Superior sintering behavior and ionic conductivity: Facile coating methods for improved electrolyte materials

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Introduction

Solid-state batteries are intensively investigated as a disruptive energy storage technology. Oxide and phosphate-based Li-ion conductors such as Li_{1.5}Al_{0.5}Ti_{1.5}(PO₄)₃ (LATP), which crystallizes in the rhombohedral NaZr₂(PO₄)₃ structure [1], and Li_{6.45}La₃Zr_{1.6}Ta_{0.4}Al_{0.05}O₁₂ (LLZ), with a garnet structure [2] show properties like improved safety, high ionic conductivities (0.1 - 1 mS/cm [3,4]); and electrochemical stability compared to incumbent Li-ion battery technologies, but also require treatment at elevated temperatures.

We present advanced processing methods for these ceramic electrolyte materials. By applying various surface coatings on electrolyte particles and electrode-electrolyte interfaces we aim to improve the sintering behavior and the electrode compatibility.

Experimental LATP Stirring of D(v,o.5) :0.212 μm LiH₂PO₄ dispersion AI[OOCH₃(OH)₂ Ti[OCH(CH₃)₂]₄ H₃PO₄ Drying Calcination Ball milling 0,01 0,1 Dispersing in solvent, addition of sol AI[OCH(CH₃)C₂H₅] precursors [(CH₃)₂CHO]₃BGelation and Drying Pressing discs and testing Solid state reaction to form Application of sintering LLZ from oxide-based additives precursors LLZ substrate coating by sol-Dispersion through gel process followed by ball-milling calcination calcined $D_{50} = 3.3 \mu m$ ball milled $D_{50} = 1.1 \mu m$ Preparation of pellets (discs) ball milled with additive through uniaxial pressing and %) 10- $(1 \%_{\text{wt}} \text{Li}_3 \text{BO}_3) \text{D}_{50} = 1,1 \ \mu\text{m}$ sintering Characterization: Crystallographic structure 0,1 Diameter [µm] Porosity Ionic conductivity e reaction

References

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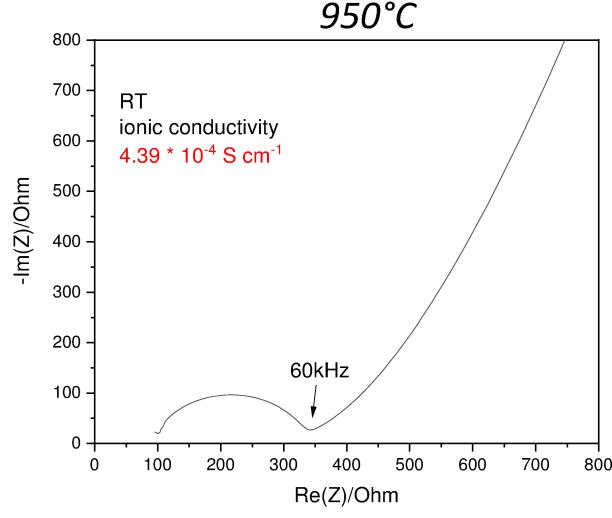
Results

LATP

As expected the relative density increases with rising sintering temperatures and the grain boundary ionic conductivity decreases with lower sintering temperatures. A maximum is present at 850°C in the sample with modification. The SEM images show additional phases for the modified sample. Cracks are also found, which are expected to have a negative effect on the ionic conductivity.

The microstructure is typical for agglomerated

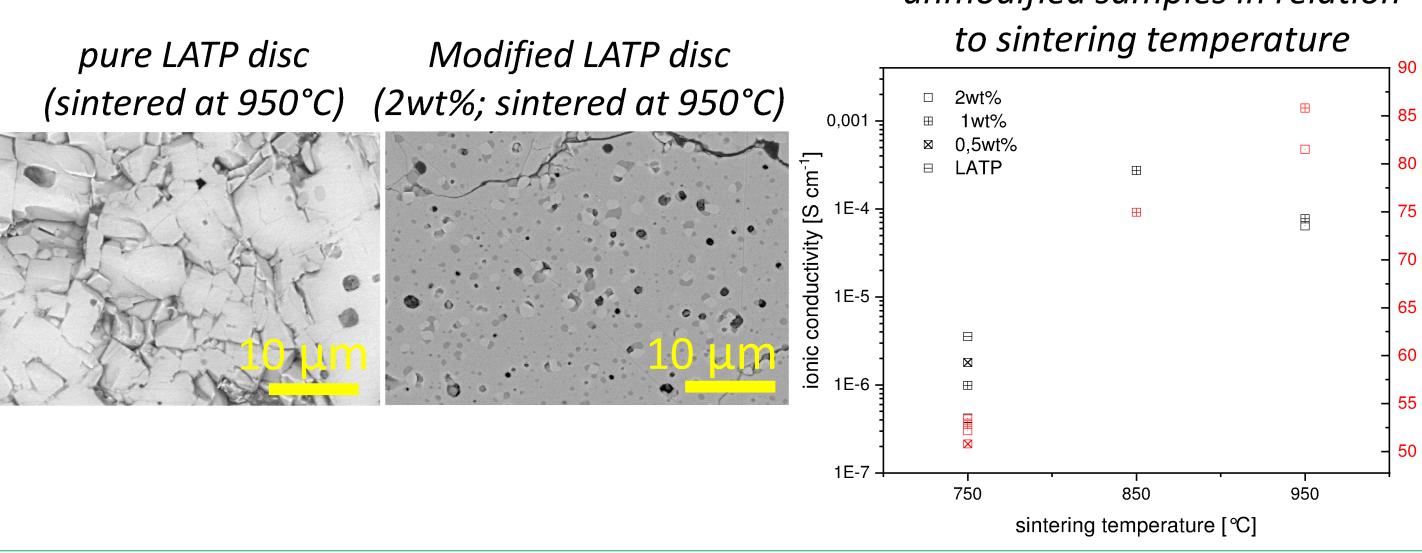
primary powders, resulting in large grains and cracks.



Impedance spectroscopy for

unmodified sample sintered at

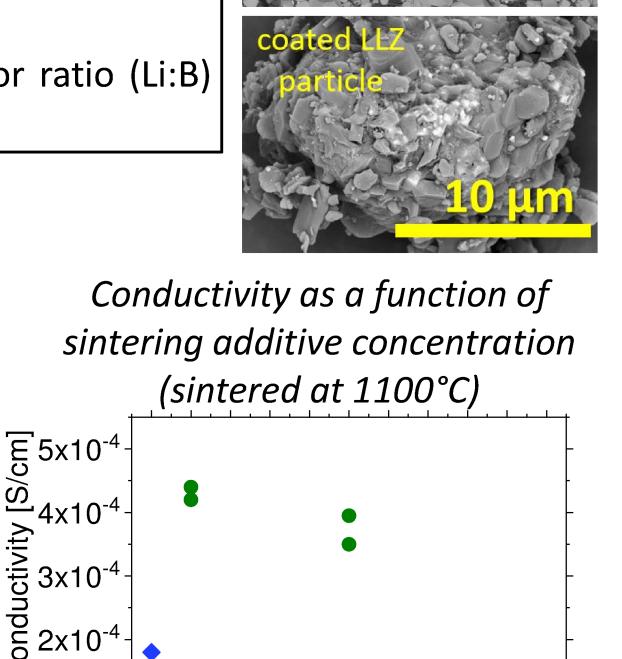
relative density and the ionic conductivity for modified and unmodified samples in relation



Spin-coating on planar pellet substrates yield model systems, which guide the development of powder coatings.

Applying the additive in powder form at low concentrations (1 wt%, 5 wt%) yielded the highest conductivity.

The coating composition depends on the precursor ratio (Li:B) and calcination temperature.



0 1 2 3 4 5 6 7 8 9 10

Li₃BO₃ Concentration [wt%]

Li-borate coating

X-ray diffractogram of Li₃BO₃ coated LLZ pellet Reference spectra from 10 15 20 25 30 35 40 45 50 55 60 database 2 Theta [°]

Summary

LATP can be modified using sol-gel precursors leading to acceptable ionic conductivities at lower temperatures

<u>일</u> 1x10⁻⁴

- Key factors to further optimize the relative density and the sintering are control of particle size, elimination of agglomeration
- The application of lithium borate powder as a sintering additive yields improved ionic conductivity
- Wet chemical processing enables the application of lithium borate coatings on planar substrates and powders. The desired structure and stoichiometry can be achieved by controlling the precursor composition and calcination temperature

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