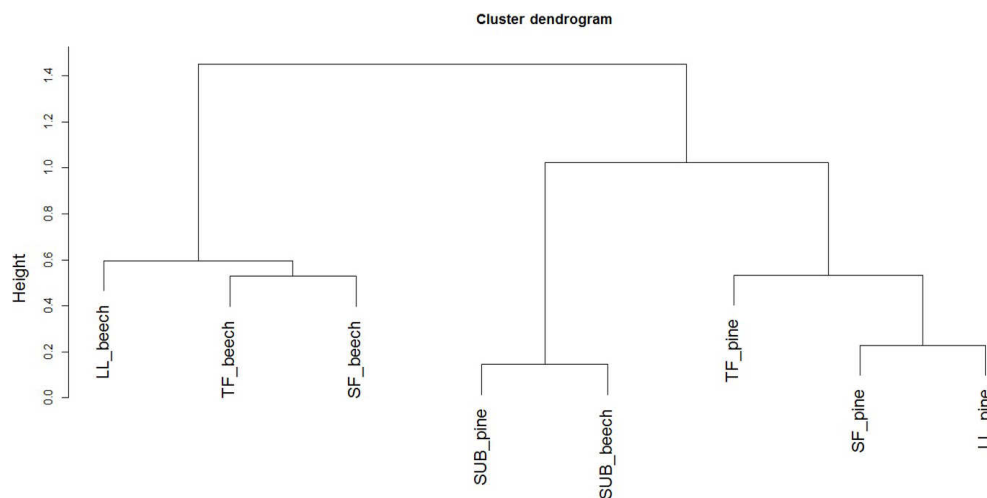


Figure 6. Cluster dendrogram of the number of molecules assigned to major groups of biomolecules (tannin-like, lignin-like, lipid-like, protein-like, amino-sugar-like, and hydrocarbon-like) according to Sleighter and Hatcher (2007) obtained from FTI-CR mass spectra of DOM samples from ecosystem fluxes of coniferous (pine) and unmanaged beech forest from Schorfheide sites. TF: throughfall; SF: stemflow; LL: litter leachate; SUB: subsoil solution.

Table 3. Number of formulae (relative shares) assigned to major groups of biomolecules according to Sleighter and Hatcher (2007) obtained from FT-ICR mass spectra of DOM samples from ecosystem fluxes of coniferous (pine) and unmanaged beech forest from the Schorfheide. TF: throughfall; SF: stemflow; LL: litter leachate; SUB: subsoil solution.

Biomolecular groups	Ecosystem flux	Formulae within each ecosystem flux			
		Coniferous forest		Unmanaged beech forest	
Lignin-like	TF	840	(53 %)	194	(20 %)
	SF	1173	(50 %)	229	(39 %)
	LL	1088	(59 %)	108	(14 %)
	SUB	2735	(66 %)	2619	(63 %)
Tannin-like	TF	96	(6 %)	345	(35 %)
	SF	309	(13 %)	231	(40 %)
	LL	503	(27 %)	583	(77 %)
	SUB	205	(5 %)	512	(12 %)
Protein-like	TF	74	(5 %)	5	(1 %)
	SF	98	(4 %)	39	(7 %)
	LL	24	(1 %)	0	(0 %)
	SUB	67	(2 %)	48	(1 %)
Amino-sugar-like	TF	17	(1 %)	3	(0 %)
	SF	35	(2 %)	10	(2 %)
	LL	8	(0 %)	0	(0 %)
	SUB	27	(1 %)	16	(0 %)
Lipid-like	TF	0	(0 %)	2	(0 %)
	SF	7	(0 %)	1	(0 %)
	LL	0	(0 %)	0	(0 %)
	SUB	5	(0 %)	3	(0 %)
Cellulose-like	TF	1	(0 %)	1	(0 %)
	SF	21	(1 %)	2	(0 %)
	LL	4	(0 %)	0	(0 %)
	SUB	53	(1 %)	52	(1 %)
Condensed-hydrocarbon-like	TF	235	(15 %)	358	(36 %)
	SF	45	(2 %)	30	(5 %)
	LL	21	(1 %)	49	(6 %)
	SUB	89	(2 %)	145	(4 %)

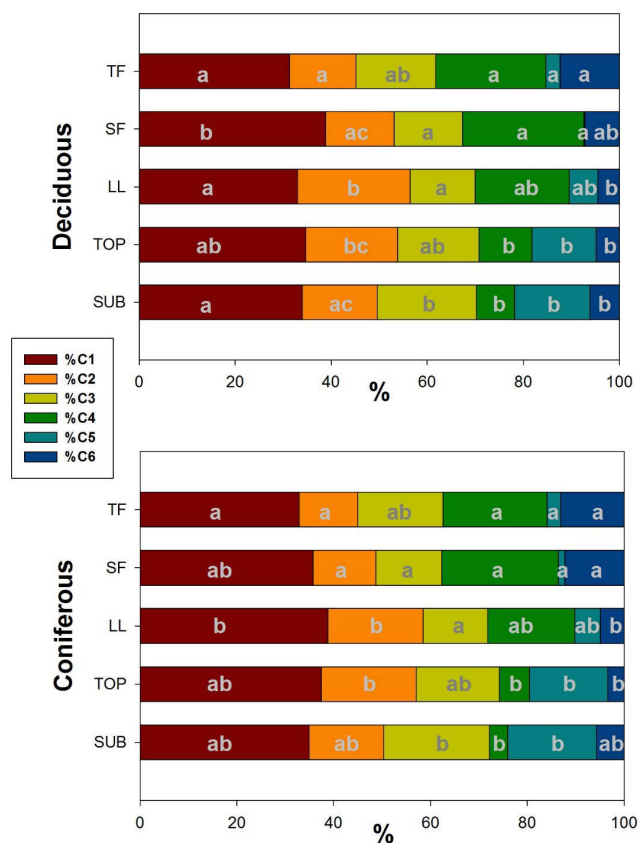
test, $p < 0.05$). In contrast, the mean contribution of lignin-like C3 to total fluorescence (13 %–22 %) was similar for both forest categories. Component C4 contributed between 4 % and 25 % to total fluorescence and showed significant differences between forest categories only for water samples collected from belowground ecosystem fluxes. Fluorescence component C5 ranged from 0 % to 18 % and similar to protein-like component C6 (3 %–13 %) showed significant differences between deciduous and coniferous forests only for SF samples.

When comparing the distribution of single PARAFAC components between samples from different ecosystem fluxes along the water flow path within each management category, we found for %C1 the smallest shares in TF samples increasing to maximum shares in SF and LL samples and again slightly decreasing from TOP to SUB samples. A sim-

ilar trend was observed for the contribution of the tannin-like component %C2 but reaching maximum shares in LL and TOP samples before decreasing again in SUB. The lignin-like fluorescence component %C3 showed an opposite trend to %C2, with the smallest contributions to total fluorescence in LL samples, increasing again via TOP to reach its maximum contribution in SUB samples (Fig. 7). We found a decreasing mean contribution of component %C4 from TF to SUB samples interrupted by a slight increase in SF samples. The reverse trend was found for the fluorescence component C5. The mean share of the protein-like component %C6 of total fluorescence was largest in TF samples. This share decreased along the flow path in LL and TOP to slightly increase SUB samples.

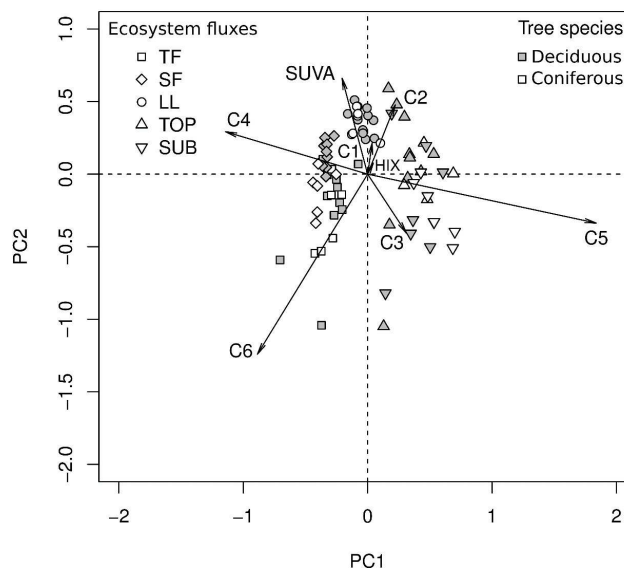
Table 4. Spearman's rho for the correlation between the percentage relative abundances of PARAFAC components (%C1–%C6) and the relative abundances of biopolymers extracted from FT-ICR-MS van Krevelen plots. Significance level: * $p < 0.05$; ** $p < 0.01$.

	%C1	%C2	%C3	%C4	%C5	%C6
DOC	−0.67*	0.73*	−0.37	0.71*	−0.15	−0.01
Lipid-like	0.6	−0.39	0.07	0.15	−0.34	0.48
Protein-like	0.43	−0.72*	0.27	0.34	−0.53	0.74*
Amino-sugar-like	0.45	−0.64*	0.21	0.42	−0.56	0.74*
Lignin-like	0.05	−0.5	0.80**	−0.41	0.31	0.18
Tannin-like	−0.29	0.72*	−0.66*	0.26	0.01	−0.49

**Figure 7.** Mean distribution of PARAFAC components in samples from different ecosystem fluxes of deciduous (age class and unmanaged) and coniferous forests. Letters (reading vertically) indicate differences between samples from different ecosystem fluxes regarding PARAFAC components within each management category (Nemenyi–Damico–Wolfe–Dunn tests). TF: throughfall; SF: stemflow; LL: litter leachate; TOP: topsoil solution; SUB: subsoil solution.

3.5 Drivers of spectroscopic DOM composition (absorbance and fluorescence)

To comprehensively assess the drivers of DOM composition we combined our absorbance and fluorescence spectroscopic results. We found a significant effect of ecosystem

**Figure 8.** PCA plot of DOM composition variables (SUVA₂₅₄, PARAFAC components C1–C6). TF: throughfall; SF: stemflow; LL: litter leachate; TOP: topsoil solution; SUB: subsoil solution. Variables $n = 7$; samples $n = 79$.

fluxes on DOM composition variables including SUVA₂₅₄ and %PARAFAC components (PERMANOVA, $p = 0.001$) explaining 67 % of sample variance. Further, a significant, albeit small, effect ($R^2 = 0.01$) was found for the main tree species (PERMANOVA, $p = 0.04$). When investigating the individual ecosystem fluxes in detail, significant differences (Wilcoxon test, $p < 0.05$) were found for samples from aboveground ecosystem fluxes between coniferous and deciduous forest stands, especially for %C2 (tannin-like) but not for %C3 (lignin-like) and %C4. Prominent differences disappeared when following the water underground, except for %C4 for which significant differences appeared. No significant effects were found for management intensity alone (PERMANOVA, $p = 0.964$).

A PCA illustrated the distinctly different DOM composition in the water samples collected from various ecosystem fluxes (Fig. 8). The first two components identified by the PCA explained 88 % of the total variance (PC1: 60 %, PC2: 28 %).

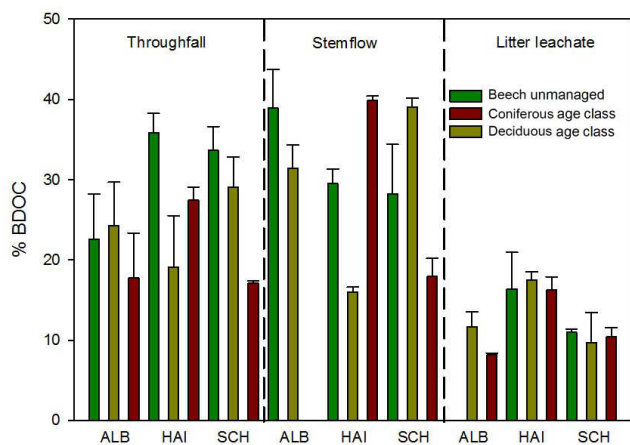


Figure 9. Percentage of biodegradable DOC (%BDOC) after 28 days of incubation in samples collected from different ecosystem fluxes. Bar chart: mean and SD of three replicates. TF: throughfall; SF: stemflow; LL: litter leachate.

PC2: 28 %). TF and SF were closely grouped together and differentiated from TOP and SUB along PC1, based most strongly on the different contributions of C4 and C5 to overall fluorescence. LL was separated in particular from TF and SUB samples along PC2, based predominantly on their larger SUVA₂₅₄ and smaller contribution of C6 to overall fluorescence (Fig. 8).

3.6 DOM biodegradability

We found a significant (Wilcoxon rank sum test, $p < 0.001$) decrease in DOC concentrations with increasing time of incubation for all samples. The decrease could be adequately described using a two-parameter single exponential model. Calculated degradation rate constants (k) were significantly different from zero for all samples and ranged between 0.004 and 0.021 d⁻¹. SF proved to be the samples with the highest extent (Table S5) and rate of DOC degradation followed by TF with slightly lower values (Fig. 9). In SF samples 15 %–40 % and in TF samples 17 %–35 % of initial DOC was degraded within 28 days. With 8 %–18 % of BDOC, LL samples showed 2 times lower values of degradation and up to 10 times lower rate constants than SF and TF. No significant differences for %BDOC and k were found between coniferous and deciduous forests.

SUVA₂₅₄ values showed a significant increase during the incubation (Wilcoxon rank sum test, $p < 0.001$) for water samples from all ecosystem fluxes. The mean increase was lowest for TF (0.5 L mg⁻¹ m⁻¹) and similar for SF and LL (1.0 L mg⁻¹ m⁻¹). We found a significant negative correlation between %BDOC and SUVA₂₅₄ (Spearman's rho, $p < 0.05$).

We applied the previously validated six-component PARAFAC model to the EEMs of samples measured be-

fore and after 28 days of incubation. A significant increase after 28 days was found for %C1 for TF and SF samples (Wilcoxon rank sum test, $p < 0.01$) but not for LL samples. We found a significant decrease in %C3 and %C4 during the incubation for TF samples only (Wilcoxon rank sum test, $p < 0.01$). Although SUVA₂₅₄ was positively correlated with PARAFAC components %C1 and %C2, and negatively correlated with %C3 and %C6, no correlations were found between %BDOC and these PARAFAC components.

4 Discussion

Change in DOC concentrations and DOM composition along the water flow path and among different forest management categories

Mean DOC concentrations of water passing through the forest ecosystems in our study followed concentrations reported in previous studies. For TF, DOC concentrations documented in the literature ranged from 2 to 35 mg L⁻¹ (Michalzik et al., 2001; Moore, 2003; Stubbins et al., 2017), for SF from 12 to 95 mg L⁻¹ (Moore, 2003; Levia et al., 2012; Stubbins et al., 2017) and for LL from 14 to 90 mg L⁻¹ (Michalzik et al., 2001; Ide et al., 2017; Stubbins et al., 2017). Investigating soil solutions, others reported DOC concentrations of 7–40 mg L⁻¹ for topsoil (Moore, 2003; Fellman et al., 2008b; Kindler et al., 2011; Ide et al., 2017) and 2–5 mg L⁻¹ for subsoil solutions (Michalzik et al., 2001; Peichl et al., 2007; Kindler et al., 2011). This pattern indicates that water is enriched in DOM during aboveground ecosystem passage and depleted while passing through mineral soil horizons.

In the study of Kindler et al. (2011), the retention of DOC in mineral soil, expressed as percentage reduction of downward DOC flux, was closely related to the ratio between organic carbon content of the mineral soil and the sum of its oxalate-extractable Fe and Al content (OC/(Fe₀ + Al₀)). This suggested that the DOC retention in mineral soils is governed by the sorption to the surfaces of Fe and Al (hydr)oxides. Because Fe and Al (hydr)oxides have a limited sorption capacity for organic matter, DOC retention in subsoils decreased exponentially with increasing organic matter coverage of the hydroxide surfaces (Kindler et al., 2011). In contrast to the findings of Kindler et al. (2011), we compare DOC concentrations, not fluxes. In order to test whether the data of our study fit the findings of Kindler et al. (2011), we plotted changes in DOC concentrations reported by Kindler et al. (2011) together with the data of this study against the ratio of OC/(Fe₀ + Al₀) in one graph (Fig. 2). Different from fluxes, which always decreased with increasing soil depth in the Kindler et al. (2011) study, DOC concentrations increased with increasing depth at the Hainich sites with the highest OC/(Fe₀ + Al₀) ratios of all study regions (Fig. 2). This increase in concentrations can be explained by a concentration effect because of evapotranspiration, in the case

that the DOC sorption capacity of pedogenic Fe and Al (hydr)oxides is saturated. Overall, the retention of DOC in the Hainich soils of this study fitted well to the DOC retention in the European dataset of Kindler et al. (2011), which showed that the regional variation in DOC retention can be as large as the variation at continental scale. The reduction of DOC concentrations between TOP and SUB significantly decreased with increasing $\text{OC}/(\text{Fe}_0 + \text{Al}_0)$ ratio ($p = 0.027$; Fig. 2), corroborating the hypothesis that sorption to pedogenic Fe and Al (hydr)oxides controlled DOC retention in mineral soils (Kindler et al., 2011). However, the results for mineral soils of the Schorfheide sites did not follow this pattern, as DOC concentrations decreased from TOP to SUB by 33 %–72 % regardless of the $\text{OC}/(\text{Fe}_0 + \text{Al}_0)$ ratio (Fig. 2). At the Schorfheide sites, processes other than sorption to Fe and Al (hydr)oxide surfaces likely governed DOC retention. The Schorfheide soils developed from fluvioglacial quartzitic sands covering carbonate-free glacial till. Because of their poor pH buffering capacity, these soils were very acidic ($\text{pH}_{\text{CaCl}_2} = 3.0\text{--}3.6$ in topsoils). The mean pH value in soil water samples of the Schorfheide sites was 4.5 in TOP solutions, increasing to 5.5 in SUB solutions. This means that Al (hydr)oxides were dissolved in the Schorfheide topsoils, increasing Al^{3+} concentrations in soil water and leachates. The pH increase to 5.5 along the way from TOP to SUB likely induced a re-precipitation of Al. We assume that dissolved organic matter transported from TOP to SUB co-precipitated together with Al^{3+} as described by Nierop et al. (2002) and Jansen et al. (2003, 2005) for acidic sandy soils from the Netherlands. If DOM was immobilized as insoluble metal-organic matter precipitate in B horizons, no limitation by available sorption sites of surfaces of pedogenic (hydr)oxides would apply, so that reductions of DOC concentrations with increasing depth in mineral soil would be independent of the soil $\text{OC}/(\text{Fe}_0 + \text{Al}_0)$ saturation index.

Consistent with findings in other studies, SUVA_{254} values of our DOM samples ranged from 1.8 to $4.7 \text{ L mg}^{-1} \text{ m}^{-1}$ for TF (Peichl et al., 2007; Inamdar et al., 2012; Stubbins et al., 2017), from 1.9 to $11.2 \text{ L mg}^{-1} \text{ m}^{-1}$ for SF (Levia et al., 2012; Stubbins et al., 2017) and between 2.7 and $5.2 \text{ L mg}^{-1} \text{ m}^{-1}$ for LL (Peichl et al., 2007; Inamdar et al., 2012). This indicates an increasing share of aromatic DOM compounds when passing through the above-ground forest ecosystem. Reported ranges for topsoil solutions ($2.2\text{--}3.9 \text{ L mg}^{-1} \text{ m}^{-1}$) and for subsoil samples ($1.4\text{--}2.7 \text{ L mg}^{-1} \text{ m}^{-1}$) are again similar to our findings (Peichl et al., 2007; Fellman et al., 2008b; Inamdar et al., 2012). The decrease in SUVA_{254} values of DOM during the mineral soil passage could be related to preferential sorption of the aromatic DOM fractions (Kaiser and Guggenberger, 2000; Peichl et al., 2007).

Our ESI FT-ICR-MS measurements of forest DOM samples generated spectra with thousands of peaks, the number and distribution of which were comparable with previous studies of natural DOM samples (e.g., Stenson et al., 2003;

Sleighter et al., 2010; Tfaily et al., 2015). Due to the ultrahigh mass resolution, combined with the exactness in the sub parts per million range, it is possible to assign molecular formulae unique to almost all of the detected masses. The molecular composition of single peaks in our forest DOM samples led to their classification in typical biomolecular groups (lignin-like, tannin-like, condensed-hydrocarbon-like, protein-like, amino-sugar-like, lipid-like and carbohydrate-like) and was similar to those reported by others for TF, SF, LL and subsoil solution samples (Tfaily et al., 2015; Ide et al., 2017; Stubbins et al., 2017). Consistent with other studies of DOM samples from aboveground ecosystems fluxes in oak and cedar forests (Stubbins et al., 2017) and in wetlands (Hertkorn et al., 2016) and a fulvic acid isolated from small lake in Antarctica (D'Andrilli et al., 2013), CHO-only compounds were the main fraction of assigned molecules.

Using the OpenFluor database (Murphy et al., 2014), we found close matches of component C1 with fluorescence components from studies in various environments characterized as “humic-like with terrestrial origin” (Santos et al., 2010; Yamashita et al., 2010; Kothawala et al., 2012; Shutova et al., 2014; Dainard et al., 2015). Studies by Stedmon et al. (2003) in a Danish estuary and by Lambert et al. (2016) with Congo River water found components with spectra matching our component C2. They also described this component as humic-like with terrestrial origin. The positive correlation with the number of m/z peaks assigned to tannin-like compounds based on their position in the van Krevelen plots ($\rho = 0.75$, Table 4) along with the high contribution of C2 to the fluorescence found in LL and TOP (Fig. 7) indicated that component C2 contained plant-derived, tannin-like components. Component C3 resembled components described as “microbially altered humic material” (Murphy et al., 2011), which were, among others, found in humic substances from sediments, in fen and bog pore water, and in lakes, streams and estuaries (Santín et al., 2009; Shutova et al., 2014; Tfaily et al., 2015; Osburn et al., 2016). C3 showed excitation and emission wavelengths ($\lambda_{\text{ex}} = 250, 300 \text{ nm}$; $\lambda_{\text{em}} = 400 \text{ nm}$) similar to those published in studies investigating fluorescence of lignin from different sources (e.g., Thruston Jr., 1970; Albinsson et al., 1999). We found a significant positive correlation of the contribution of C3 to total fluorescence with the number of m/z peaks assigned to lignin-like compounds detected using FT-ICR-MS ($\rho = 0.80$, Table 4). Therefore, we suggest that the share of C3 to total fluorescence reflected lignin and lignin-derived degradation products in DOM. As another humic-like component, C4 was termed “C peak” by Coble (1996), which matched the fluorescence components found by Kothawala et al. (2012) studying Swedish lakes as well as by studies investigating river and lake water (Lambert et al., 2016; Osburn et al., 2016). The humic-like component C5 only matched a component in the OpenFluor database that was reported by Lambert et al. (2016) studying Congo River water. The component C5 also falls into the excitation and emission range

of a component described as “humic-like C” by Coble et al. (2014) with sources referred also as humic and terrestrial. The fluorescence of component C6 was similar to the fluorescence of tryptophan and was therefore described as protein-like, representing fluorescence of free amino acids and such bound in proteins. The component was included in numerous PARAFAC models of fluorescence of DOM from various environments (e.g., Murphy et al., 2013; Yu et al., 2015). The positive correlation between the protein-like as well as amino-sugar-like fraction of FT-ICR-MS data and %C6 ($\rho = 0.74$, Table 4) seems to confirm that protein-like fluorescence represented the fluorescence of proteins.

Our results showed distinct differences in DOC concentrations and DOM properties between solution samples from different forest ecosystem fluxes. TF was enriched in DOC ($9\text{--}17\text{ mg L}^{-1}$) relative to precipitation ($2\text{--}5\text{ mg L}^{-1}$) measured at the same sites during the same sampling period. In line with other studies (Peichl et al., 2007; Inamdar et al., 2012), low values for optical DOM properties like SUVA₂₅₄ and humic PARAFAC components C1 and C2 indicated a less humic-like and less aromatic DOM composition in TF compared with the other aboveground ecosystem fluxes. According to this interpretation, we would expect low percentages of molecules assigned to the lignin, tannin and condensed hydrocarbon fractions gained by FT-ICR-MS analysis of TF samples. However, this was only found for tannin-like not for lignin-like compounds or condensed hydrocarbons (Table 3).

Elevated shares of condensed hydrocarbons in TF compared with the other ecosystem fluxes (Table 3) agreed with findings of Stubbins et al. (2017) studying oak and cedar TF and SF samples. They suggested that atmospheric deposition of combustion products (primary aerosol) in combination with their reaction products (secondary aerosol) caused this noticeable fraction in DOM, due to accumulation from atmospheric aerosols on leaf surfaces. Combustion products have been shown to contribute to the designated condensed hydrocarbon fraction in the van Krevelen diagram (Kim et al., 2003, 2004).

TF samples were also richest in N-containing compounds, as shown by the highest relative contribution of component C6 (Fig. 7) and of the protein-associated fraction of FT-ICR-MS molecules (Table 3) of all ecosystem fluxes. In addition to free and bound proteins, atmospheric trace gases of bio- and anthropogenic origin react preferentially in the night with NO₃ radicals, generating nitrogen organic compounds (Ervens and Volkamer, 2010; Farmer et al., 2010; Lee et al., 2016; Ng et al., 2017) that could deposit in the (wet) canopy. This observation is also in line with findings in other studies (Inamdar et al., 2012; Ide et al., 2017).

While following the water passage downward into subsoil layers, the decreasing DOC concentrations and SUVA₂₄₅ values, as well as decreasing percentages of tannin-like compounds (Table 3), were in line with a preferential sorption of aromatic, polyphenolic DOM in mineral soils (Kaiser and

Guggenberger, 2000; Avneri-Katz et al., 2017). We had expected the fraction of molecules assigned to lignin in the van Krevelen plot to follow this behavior. Contrarily, we found increasing concentrations of lignin-like compounds with increasing soil depth, which were sorbed less strongly in the mineral phase than the highly oxidized compounds associated with tannin-like molecules. Additional evidence of the ongoing microbial processing of DOM along the flow path is the increasing share of the microbial-derived PARAFAC component C3 (Fig. 7). The accumulation of lignin-like compounds may also be explained by a different interpretation of the van Krevelen diagram. It was suggested by others that the space covered by lignin molecules in the van Krevelen diagram should not only be linked to higher plant source material, but also to other types of compounds proposed to be refractory, including nonaromatic compounds like carboxylic-rich alicyclic molecules (CRAM, Hertkorn et al., 2006; Stubbins et al., 2010; D'Andrilli et al., 2013).

We found only a few significant differences between the distribution of PARAFAC components between different forest management categories prior to and after incubation. This could be attributable to balanced changes in the relative shares of PARAFAC components used for comparison. Additionally, only parts of the DOM are able to absorb light potentially emitting light by fluorescence (Aiken, 2014). The combination of the low sensitivity of fluorescence and only small differences and/or changes in DOM composition during incubation might cause no visible changes in fluorescence.

Cluster analysis of biomolecules according to molecular composition (Fig. 6) revealed the influence of tree species on aboveground DOM characteristics. Following the water downward, DOM properties assessed with FT-ICR-MS of coniferous stands and deciduous forests from the same site converged, so that both SUB samples of both forest types grouped in one cluster. The same observation was true for all fluorescence components, except %C4, confirming that significant differences in properties detected with FT-ICR-MS disappeared between TOP and SUB samples.

We found significant differences between deciduous and coniferous forests in DOC concentrations of all solution samples collected from aboveground ecosystem fluxes (Fig. 1). Higher DOC concentrations in coniferous than beech forests might partly be attributable to differences in tree traits, like canopy and bark structure, and thus different water–vegetation contact times (Guggenberger et al., 1994).

The compositional differences between coniferous and deciduous aboveground DOM were mainly related to differences in the fractions associated with aromatic compounds like lignin and tannin (Table 3). We found a higher share of lignin-like compounds for water samples from pine compared to beech forests as revealed in the patterns of the van Krevelen plots (Fig. 5), which agreed with findings of Ide et al. (2017). Additionally, we found different lignin / tannin ratios for both management categories. While pine samples exhibited up to 10-fold higher shares of lignin-like than tannin-

like compounds, the ratio was close to or even smaller than 1, especially in LL beech samples (Table 3). Tannins are secondary plant metabolites and play a role in herbivore defence and additionally may affect ecosystem processes (Kraus et al., 2003). A particularly large number of tannin-like molecules in solution samples from beech forests was also reflected in significantly higher shares of PARAFAC component C2 for TF, SF and LL samples in solution samples of beech than of pine (Fig. 7). This agrees with findings of Lorenz et al. (2004), who reported higher concentrations of tannins in beech leaf litter than in pine needles. A higher share of phenolic carbon in beech than spruce solution samples from the same plots than this study was also found by Bischoff et al. (2015), based on ^{13}C NMR analysis.

In addition to the effect of different main tree species, we found no statistically significant effect of management practice on DOM composition. There were no differences between deciduous age class and unmanaged beech forests as well as no influence of forest management intensity (ForMI) on optical DOM properties and DOC concentrations. With the ForMI, we applied an index that is only based on attributes related to aboveground vegetation (harvested tree volume, nonnatural tree species, and deadwood volume with saw cuts).

4.1 Change in biodegradability along the water flow path and among different forest management categories

The biodegradability of DOM in our solution samples was mainly determined by the type of ecosystem flux. The amount of %BDOC in all our samples of ecosystem fluxes was in the range found by Qualls and Haines (1992) in deciduous forests samples (22 %–57 %). With a cumulative degradation of up to 40 % of the initial DOC concentration as well as the highest degradation rates, DOM from SF samples was most bioavailable. TF samples with BDOC up to 36 % contained DOM that seemed slightly less bioavailable. This corresponded to the results of Howard et al. (2018) reporting BDOC in an interquartile range of 36 %–73 % for cedar throughfall and stemflow samples. The lowest degradation rates and, thus, the most stable DOM were found in LL (8 %–18 %), comparable with results of Kalbitz et al. (2003), who reported mean values of 8 % BDOC when incubating extracts from spruce and beech forest fermentation layers (Oa horizons). In addition to other factors, nutrient availability can affect biological degradation of organic matter in ecosystem samples. In our study no additional nutrients were added to compensate for possible limitations. We calculated maximum nutrient demands for the mineralized organic carbon in our samples by using values for bacterial growth requirement of nitrogen and phosphorus suggested by Fellmann et al. (2008b) and measured concentrations of N and P in the solution samples prior to pooling for the incubation experiment (Table S6). The results suggested that constrained biodegra-

dation due to nutrient limitation in TF and LL samples was not likely. Low concentrations for phosphorus in SF samples may, however, have had a limiting effect of biological degradation and the amount of %BDOC could be higher than measured, thus even increasing the difference in the biodegradability of DOM between the samples of SF and those of TF and LL.

Given the findings that carbohydrates and amino acids were typically preferentially utilized by microorganisms during degradation of different compounds in DOM solutions (Volk et al., 1997; Amon et al., 2001; Kalbitz et al., 2003), we expected a significant decrease in %C6 after 28 days of incubation. The fact that we found no significant change in %C6 during incubation might indicate that amino acids were bound in and on other, less degradable organic substances, so that they were protected against degradation (Volk et al., 1997). However, phenolic compounds such as tannins and simple phenols have also been shown to contribute to those regions of fluorescence (Goldberg and Weiner, 1993; Maie et al., 2007; Hernes et al., 2009).

Consistent with other studies (Kalbitz et al., 2003; Fellmann et al., 2008a), we found a negative correlation between %BDOC and aromaticity indicators (SUVA_{254}). This supported the assumption that especially aromatic structures are stable against rapid biological degradation. The significant positive correlation between SUVA_{254} and %C1 combined with the significant increase in component C1 after 28 days of incubation indicated either a transformation of former nonaromatic into aromatic compounds or a relative accumulation of the latter.

The larger share of condensed hydrocarbons in TF could explain the reduced biodegradability of DOM in TF compared to SF. LL showed the highest portion of aromatic DOM compounds. This was indicated by the highest SUVA_{254} values, the highest percentage of the tannin-associated PARAFAC component C2 (Fig. 7), and the highest share of the tannin-like and lignin-like molecules (Table 3). This observation coincided with studies of Peichl et al. (2007) and Inamdar et al. (2012) and could explain the lowest amount of %BDOC in LL compared to TF and SF.

5 Conclusion

There are distinct changes in DOC concentrations, chemical DOM composition and DOM biodegradability along the water flow path through European forest ecosystems. Aboveground DOM composition was influenced by forest management, namely selection of the main tree species (deciduous versus coniferous), but not by management intensity (age-class beech versus unmanaged beech forests; ForMI). Biodegradability mainly depended on the type of ecosystem flux, with SF containing the most biodegradable DOM and LL the least. The systematic changes in DOM properties suggest that the biotransformation and degradation of organic

molecules in combination with their interaction with the soil solid phase cause an alignment of the composition of DOM from different sources along the water flow path through forest ecosystems, producing a characteristic pattern of organic compounds in mineral soil solutions.

Data availability. Data are available at <https://doi.org/10.14279/depositonce-8321> (Thieme et al., 2019) or via data search on the Biodiversity Exploratories Information System home page: Biodiversity Exploratories <https://www.bexis.uni-jena.de/PublicData/About.aspx> (last access: 3 April 2019).

Supplement. The supplement related to this article is available online at: <https://doi.org/10.5194/bg-16-1411-2019-supplement>.

Author contributions. JS, MK, WW, BM, LT, MTS, SB and DH formulated the hypotheses and designed the sampling strategy. LT, SB and MTS collected the samples. LT and DG performed fluorescence measurements and all statistical analysis. DH and BS measured and processed FT-ICR-MS data for assignment of DOM molecular formulae. SB and UNM conducted the biodegradation experiment. All authors interpreted the results and wrote the paper.

Competing interests. The authors declare that they have no conflict of interest.

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References

- Ad-Hoc-Arbeitsgruppe Boden (der Staatlichen Geologischen Dienste und der Bundesanstalt für Geowissenschaften und Rohstoffe): Bodenkundliche Kartieranleitung (KA5), Schweitzerbart'sche Verlagsbuchhandlung, Stuttgart, 141–142, 2005.
- Aiken, G. R.: Fluorescence and dissolved organic matter: a chemist's perspective, in: Aquatic Organic Matter Fluorescence, 1st Edn., edited by: Coble, P. G., Lead, J. R., Baker, A., Reynolds, D. M., and Spencer, R. G., Cambridge environmental chemistry series, Cambridge University Press, 35–74, 2014.
- Aitkenhead-Peterson, J. A., McDowell, W. H., and Neff, J. C.: Sources, Production, and Regulation of Allochthonous Dissolved Sources, Production, and Regulation of Allochthonous Dissolved Organic Matter Inputs to Surface Waters to Surface Waters, in: Aquatic ecosystems: Interactivity of Dissolved Organic Matter, edited by: Findlay, S. and Sinsabaugh, R., 25–70, Academic Press, Burlington, 2003.
- Albinsson, B., Li, S., Lundquist, K., and Stomberg, R.: The origin of lignin fluorescence, *J. Mol. Struct.*, 508, 19–27, [https://doi.org/10.1016/S0022-2860\(98\)00913-2](https://doi.org/10.1016/S0022-2860(98)00913-2), 1999.
- Amon, R. M. W., Fitznar, H.-P., and Benner, R.: Linkages among the bioreactivity, chemical composition, and diagenetic state of marine dissolved organic matter, *Limnol. Oceanogr.*, 46, 287–297, <https://doi.org/10.4319/lo.2001.46.2.0287>, 2001.
- Arnstadt, T., Hoppe, B., Kahl, T., Kellner, H., Krüger, D., Bauhus, J., and Hofrichter, M.: Dynamics of fungal community composition, decomposition and resulting deadwood properties in logs of *Fagus sylvatica*, *Picea abies* and *Pinus sylvestris*, *Forest Ecol. Manag.*, 382, 129–142, <https://doi.org/10.1016/j.foreco.2016.10.004>, 2016.
- Augusto, L., Ranger, J., Binkley, D., and Rothe, A.: Impact of several common tree species of European temperate forests on soil fertility, *Ann. For. Sci.*, 59, 233–253, <https://doi.org/10.1051/forest:2002020>, 2002.
- Avneri-Katz, S., Young, R. B., McKenna, A. M., Chen, H., Corilo, Y. E., Polubesova, T., Borch, T., and Chefetz, B.: Adsorptive fractionation of dissolved organic matter (DOM) by mineral soil: Macroscale approach and molecular insight, *Org. Geochem.*, 103, 113–124, <https://doi.org/10.1016/j.orggeochem.2016.11.004>, 2017.
- Baetz, U. and Martinoia, E.: Root exudates: the hidden part of plant defense, *Trends Plant Sci.*, 19, 90–98, <https://doi.org/10.1016/j.tplants.2013.11.006>, 2014.
- Bantle, A., Borken, W., Ellerbrock, R. H., Schulze, E.-D., Weisser, W. W., and Matzner, E.: Quantity and quality of dissolved organic carbon released from coarse woody debris of different tree species in the early phase of decomposition, *Forest Ecol. Manag.*, 329, 287–294, <https://doi.org/10.1016/j.foreco.2014.06.035>, 2014.
- Benner, R.: Chapter 3 – Chemical Composition and Reactivity, in: Biogeochemistry of Marine Dissolved Organic Matter, edited by: Hansell, D. A. and Carlson, C. A., Academic Press, San Diego, 59–90, 2002.

- Bischoff, S., Schwarz, M. T., Siemens, J., Thieme, L., Wilcke, W., and Michalzik, B.: Properties of dissolved and total organic matter in throughfall, stemflow and forest floor leachate of central European forests, *Biogeosciences*, 12, 2695–2706, <https://doi.org/10.5194/bg-12-2695-2015>, 2015.
- Bolan, N. S., Adriano, D. C., Kunhikrishnan, A., James, T., McDowell, R., and Senesi, N.: Dissolved Organic Matter: Biogeochemistry, Dynamics, and Environmental Significance in Soils, in: *Advances in Agronomy*, 1–75, Academic Press, Burlington, 2011.
- Cleveland, C. C., Neff, J. C., Townsend, A. R., and Hood, E.: Composition, Dynamics, and Fate of Leached Dissolved Organic Matter in Terrestrial Ecosystems: Results from a Decomposition Experiment, *Ecosystems*, 7, 175–285, <https://doi.org/10.1007/s10021-003-0236-7>, 2004.
- Coble, P. G.: Characterization of marine and terrestrial DOM in seawater using excitation-emission matrix spectroscopy, *Mar. Chem.*, 51, 325–346, [https://doi.org/10.1016/0304-4203\(95\)00062-3](https://doi.org/10.1016/0304-4203(95)00062-3), 1996.
- Coble, P. G., Lead, J. R., Baker, A., Reynolds, D. M., and Spencer, R. G. (Eds.): *Aquatic Organic Matter Fluorescence*, 1st Edn., Cambridge environmental chemistry series, Cambridge University Press, 2014.
- Cuss, C. W. and Guéguen, C.: Distinguishing dissolved organic matter at its origin: Size and optical properties of leaf-litter leachates, *Chemosphere*, 92, 1483–1489, <https://doi.org/10.1016/j.chemosphere.2013.03.062>, 2013.
- D'Andrilli, J., Foreman, C. M., Marshall, A. G., and McKnight, D. M.: Characterization of IHSS Pony Lake fulvic acid dissolved organic matter by electrospray ionization Fourier transform ion cyclotron resonance mass spectrometry and fluorescence spectroscopy, *Org. Geochem.*, 65, 19–28, <https://doi.org/10.1016/j.orggeochem.2013.09.013>, 2013.
- Dainard, P. G., Guéguen, C., McDonald, N., and Williams, W. J.: Photobleaching of fluorescent dissolved organic matter in Beaufort Sea and North Atlantic Subtropical Gyre, *Mar. Chem.*, 177, 630–637, <https://doi.org/10.1016/j.marchem.2015.10.004>, 2015.
- Don, A. and Kalbitz, K.: Amounts and degradability of dissolved organic carbon from foliar litter at different decomposition stages, *Soil Biol. Biochem.*, 37, 2171–2179, <https://doi.org/10.1016/j.soilbio.2005.03.019>, 2005.
- Ervens, B. and Volkamer, R.: Glyoxal processing by aerosol multiphase chemistry: towards a kinetic modeling framework of secondary organic aerosol formation in aqueous particles, *Atmos. Chem. Phys.*, 10, 8219–8244, <https://doi.org/10.5194/acp-10-8219-2010>, 2010.
- Farmer, D. K., Matsunaga, A., Docherty, K. S., Surratt, J. D., Seinfeld, J. H., Ziemann, P. J., and Jimenez, J. L.: Response of an aerosol mass spectrometer to organonitrates and organosulfates and implications for atmospheric chemistry, *P. Natl. Acad. Sci. USA*, 107, 6670–6675, <https://doi.org/10.1073/pnas.0912340107>, 2010.
- Fellman, J. B., D'Amore, D. V., and Hood, E.: An evaluation of freezing as a preservation technique for analyzing dissolved organic C, N and P in surface water samples, *Sci. Total Environ.*, 392, 305–312, <https://doi.org/10.1016/j.scitotenv.2007.11.027>, 2008a.
- Fellman, J. B., D'Amore, D. V., Hood, E., and Boone, R. D.: Fluorescence characteristics and biodegradability of dissolved organic matter in forest and wetland soils from coastal temperate watersheds in southeast Alaska, *Biogeochemistry*, 88, 169–184, <https://doi.org/10.1007/s10533-008-9203-x>, 2008b.
- Fischer, M., Bossdorf, O., Gockel, S., Hänsel, F., Hemp, A., Hesenmöller, D., Korte, G., Nieschulze, J., Pfeiffer, S., Prati, D., Renner, S., Schöning, I., Schumacher, U., Wells, K., Buscot, F., Kalko, E. K. V., Linsenmair, K. E., Schulze, E.-D., and Weisser, W. W.: Implementing large-scale and long-term functional biodiversity research: The Biodiversity Exploratories, *Basic Appl. Ecol.*, 11, 473–485, <https://doi.org/10.1016/j.baec.2010.07.009>, 2010.
- Fox, J. and Weisberg, S.: *An R Companion to Applied Regression*, 2nd Edn., Sage, Thousand Oaks, CA, 2011.
- Gödde, M., David, M. B., Christ, M. J., Kaupenjohann, M., and Vance, G. F.: Carbon mobilization from the forest floor under red spruce in the northeastern U.S.A, *Soil Biol. Biochem.*, 28, 1181–1189, [https://doi.org/10.1016/0038-0717\(96\)00130-7](https://doi.org/10.1016/0038-0717(96)00130-7), 1996.
- Goldberg, M. C. and Weiner, E. R.: Fluorescence Spectroscopy in Environmental and Hydrological Sciences, in: *Fluorescence Spectroscopy: New Methods and Applications*, edited by: Wolfbeis, O. S., Springer Berlin Heidelberg, Berlin, Heidelberg, 213–241, 1993.
- Goldmann, K., Schöning, I., Buscot, F., and Wubet, T.: Forest Management Type Influences Diversity and Community Composition of Soil Fungi across Temperate Forest Ecosystems, *Front. Microbiol.*, 6, 1300, <https://doi.org/10.3389/fmicb.2015.01300>, 2015.
- Guggenberger, G., Zech, W., and Schulten, H.-R.: Formation and mobilization pathways of dissolved organic matter: evidence from chemical structural studies of organic matter fractions in acid forest floor solutions, *Org. Geochem.*, 21, 51–66, [https://doi.org/10.1016/0146-6380\(94\)90087-6](https://doi.org/10.1016/0146-6380(94)90087-6), 1994.
- Hagedorn, F., Saurer, M., and Blaser, P.: A ^{13}C tracer study to identify the origin of dissolved organic carbon in forested mineral soils, *European J. Soil Sci.*, 55, 91–100, <https://doi.org/10.1046/j.1365-2389.2003.00578.x>, 2004.
- Hernes, P. J., Bergamaschi, B. A., Eckard, R. S., and Spencer, R. G. M.: Fluorescence-based proxies for lignin in freshwater dissolved organic matter, *J. Geophys. Res.-Biogeo.*, 114, G00F03, <https://doi.org/10.1029/2009JG000938>, 2009.
- Hertkorn, N., Benner, R., Frommberger, M., Schmitt-Kopplin, P., Witt, M., Kaiser, K., Kettrup, A., and Hedges, J. I.: Characterization of a major refractory component of marine dissolved organic matter, *Geochim. Cosmochim. Ac.*, 70, 2990–3010, <https://doi.org/10.1016/j.gca.2006.03.021>, 2006.
- Hertkorn, N., Harir, M., Cawley, K. M., Schmitt-Kopplin, P., and Jaffé, R.: Molecular characterization of dissolved organic matter from subtropical wetlands: a comparative study through the analysis of optical properties, NMR and FTICR/MS, *Biogeosciences*, 13, 2257–2277, <https://doi.org/10.5194/bg-13-2257-2016>, 2016.
- Hothorn, T., Hornik, K., van de Wiel, M. A., and Zeileis, A.: A Lego System for Conditional Inference, *Am. Stat.*, 60, 257–263, 2006.
- Howard, D. H., Stan, J. T. V., Whitetree, A., Zhu, L., and Stubbins, A.: Interstorm Variability in the Biolability of Tree-Derived Dissolved Organic Matter (Tree-DOM) in Throughfall and Stemflow, *Forests*, 9, 236, <https://doi.org/10.3390/f9050236>, 2018.
- Ide, J., Ohashi, M., Takahashi, K., Sugiyama, Y., Piirainen, S., Kortelainen, P., Fujitake, N., Yamase, K., Ohte, N., Moritani, M., Hara, M., and Finér, L.: Spatial variations in the molecular diversity of dissolved organic matter in water moving through

- a boreal forest in eastern Finland, *Sci. Rep.-UK*, 7, 42102, <https://doi.org/10.1038/srep42102>, 2017.
- Inamdar, S., Finger, N., Singh, S., Mitchell, M., Levia, D., Bais, H., Scott, D., and McHale, P.: Dissolved organic matter (DOM) concentration and quality in a forested mid-Atlantic watershed, USA, *Biogeochemistry*, 108, 55–76, <https://doi.org/10.1007/s10533-011-9572-4>, 2012.
- Jansen, B., Nierop, K. G. J., and Verstraten, J. M.: Mobility of Fe(II), Fe(III) and Al in acidic forest soils mediated by dissolved organic matter: influence of solution pH and metal / organic carbon ratios, *Geoderma*, 113, 323–340, 2003.
- Jansen, B., Nierop, K. G. J., and Verstraten, J. M.: Mechanisms controlling the mobility of dissolved organic matter, aluminium and iron in podzol B horizons, *Eur. J. Soil Sci.*, 56, 537–550, <https://doi.org/10.1111/j.1365-2389.2004.00686.x>, 2005.
- Kahl, T. and Bauhus, J.: An index of forest management intensity based on assessment of harvested tree volume, tree species composition and dead wood origin, *Nat. Conserv.*, 7, 15–27, <https://doi.org/10.3897/natureconservation.7.7281>, 2014.
- Kahl, T., Mund, M., Bauhus, J., and Schulze, E.-D.: Dissolved Organic Carbon from European Beech Logs: Patterns of Input to and Retention by Surface Soil, *Ecoscience*, 19, 364–373, <https://doi.org/10.2980/19-4-3501>, 2012.
- Kaiser, K. and Guggenberger, G.: The role of DOM sorption to mineral surfaces in the preservation of organic matter in soils, *Org. Geochem.*, 31, 711–725, [https://doi.org/10.1016/S0146-6380\(00\)00046-2](https://doi.org/10.1016/S0146-6380(00)00046-2), 2000.
- Kaiser, K. and Kalbitz, K.: Cycling downwards – dissolved organic matter in soils, *Soil Biol. Biochem.*, 52, 29–32, <https://doi.org/10.1016/j.soilbio.2012.04.002>, 2012.
- Kaiser, K., Guggenberger, G., and Zech, W.: Sorption of DOM and DOM fractions to forest soils, *Geoderma*, 74, 281–303, [https://doi.org/10.1016/S0016-7061\(96\)00071-7](https://doi.org/10.1016/S0016-7061(96)00071-7), 1996.
- Kalbitz, K., Schmerwitz, J., Schwesig, D., and Matzner, E.: Biodegradation of soil-derived dissolved organic matter as related to its properties, *Geoderma*, 113, 273–291, [https://doi.org/10.1016/S0016-7061\(02\)00365-8](https://doi.org/10.1016/S0016-7061(02)00365-8), 2003.
- Kim, S., Kramer, R. W., and Hatcher, P. G.: Graphical Method for Analysis of Ultrahigh-Resolution Broadband Mass Spectra of Natural Organic Matter, the Van Krevelen Diagram, *Anal. Chem.*, 75, 5336–5344, <https://doi.org/10.1021/ac034415p>, 2003.
- Kim, S., Kaplan, L. A., Benner, R., and Hatcher, P. G.: Hydrogen-deficient molecules in natural riverine water samples – evidence for the existence of black carbon in DOM, *Mar. Chem.*, 92, 225–234, <https://doi.org/10.1016/j.marchem.2004.06.042>, 2004.
- Kindler, R., Siemens, J., Kaiser, K., Walmsley, D. C., Bernhofer, C., Buchmann, N., Cellier, P., Eugster, W., Gleixner, G., Grünwald, T., Heim, A., Ibrom, A., Jones, S. K., Jones, M., Klumpp, K., Kutsch, W., Larsen, K. S., Lehuger, S., Loubet, B., McKenzie, R., Moors, E., Osborne, B., Pilegaard, K., Rebmann, C., Saubders, M., Schmidt, M. I., Schrumpf, M., Seyfferth, J., Skiba, U., Sousana, J.-F., Sutton, M. A., Trfs, C., Vowinkel, B., Zeeman, M. J., and Kaupenjohann, M.: Dissolved carbon leaching from soil is a crucial component of the net ecosystem carbon balance, *Glob. Change Biol.*, 17, 1167–1185, <https://doi.org/10.1111/j.1365-2486.2010.02282.x>, 2011.
- Klotzbücher, T., Kaiser, K., Filley, T. R., and Kalbitz, K.: Processes controlling the production of aromatic water-soluble organic matter during litter decomposition, *Soil Biol. Biochem.*, 67, 133–139, <https://doi.org/10.1016/j.soilbio.2013.08.003>, 2013.
- Kothawala, D. N., Wachenfeldt, E., Koehler, B., and Tranvik, L. J.: Selective loss and preservation of lake water dissolved organic matter fluorescence during long-term dark incubations, *Sci. Total Environ.*, 433, 238–246, <https://doi.org/10.1016/j.scitotenv.2012.06.029>, 2012.
- Kraus, T. C., Dahlgren, R. A., and Zasoski, R. J.: Tannins in nutrient dynamics of forest ecosystems – a review, *Plant Soil*, 256, 41–66, <https://doi.org/10.1023/A:1026206511084>, 2003.
- Lakowicz, J. R.: Principles of Fluorescence Spectroscopy, 3rd Edn., Springer Science+Business Media, <https://doi.org/10.1007/978-0-387-46312-4>, 2006.
- Lambert, T., Bouillon, S., Darchambeau, F., Massicotte, P., and Borges, A. V.: Shift in the chemical composition of dissolved organic matter in the Congo River network, *Biogeosciences*, 13, 5405–5420, <https://doi.org/10.5194/bg-13-5405-2016>, 2016.
- Lee, B. H., Mohr, C., Lopez-Hilfiker, F. D., Lutz, A., Hallquist, M., Lee, L., Romer, P., Cohen, R. C., Iyer, S., Kurten, T., Hu, W. W., Day, D. A., Campuzano-Jost, P., Jimenez, J. L., Xu, L., Ng, N. L., Guo, H. Y., Weber, R. J., Wild, R. J., Brown, S. S., Koss, A., de Gouw, J., Olson, K., Goldstein, A. H., Seco, R., Kim, S., McAvey, K., Shepson, P. B., Starn, T., Baumann, K., Edgerton, E. S., Liu, J. M., Shilling, J. E., Miller, D. O., Brune, W., Schobesberger, S., D'Ambro, E. L., and Thornton, J. A.: Highly functionalized organic nitrates in the southeast United States: Contribution to secondary organic aerosol and reactive nitrogen budgets, *P. Natl. Acad. Sci. USA*, 113, 1516–1521, <https://doi.org/10.1073/pnas.1508108113>, 2016.
- Levia, D. F. and Germer, S.: A review of stemflow generation dynamics and stemflow-environment interactions in forests and shrublands, *Rev. Geophys.*, 53, 673–714, <https://doi.org/10.1002/2015RG000479>, 2015.
- Levia, D. F., van Stan, J. T., Inamdar, S. P., Jarvis, M. T., Mitchell, M. J., Mage, S. M., Scheick, C. E., and Mchale, P. J.: Stemflow and dissolved organic carbon cycling: temporal variability in concentration, flux, and UV-Vis spectral metrics in a temperate broadleaved deciduous forest in the eastern United States, *Can. J. Forest Res.*, 42, 207–216, <https://doi.org/10.1139/x11-173>, 2012.
- Lindow, S. E. and Brandl, M. T.: Microbiology of the Phyllosphere, *Appl. Environ. Microb.*, 2003, 1875–1883, 2003.
- Lorenz, K., Preston, C. M., Krumrei, S., and Feger, K.-H.: Decomposition of needle/leaf litter from Scots pine, black cherry, common oak and European beech at a conurbation forest site, *Eur. J. For. Res.*, 123, 177–188, <https://doi.org/10.1007/s10342-004-0025-7>, 2004.
- Magnússon, R. Í., Tieteme, A., Cornelissen, J. H., Hefting, M. M., and Kalbitz, K.: Tamm Review: Sequestration of carbon from coarse woody debris in forest soils, *Forest Ecol. Manag.*, 377, 1–15, <https://doi.org/10.1016/j.foreco.2016.06.033>, 2016.
- Maie, N., Scully, N. M., Pisani, O., and Jaffé, R.: Composition of a protein-like fluorophore of dissolved organic matter in coastal wetland and estuarine ecosystems, *Water Res.*, 41, 563–570, <https://doi.org/10.1016/j.watres.2006.11.006>, 2007.
- Marschner, B. and Kalbitz, K.: Controls of bioavailability and biodegradability of dissolved organic matter in soils, *Geoderma*, 113, 211–235, [https://doi.org/10.1016/S0016-7061\(02\)00362-2](https://doi.org/10.1016/S0016-7061(02)00362-2), 2003.

- MATLAB: MATLAB and Statistics Toolbox, The MathWorks, Inc., Natick, Massachusetts, USA, 2015.
- Michalzik, B., Kalbitz, K., Park, J.-H., Solinger, S., and Matzner, E.: Fluxes and concentrations of dissolved organic carbon and nitrogen – a synthesis for temperate forests, *Biogeochemistry*, 52, 173–205, <https://doi.org/10.1023/A:1006441620810>, 2001.
- Michalzik, B., Levia, D. F., Bischoff, S., Nätke, K., and Richter, S.: Effects of aphid infestation on the biogeochemistry of the water routed through European beech (*Fagus sylvatica* L.) saplings, *Biogeochemistry*, 129, 197–214, <https://doi.org/10.1007/s10533-016-0228-2>, 2016.
- Moore, T. R.: Dissolved organic carbon in a northern boreal landscape, *Global Biogeochem. Cy.*, 17, 1109, <https://doi.org/10.1029/2003GB002050>, 2003.
- Müller, T., Strobel, K., and Ulrich, A.: Microorganisms in the phyllosphere of temperate forest ecosystems in a changing environment, in: *Microbial Ecology of Aerial Plant Surfaces*, edited by: Bailey, M. J., Lilley, A. K., Timms-Wilson, T. M., and Spencer-Phillips, P. T. N., CAB International, Wallingford, UK, 51–66, 2006.
- Murphy, K. R., Butler, K. D., Spencer, R. G. M., Stedmon, C. A., Boehme, J. R., and Aiken, G. R.: Measurement of Dissolved Organic Matter Fluorescence in Aquatic Environments: An Interlaboratory Comparison, *Environ. Sci. Technol.*, 44, 9405–9412, <https://doi.org/10.1021/es102362t>, 2010.
- Murphy, K. R., Hambly, A., Singh, S., Henderson, R. K., Baker, A., Stuetz, R., and Khan, S. J.: Organic Matter Fluorescence in Municipal Water Recycling Schemes: Toward a Unified PARAFAC Model, *Environ. Sci. Technol.*, 45, 2909–2916, <https://doi.org/10.1021/es103015e>, 2011.
- Murphy, K. R., Stedmon, C. A., Graeber, D., and Bro, R.: Fluorescence spectroscopy and multi-way techniques. PARAFAC, *Anal. Methods-UK*, 5, 6557, <https://doi.org/10.1039/c3ay41160e>, 2013.
- Murphy, K. R., Stedmon, C. A., Wenig, P., and Bro, R.: OpenFluor – an online spectral library of auto-fluorescence by organic compounds in the environment, *Anal. Methods-UK*, 6, 658–661, <https://doi.org/10.1039/C3AY41935E>, 2014.
- Nierop, K. G. J., Jansen, B., and Verstraten, J. M.: Dissolved organic matter, aluminium and iron interactions: precipitation induced by metal/carbon ratio, pH and competition, *Sci. Total. Environ.*, 300, 201–211, 2002.
- Ng, N. L., Brown, S. S., Archibald, A. T., Atlas, E., Cohen, R. C., Crowley, J. N., Day, D. A., Donahue, N. M., Fry, J. L., Fuchs, H., Griffin, R. J., Guzman, M. I., Herrmann, H., Hodzic, A., Iinuma, Y., Jimenez, J. L., Kiendler-Scharr, A., Lee, B. H., Luecken, D. J., Mao, J., McLaren, R., Mutzel, A., Osthoff, H. D., Ouyang, B., Picquet-Varraut, B., Platt, U., Pye, H. O. T., Rudich, Y., Schwantes, R. H., Shiraiwa, M., Stutz, J., Thornton, J. A., Tilgner, A., Williams, B. J., and Zaveri, R. A.: Nitrate radicals and biogenic volatile organic compounds: oxidation, mechanisms, and organic aerosol, *Atmos. Chem. Phys.*, 17, 2103–2162, <https://doi.org/10.5194/acp-17-2103-2017>, 2017.
- Ohno, T.: Fluorescence Inner-Filtering Correction for Determining the Humification Index of Dissolved Organic Matter, *Environ. Sci. Technol.*, 36, 742–746, <https://doi.org/10.1021/es0155276>, 2002.
- Oksanen, J., Blanchet, F. G., Kindt, R., Legendre, P., Minchin, P. R., O'Hara, R. B., Simpson, G. L., Solymos, P., Stevens, M. H. H., and Wagner, H.: *vegan: Community Ecology Package*, R package version 2.2–1, available at: <http://CRAN.R-project.org/package=vegan> (last access: 12 November 2015), 2015.
- Osburn, C. L., Boyd, T. J., Montgomery, M. T., Bianchi, T. S., Coffin, R. B., and Paerl, H. W.: Optical Proxies for Terrestrial Dissolved Organic Matter in Estuaries and Coastal Waters, *Front. Mar. Sci.*, 2, 127, <https://doi.org/10.3389/fmars.2015.00127>, 2016.
- Peichl, M., Moore, T. R., Arain, M. A., Dalva, M., Brodkey, D., and McLaren, J.: Concentrations and fluxes of dissolved organic carbon in an age-sequence of white pine forests in Southern Ontario, Canada, *Biogeochemistry*, 86, 1–17, <https://doi.org/10.1007/s10533-007-9138-7>, 2007.
- Qualls, R. and Haines, L. B.: Biodegradability of Dissolved Organic Matter in Forest Throughfall, Soil Solution, and Stream Water, *Soil Sci. Soc. Am. J.*, 56, 578–586, 1992.
- R Core Team: R: A language and environment for, R Foundation for Statistical Computing, Vienna, Austria, 2015.
- R Core Team: R: A Language and Environment for Statistical Computing, Vienna, Austria, available at: <https://www.R-project.org/> (last access: 21 March 2019), 2018.
- Santín, C., Yamashita, Y., Otero, X. L., Álvarez, M., and Jaffé, R.: Characterizing humic substances from estuarine soils and sediments by excitation-emission matrix spectroscopy and parallel factor analysis, *Biogeochemistry*, 96, 131–147, <https://doi.org/10.1007/s10533-009-9349-1>, 2009.
- Santos, P., Otero, M., Santos, E., and Duarte, A. C.: Molecular fluorescence analysis of rainwater: Effects of sample preservation, *Talanta*, 82, 1616–1621, <https://doi.org/10.1016/j.talanta.2010.07.048>, 2010.
- Shutova, Y., Baker, A., Bridgeman, J., and Henderson, R. K.: Spectroscopic characterisation of dissolved organic matter changes in drinking water treatment: From 5PARAFAC6 analysis to online monitoring wavelengths, *Water Res.*, 54, 159–169, <https://doi.org/10.1016/j.watres.2014.01.053>, 2014.
- Sleighter, R. L. and Hatcher, P. G.: The application of electrospray ionization coupled to ultrahigh resolution mass spectrometry for the molecular characterization of natural organic matter, *J. Mass Spectrom.*, 42, 559–574, <https://doi.org/10.1002/jms.1221>, 2007.
- Sleighter, R. L., Liu, Z., Xue, J., and Hatcher, P. G.: Multivariate Statistical Approaches for the Characterization of Dissolved Organic Matter Analyzed by Ultrahigh Resolution Mass Spectrometry, *Environ. Sci. Technol.*, 44, 7576–7582, <https://doi.org/10.1021/es1002204>, 2010.
- Stedmon, C. A., Markager, S., and Bro, R.: Tracing dissolved organic matter in aquatic environments using a new approach to fluorescence spectroscopy, *Mar. Chem.*, 82, 239–254, [https://doi.org/10.1016/S0304-4203\(03\)00072-0](https://doi.org/10.1016/S0304-4203(03)00072-0), 2003.
- Stenson, A. C., Marshall, A. G., and Cooper, W. T.: Exact Masses and Chemical Formulas of Individual Suwannee River Fulvic Acids from Ultrahigh Resolution Electrospray Ionization Fourier Transform Ion Cyclotron Resonance Mass Spectra, *Anal. Chem.*, 75, 1275–1284, <https://doi.org/10.1021/ac026106p>, 2003.
- Stubbins, A., Silva, L. M., Dittmar, T., and van Stan, J. T.: Molecular and Optical Properties of Tree-Derived Dissolved Organic Matter in Throughfall and Stemflow from Live Oaks and Eastern Red Cedar, *Front. Earth Sci.*, 5, 22, <https://doi.org/10.3389/feart.2017.00022>, 2017.

- Stubbins, A., Spencer, R. G., Chen, H., Hatcher, P. G., Mopper, K., Hernes, P. J., Mwamba, V. L., Mangangu, A. M., Wabakanghanzi, J. N., and Six, J.: Illuminated darkness: Molecular signatures of Congo River dissolved organic matter and its photochemical alteration as revealed by ultrahigh precision mass spectrometry, *Limnol. Oceanogr.*, 55, 1467–1477, <https://doi.org/10.4319/lo.2010.55.4.1467>, 2010.
- Tfaily, M. M., Corbett, J. E., Wilson, R., Chanton, J. P., Glaser, P. H., Cawley, K. M., Jaffé, R., and Cooper, W. T.: Utilization of PARAFAC-Modeled Excitation-Emission Matrix (EEM) Fluorescence Spectroscopy to Identify Biogeochemical Processing of Dissolved Organic Matter in a Northern Peatland, *Photochem. Photobiol.*, 91, 684–695, <https://doi.org/10.1111/php.12448>, 2015.
- Thieme, L., Graeber, D., Kaupenjohann, M., and Siemens, J.: Fast-freezing with liquid nitrogen preserves bulk dissolved organic matter concentrations, but not its composition, *Biogeosciences*, 13, 4697–4705, <https://doi.org/10.5194/bg-13-4697-2016>, 2016.
- Thieme, L., Graeber, D., Hofmann, D., Bischoff, S., Schwarz, M. T., Steffen, B., Meyer, U.-N., Kaupenjohann, M., Wilke, W., Michalzik, B., and Siemens, J.: Dissolved organic matter characteristics of deciduous and coniferous forests with variable management: different at the source, aligned in the soil, *Generic Research Data*, <https://doi.org/10.14279/depositonce-8321>, 2019.
- Thruston Jr., A. D.: A Fluorometric Method for the Determination of Lignin Sulfonates in Natural Waters, *Water Pollution Control Federation*, 42, 1551–1555, 1970.
- Tückmantel, T., Leuschner, C., Preusser, S., Kandeler, E., Angst, G., Mueller, C. W., and Meier, I. C.: Root exudation patterns in a beech forest: Dependence on soil depth, root morphology, and environment, *Soil Biol. Biochem.*, 107, 188–197, 2017.
- van Dam, N. M. and Bouwmeester, H. J.: Metabolomics in the Rhizosphere: Tapping into Belowground Chemical Communication, *Trends Plant Sci.*, 21, 256–265, <https://doi.org/10.1016/j.tplants.2016.01.008>, 2016.
- van Krevelen, D. W.: Graphical-Statistical Method for the Study of Structure and Reaction Processes of Coal, *Fuel*, 29, 228–269, 1950.
- Van Stan, J. T. and Stubbins, A.: Tree-DOM: Dissolved organic matter in throughfall and stemflow, *Limnol. Oceanogr.*, 3, 199–214, <https://doi.org/10.1002/lo.10059>, 2018.
- Volk, C. J., Volk, C. B., and Kaplan, L. A.: Chemical composition of biodegradable dissolved organic matter in streamwater, *Limnol. Oceanogr.*, 42, 39–44, <https://doi.org/10.4319/lo.1997.42.1.0039>, 1997.
- Weishaar, J. L., Aiken, G. R., Bergamaschi, B. A., Fram, M. S., Fujii, R., and Mopper, K.: Evaluation of Specific Ultraviolet Absorbance as an Indicator of the Chemical Composition and Reactivity of Dissolved Organic Carbon, *Environ. Sci. Technol.*, 37, 4702–4708, <https://doi.org/10.1021/es030360x>, 2003.
- Wickland, K. P., Neff, J. C., and Aiken, G. R.: Dissolved organic carbon in Alaskan boreal forest: Sources, chemical characteristics, and biodegradability, *Ecosystems*, 10, 1323–1340, <https://doi.org/10.1007/s10021-007-9101-4>, 2007.
- Yamashita, Y., Maie, N., Briceño, H., and Jaffé, R.: Optical characterization of dissolved organic matter in tropical rivers of the Guayana Shield, Venezuela, *J. Geophys. Res.-Biogeo.*, 115, G00F10, <https://doi.org/10.1029/2009JG000987>, 2010.
- Yano, Y., McDowell, W. H., and Aber, J. D.: Biodegradable dissolved organic carbon in forest soil solution and effects of chronic nitrogen deposition, *Soil Biol. Biochem.*, 32, 1743–1751, [https://doi.org/10.1016/S0038-0717\(00\)00092-4](https://doi.org/10.1016/S0038-0717(00)00092-4), 2000.
- Yu, H., Liang, H., Qu, F., Han, Z.-s., Shao, S., Chang, H., and Li, G.: Impact of dataset diversity on accuracy and sensitivity of parallel factor analysis model of dissolved organic matter fluorescence excitation-emission matrix, *Sci. Rep.-UK*, 5, 10207, <https://doi.org/10.1038/srep10207>, 2015.