





### Neutron protein crystallography at the Heinz Maier-Leibnitz Zentrum (MLZ): New developments and recent application examples

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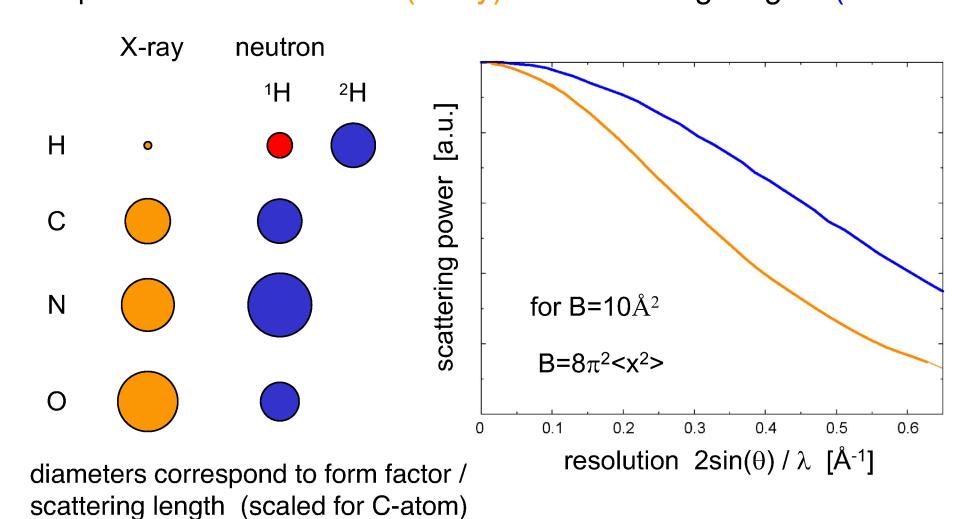
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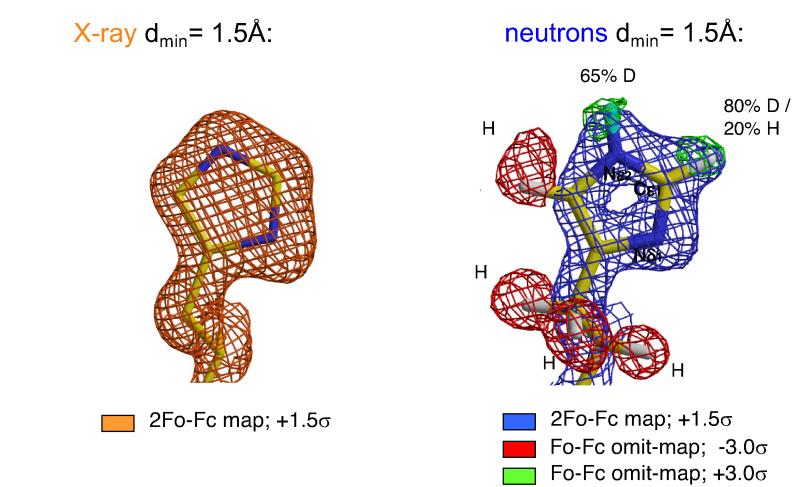
### Neutron structure determination: hydrogen atoms can be resolved even at a resolution of d<sub>min</sub>≈2.5Å → protonation states of amino acid side chains

- protonation states of amino acid side chains
- deuterium exchange as a measure of flexibility and accessibility (discrimination between H / D)
- solvent structure including hydrogen atoms can be analysed
- discrimination between neighbors in the periodic table is possible: e.g. **N** and **O**, **Fe** and **Mn**
- → B-factors (<x²>) of the hydrogen atoms can be compared with data of other techniques
- no radiation damage compared to measurements at synchrotrons

#### Comparison of form factors (X-ray) and scattering lengths (neutrons):

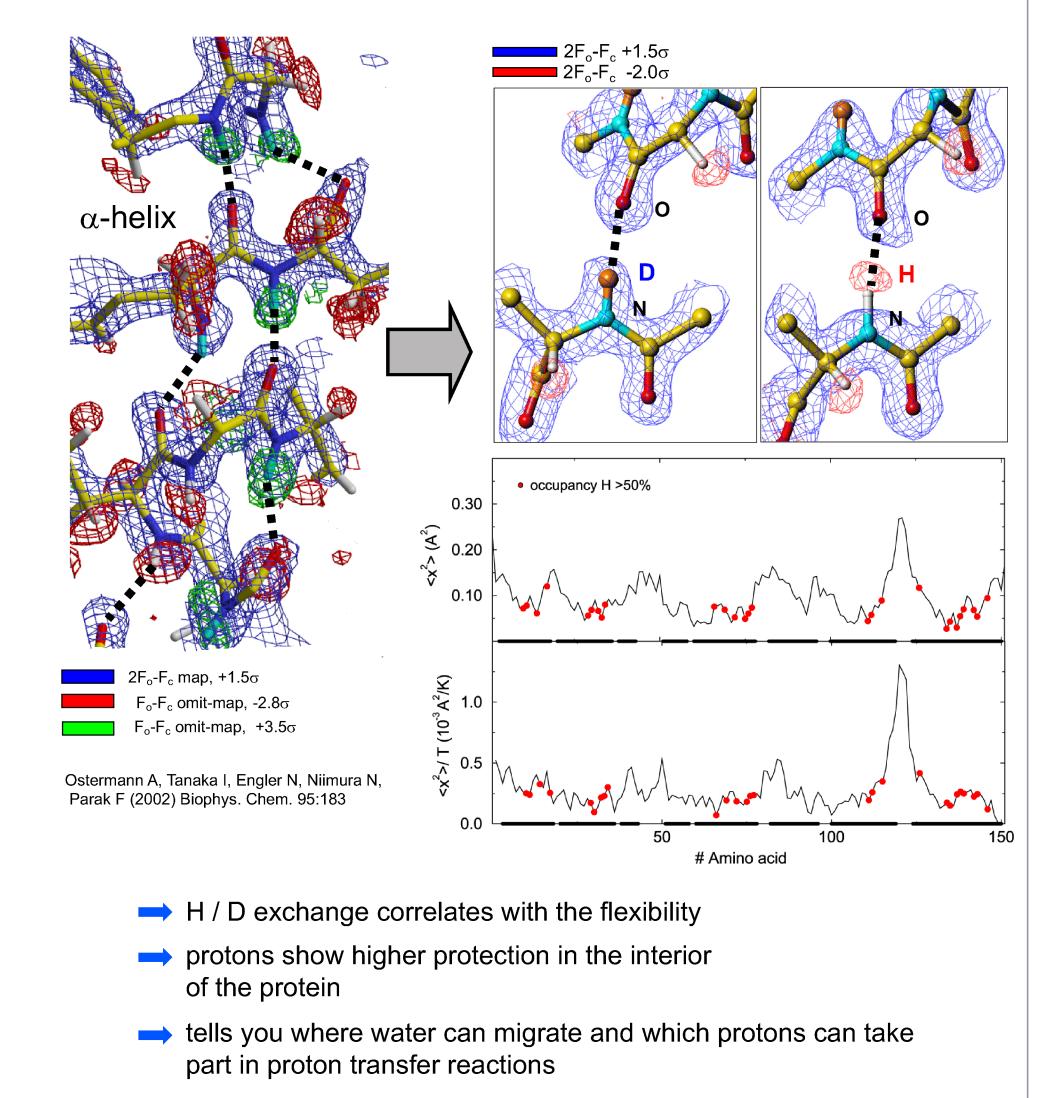


#### Amino acid protonation states:

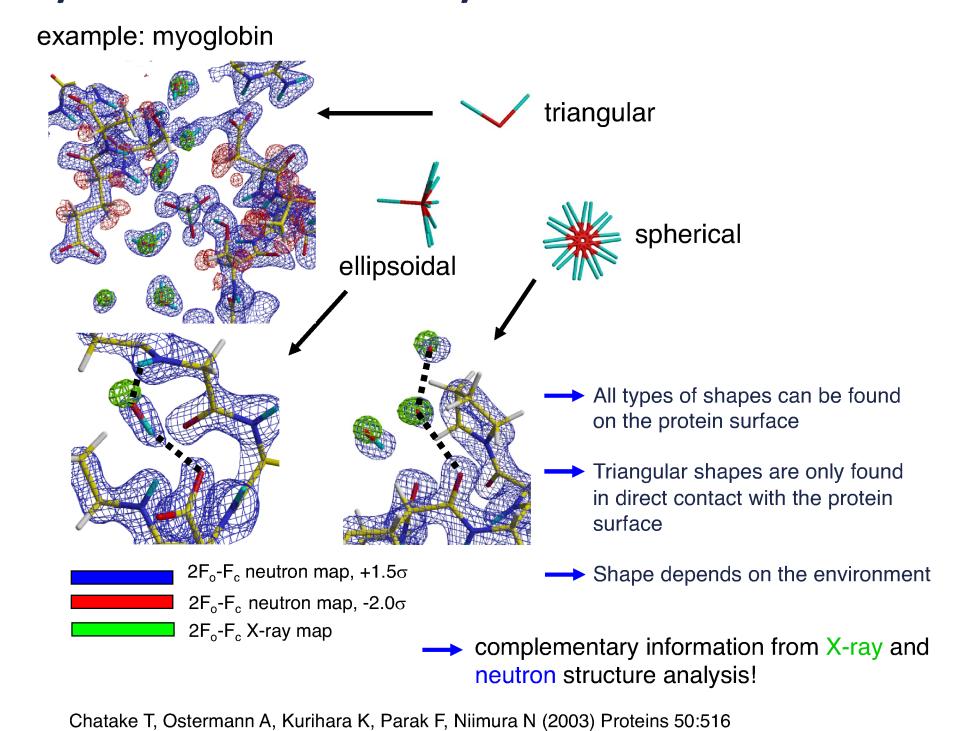


Niimura N, Chatake T, Ostermann A, Kurihara K, Tanaka T. (2003) Z. Kristallogr. 218:96

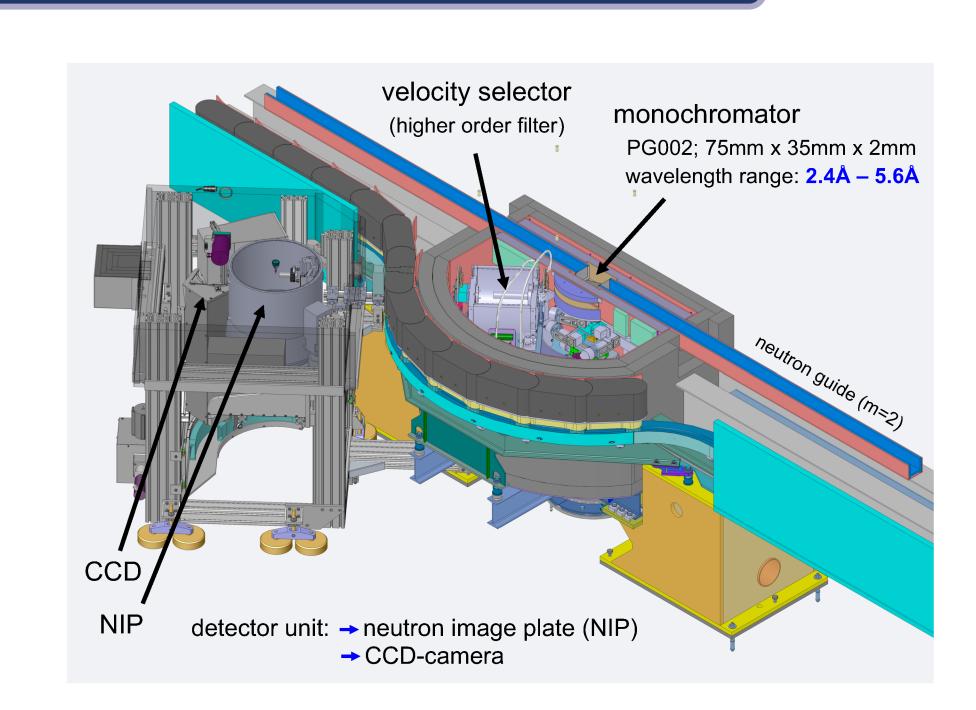
#### Analysis of H/D-exchange:



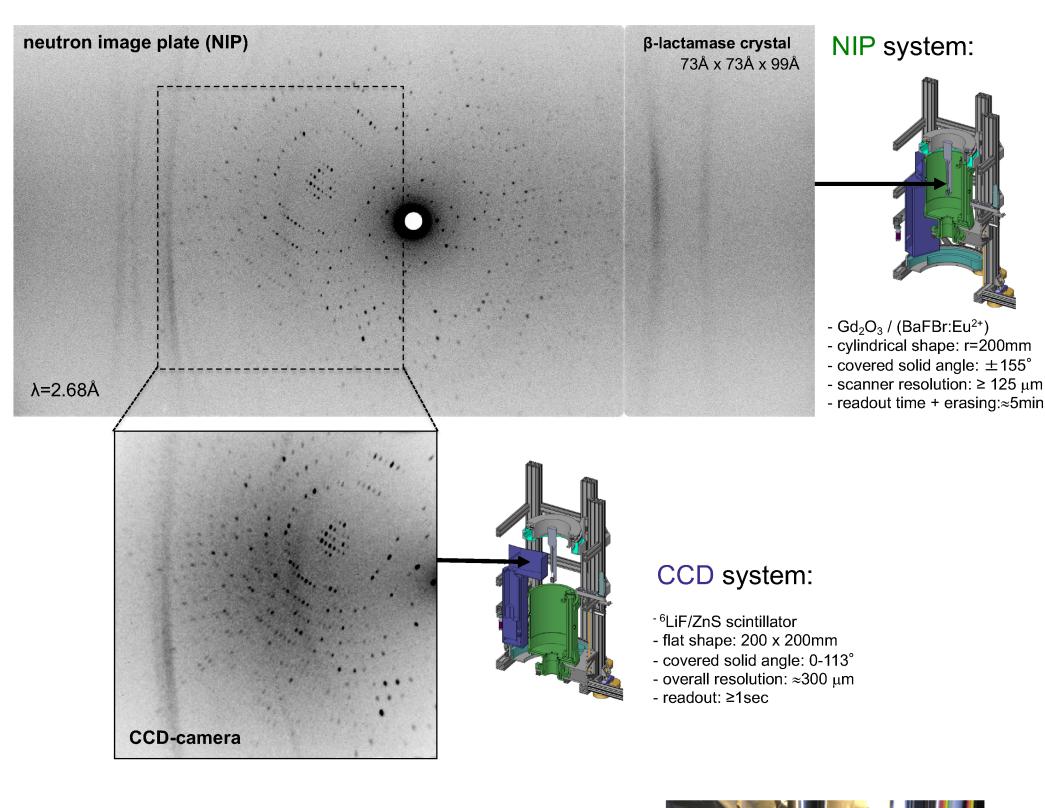
#### **Hydration structure analysis:**



#### The diffractometer BIODIFF:



#### NIP and CCD detector system:



#### Sample environment:

Cryostream & mini-kappa-goniometer

optimizing datacollection strategy
save precious beam time /

no manual re-mounting of crystal necessary for changing the orientation under cryo-condition

increase data set completeness

# Sixis

#### **Example user data-sets:**

## Compound I of cytochrome c peroxidase @100K Casadei et al. (2014) Science 345: 193 His 52 His 52 His 52

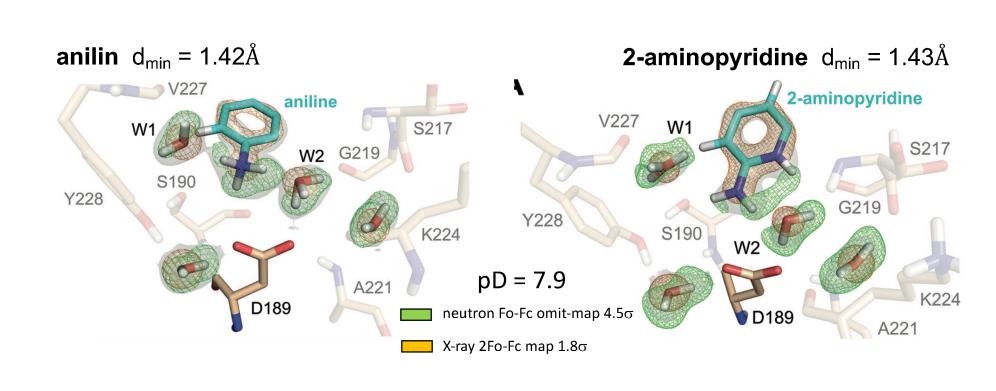
- → The oxygen atom bound to iron (IV) is <u>not</u> protonated!
- → but His 52 is double protonated!
  - Reaction mechanism needs to be reconsidered!

#### Charges shift protonation: inhibitor binding to trypsin

Schiebel J. et al. (2017) Angewandte Chemie Int. Ed. 56: 4887

Trypsin as model system for the important family of serine proteases

Question: do inhibitors with less basic properties become protonated upon binding?



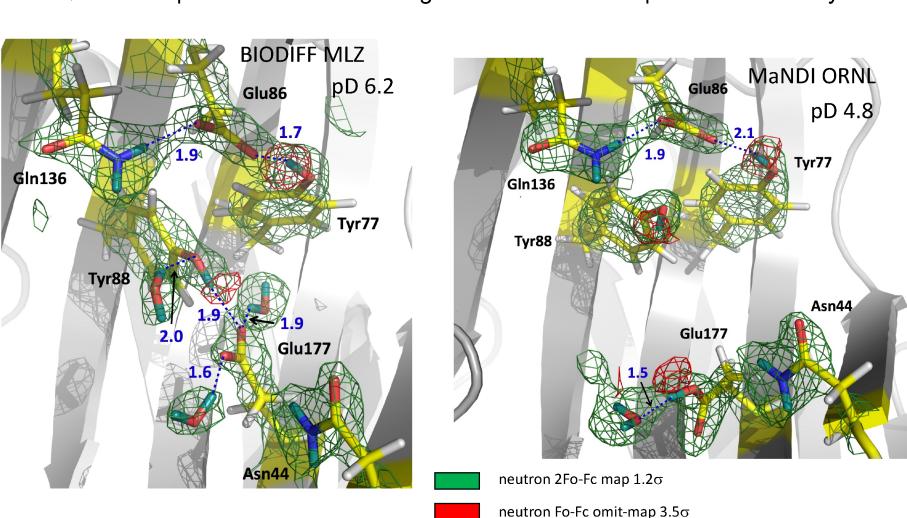
- Despite its low pK<sub>a</sub> of 4.6 the amino group of aniline becomes protonated; Asp189 induces a K<sub>a</sub> shift of four orders of magnitude
- → Whereas in aminopyridine (pK<sub>a</sub> of 6.9), the pyridine nitrogen picks up the proton although its amino group is 1.6Å closer to Asp189
- Therefore, apart from charge-charge distances, tautomer stability is essential for the resulting protonation pattern
- → Correct prediction of such properties is key in drug development !

#### Facilitating processing of biomass

Wan Q. et al., PNAS (2015) 112(40): 12384



- Plant biomass is pre-treated in a very alkaline environment. The goal is to alter the enzymes xylanase to allow it to function effectively in a basic environment.
- This requires detailed knowledge of the reaction sequence of the enzyme!



- The catalytic glutamate residue alternates between two conformations bearing different basicities, first to obtain a proton from the bulk solvent, and then to deliver it to the glycosidic oxygen to initiate the hydrolysis reaction
- Using this knowledge, work on altering the enzyme in a way that allows efficient biomass decomposition even in high pH environments can begin

Next proposal deadline: September 13th, 2019

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