# Superior sintering behavior and ionic conductivity: Facile coating methods for improved electrolyte materials

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## Introduction

Solid-state batteries are intensively investigated as a disruptive energy storage technology. Oxide and phosphate-based Li-ion conductors such as  $\text{Li}_{1.5}\text{Al}_{0.5}\text{Ti}_{1.5}(\text{PO}_4)_3$  (LATP), which crystallizes in the rhombohedral  $\text{NaZr}_2(\text{PO}_4)_3$  structure [1], and  $\text{Li}_{6.45}\text{La}_3\text{Zr}_{1.6}\text{Ta}_{0.4}\text{Al}_{0.05}\text{O}_{12}$  (LLZ), with a garnet structure [2] show properties like improved safety, high ionic conductivities (0.1 – 1 mS/cm [3,4]); and electrochemical stability compared to incumbent Li-ion battery technologies, but also require treatment at elevated temperatures.

We present advanced processing methods for these ceramic electrolyte materials. By applying various surface coatings on electrolyte particles and electrode-electrolyte interfaces we aim to improve the sintering behavior and the electrode compatibility.

#### **Experimental LATP** Stirring of :0.212 µm D(v,o.5) LiH<sub>2</sub>PO<sub>4</sub> dispersion AI[OOCH<sub>3</sub>(OH)<sub>2</sub> Ti[OCH(CH<sub>3</sub>)<sub>2</sub>]<sub>4</sub> H<sub>3</sub>PO<sub>4</sub> Drying Calcination Ball milling 0,001 0,01 0,1 100 1000 diameter (µm) Dispersing in solvent, addition of sol AI[OCH(CH<sub>3</sub>)C<sub>2</sub>H<sub>5</sub>] precursors [(CH<sub>3</sub>)<sub>2</sub>CHO]<sub>3</sub>BGelation and Drying Pressing discs and testing Solid state reaction to form Application of sintering LLZ from oxide-based additives precursors LLZ substrate coating by sol-Dispersion through gel process followed by ball-milling calcination calcined $D_{50} = 3.3 \mu m$ ball milled $D_{50} = 1.1 \mu m$ Preparation of pellets (discs) ball milled with additive through uniaxial pressing and %) 10- $(1 \%_{\text{wt}} \text{Li}_3 \text{BO}_3) D_{50} = 1,1 \mu\text{m}$ sintering Characterization: Crystallographic structure 0,1 Diameter [µm] Porosity Ionic conductivity e reaction

## References

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## Results

#### LATP

As expected the relative density increases with rising sintering temperatures and the grain boundary ionic conductivity decreases with lower sintering temperatures. A maximum is present at 850°C in the sample with modification. The SEM images show additional phases for the modified sample. Cracks are also found, which are expected to have a negative effect on the ionic conductivity.

The microstructure is typical for agglomerated primary powders, resulting in large grains and cracks.

pure LATP disc Modified LATP disc (sintered at 950°C) (2wt%; sintered at 950°C)

unmodified sample sintered at 950°C

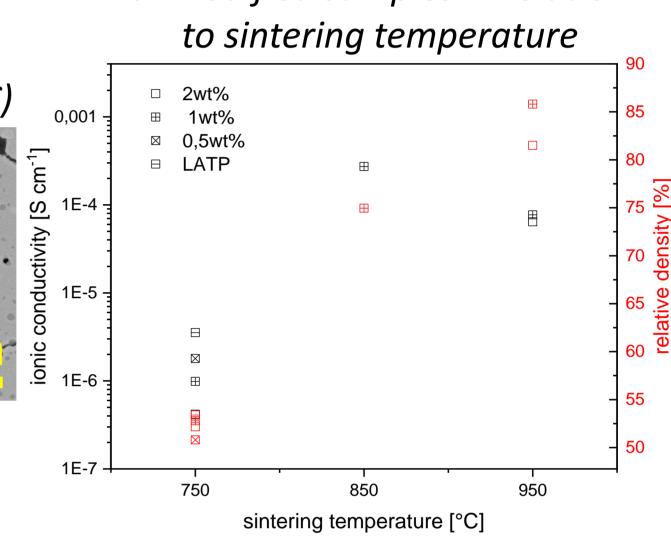
RT ionic conductivity
4.39 \* 10<sup>-4</sup> S cm<sup>-1</sup>

60kHz

Impedance spectroscopy for

relative density and the ionic conductivity for modified and unmodified samples in relation to sintering temperature

Re(Z)/Ohm



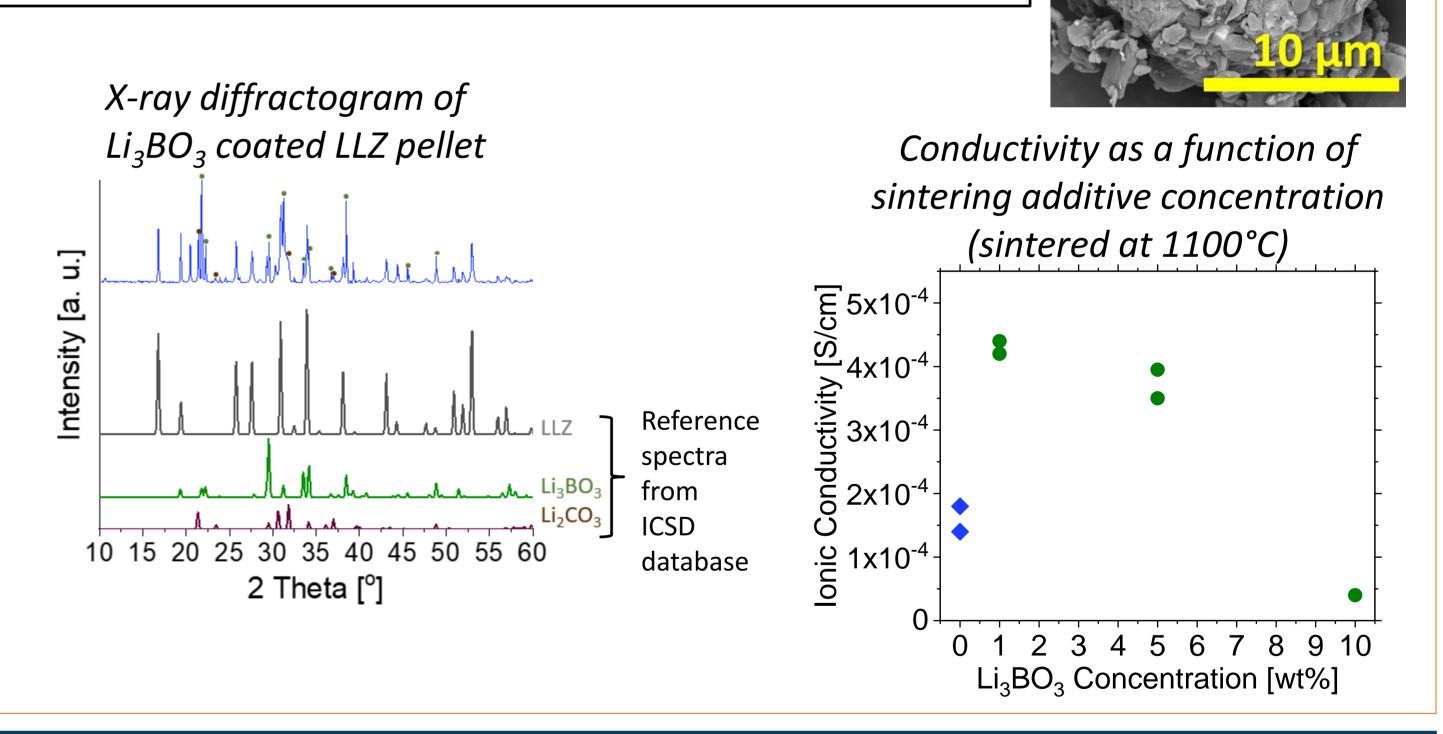
Li-borate coating

#### LLZ

Spin-coating on planar pellet substrates yield model systems, which guide the development of powder coatings.

Applying the additive in powder form at low concentrations (1 wt%, 5 wt%) yielded the highest conductivity.

The coating composition depends on the precursor ratio (Li:B) and calcination temperature.



### Summary

- LATP can be modified using sol-gel precursors leading to acceptable ionic conductivities at lower temperatures
- Key factors to further optimize the relative density and the sintering are control of particle size, elimination of agglomeration
- The application of lithium borate powder as a sintering additive yields improved ionic conductivity
- Wet chemical processing enables the application of lithium borate coatings on planar substrates and powders. The desired structure and stoichiometry can be achieved by controlling the precursor composition and calcination temperature

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