

Superior sintering behavior and ionic conductivity: Facile coating methods for improved electrolyte materials

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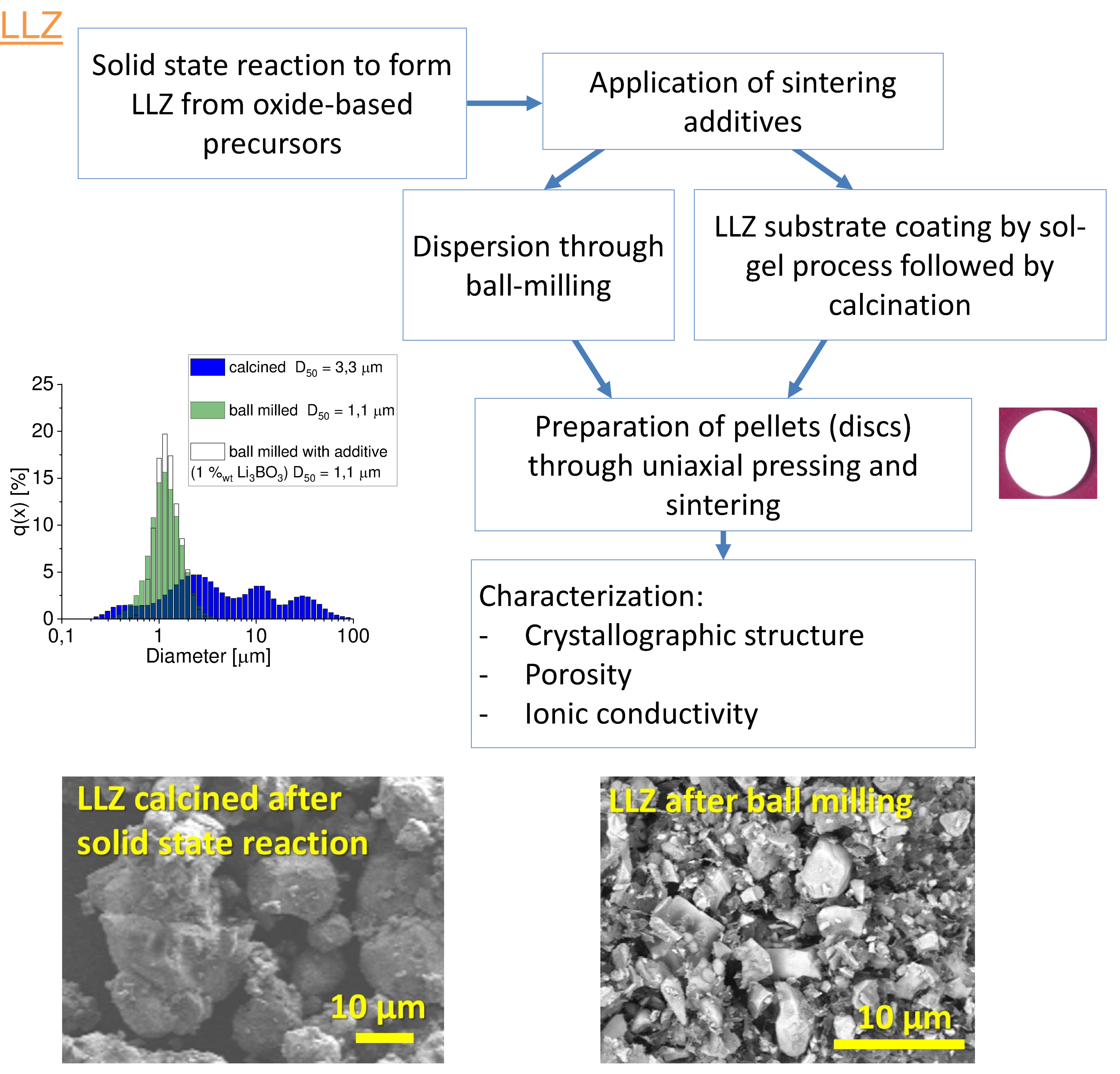
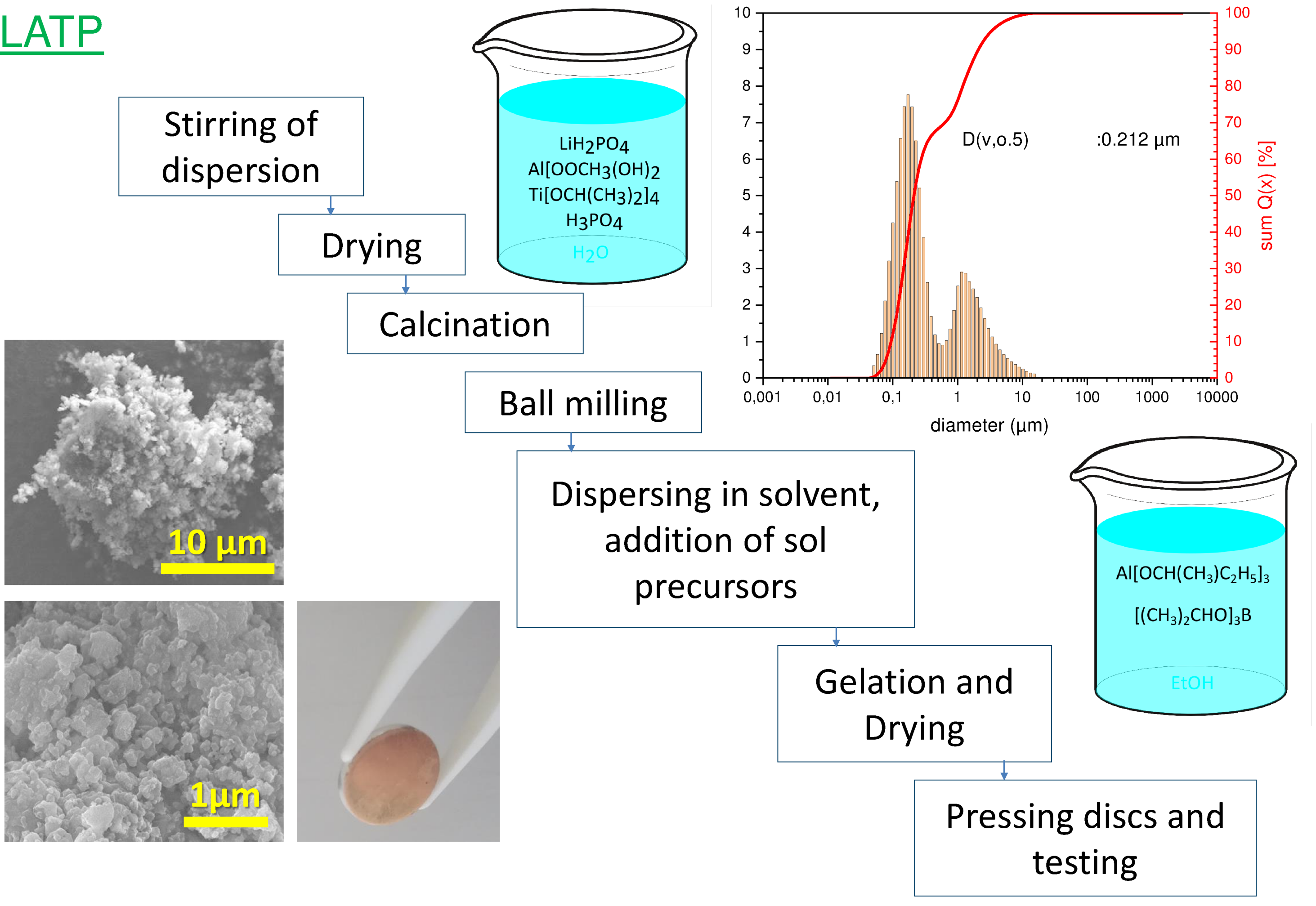
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Introduction

Solid-state batteries are intensively investigated as a disruptive energy storage technology. Oxide and phosphate-based Li-ion conductors such as $\text{Li}_{1.5}\text{Al}_{0.5}\text{Ti}_{1.5}(\text{PO}_4)_3$ (LATP), which crystallizes in the rhombohedral $\text{NaZr}_2(\text{PO}_4)_3$ structure [1], and $\text{Li}_{6.45}\text{La}_3\text{Zr}_{1.6}\text{Ta}_{0.4}\text{Al}_{0.05}\text{O}_{12}$ (LLZ), with a garnet structure [2] show properties like improved safety, high ionic conductivities (0.1 – 1 mS/cm [3,4]); and electrochemical stability compared to incumbent Li-ion battery technologies, but also require treatment at elevated temperatures.

We present advanced processing methods for these ceramic electrolyte materials. By applying various surface coatings on electrolyte particles and electrode-electrolyte interfaces we aim to improve the sintering behavior and the electrode compatibility.

Experimental



References

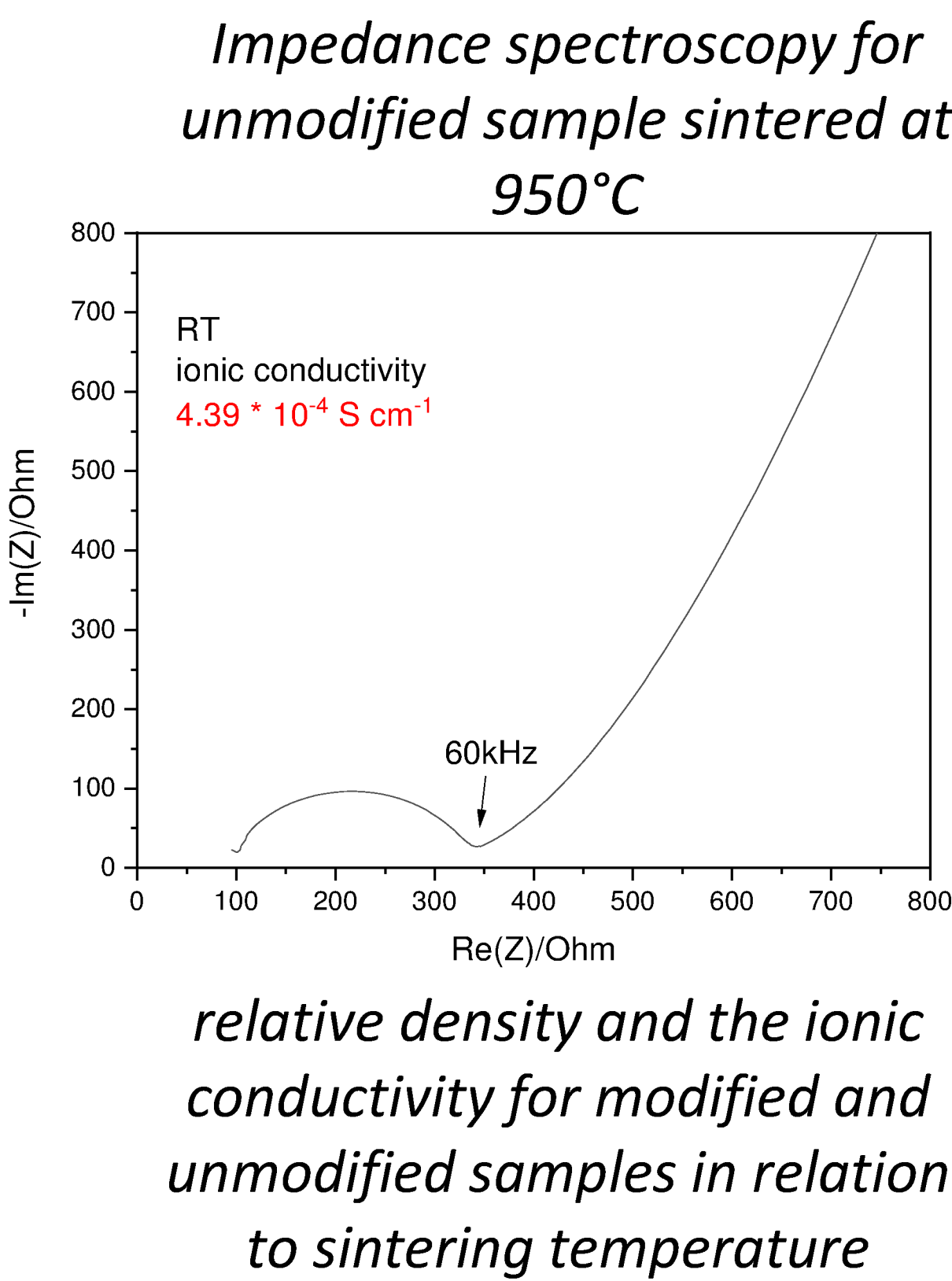
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Results

LATP

As expected the relative density increases with rising sintering temperatures and the grain boundary ionic conductivity decreases with lower sintering temperatures. A maximum is present at 850°C in the sample with modification. The SEM images show additional phases for the modified sample. Cracks are also found, which are expected to have a negative effect on the ionic conductivity.

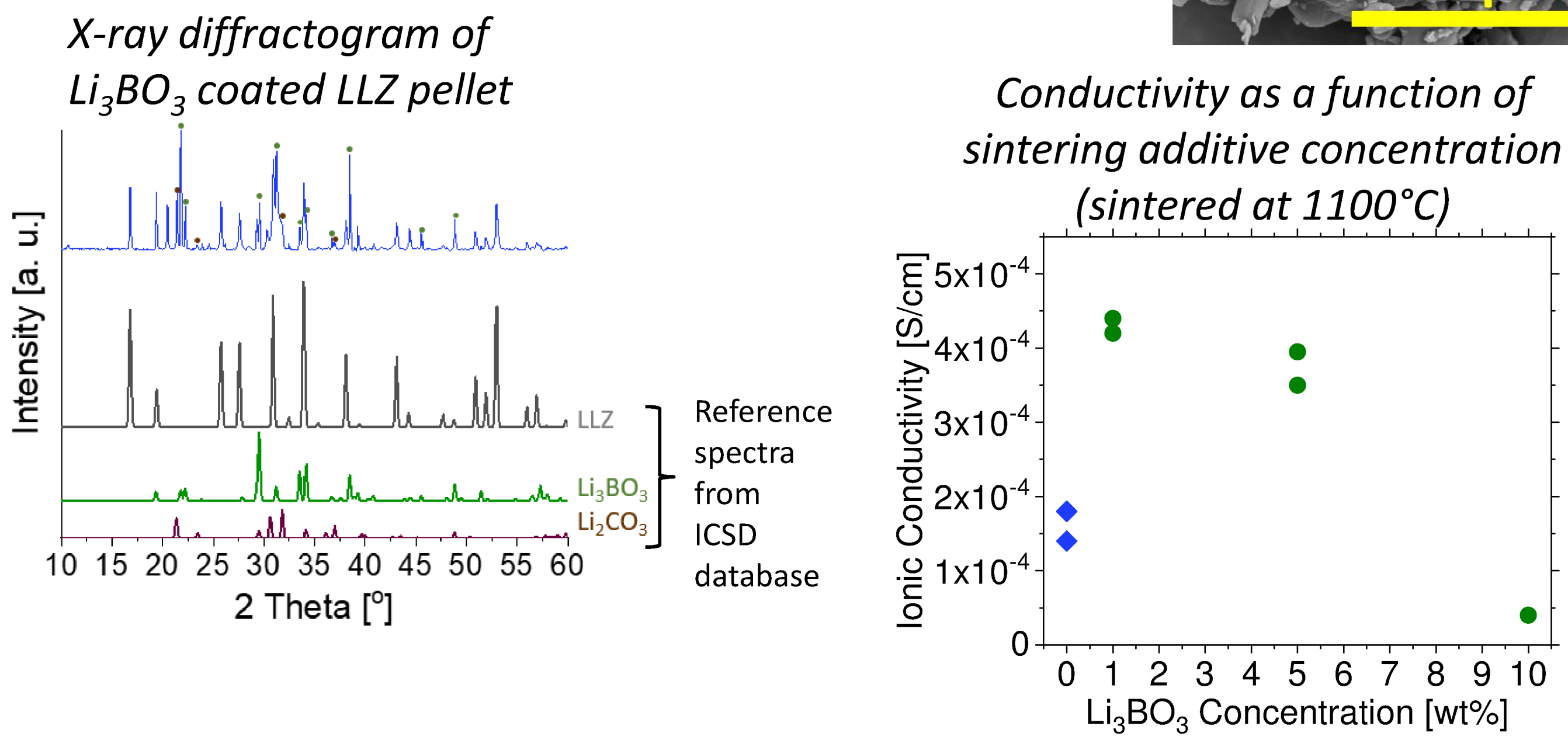
The microstructure is typical for agglomerated primary powders, resulting in large grains and cracks.



LLZ

Spin-coating on planar pellet substrates yield model systems, which guide the development of powder coatings.

Applying the additive in powder form at low concentrations (1 wt%, 5 wt%) yielded the highest conductivity. The coating composition depends on the precursor ratio (Li:B) and calcination temperature.



Summary

- LATP can be modified using sol-gel precursors leading to acceptable ionic conductivities at lower temperatures
- Key factors to further optimize the relative density and the sintering are control of particle size, elimination of agglomeration
- The application of lithium borate powder as a sintering additive yields improved ionic conductivity
- Wet chemical processing enables the application of lithium borate coatings on planar substrates and powders. The desired structure and stoichiometry can be achieved by controlling the precursor composition and calcination temperature

Acknowledgments

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