

Thermophoresis of ionic solutes: Preliminary step to protein-ligand interactions

S. Mohanakumar¹ and S. Wiegand^{1,2}

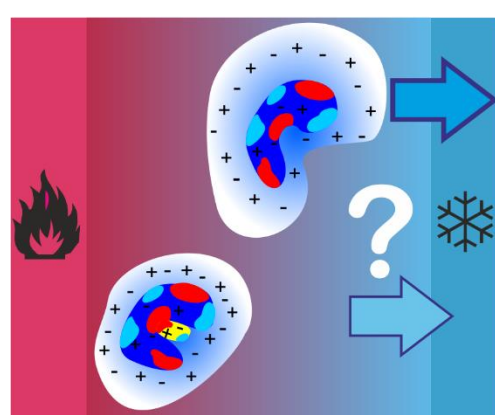
¹IBI-4: Biomacromolecular Systems & Processes, Forschungszentrum Jülich, 52428 Jülich, Germany

²Chemistry Department – Physical Chemistry, University Cologne, D-50939 Cologne, Germany

Motivation [1], [4]

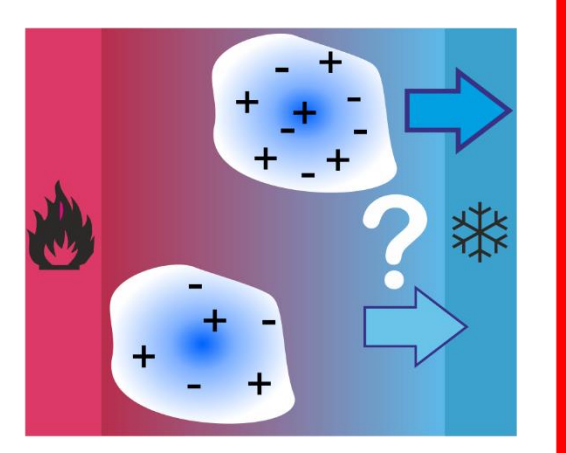
Long term goal: protein-ligand binding

- hydrophilic groups
- hydrophobic groups
- buffer (contains salt)

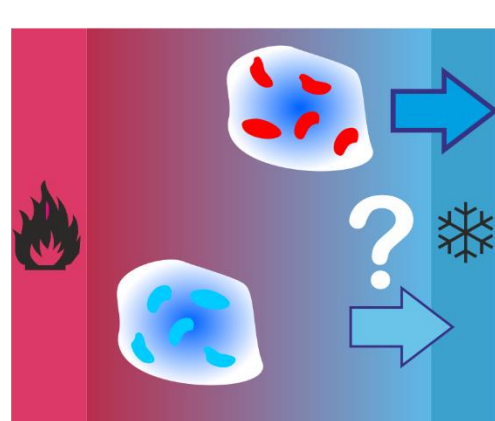


ionic solutes

Reduce Complexity non-ionic solutes



Ion specific effects?



Thermophoresis [2]



$$\vec{j} = -\rho D \vec{\nabla} c - \rho c(1-c) D_T \vec{\nabla} T$$

Steady state: $\vec{j} = 0$

$$S_T = \frac{\text{Thermodiffusion}}{\text{Diffusion}} = \frac{D_T}{D} = -\frac{1}{c(1-c)} \frac{\Delta c}{\Delta T}$$

Systems [3]



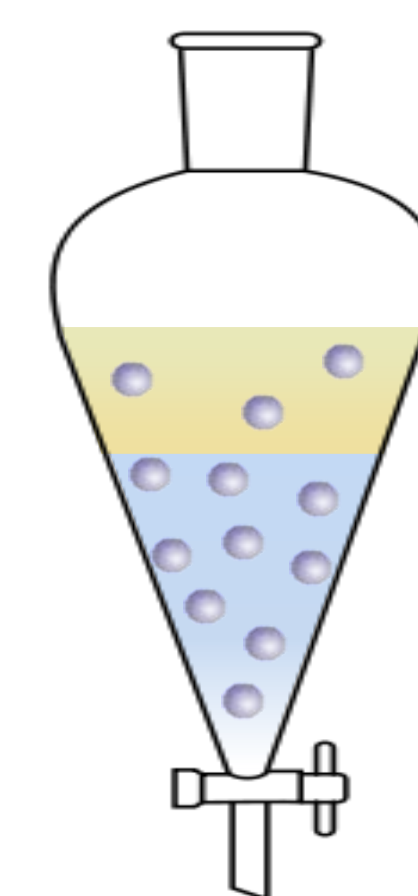
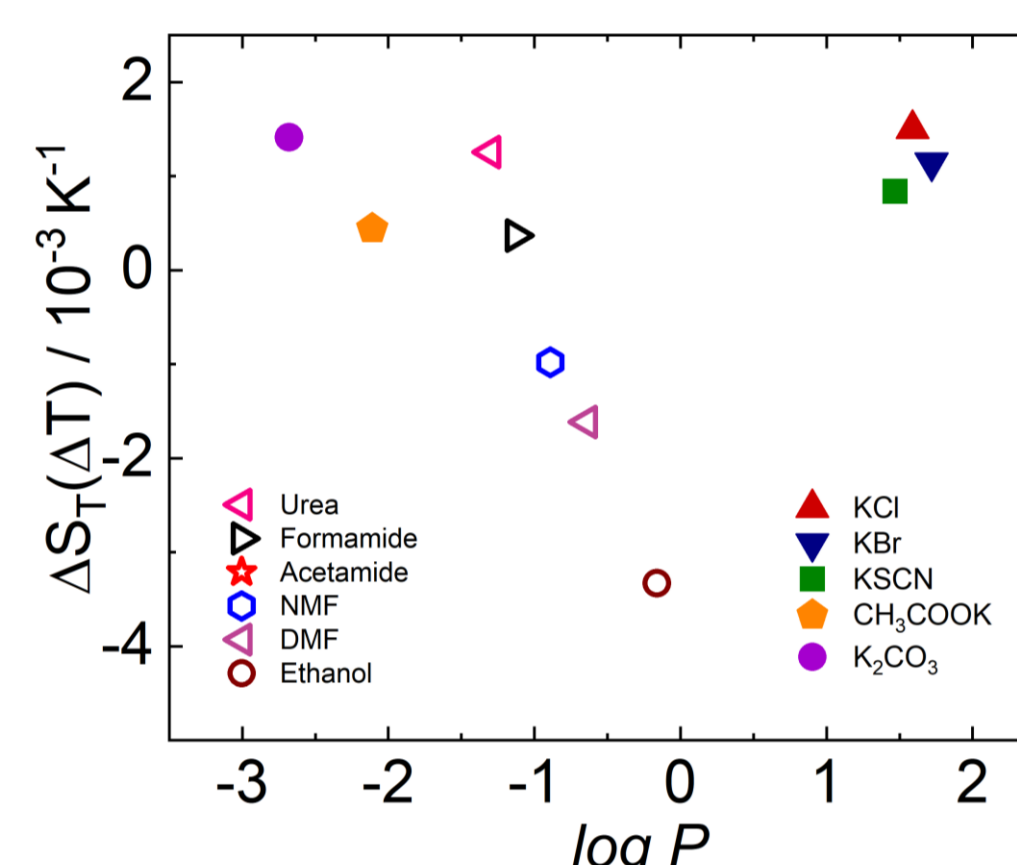
more
hydrophilic

HOFMEISTER SERIES

more
hydrophobic

TDFRS: results [6], [7]

2. $\Delta S_T(\Delta T)$ vs log P



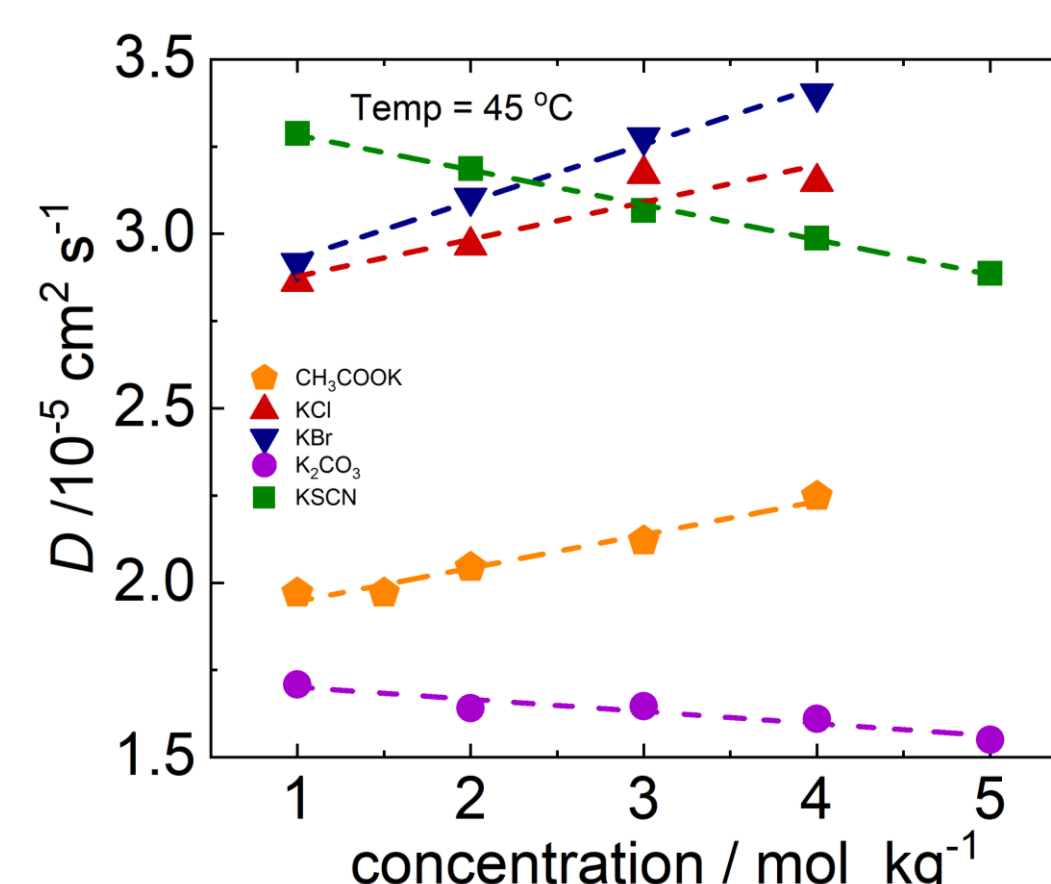
organic phase
→ c_{org}

water → c_{aq}

$$P = \frac{c_{\text{org}}}{c_{\text{aq}}}$$

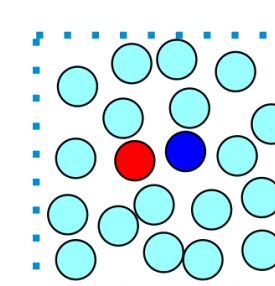
- Hydrophilic salts follow trend of non-ionic solutes

3. Diffusion coefficient

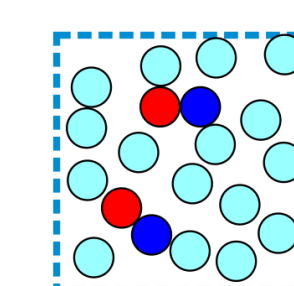


- KCl, KBr, CH₃COOK: follows predicted behavior^[7,6]
- KSCN, K₂CO₃: decrease in D with concentration

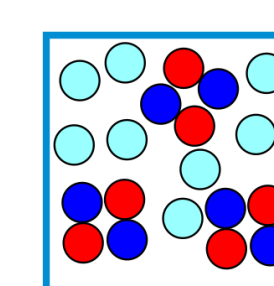
What is the diffusing entity?



ion



ion-pairs



ion clusters

Salt concentration

Conclusion [8]

- Salt specific effects can be observed
- Thermophoresis scale differs from Hofmeister series



more
hydrophilic

THERMODIFFUSION

more
hydrophobic

- Different entities as a function of concentration respond to temperature gradients

References

- [1] D. Niether et al., *Polymers*, **12** (2020), 376 [2] S. Di Lecce et al., *Phys. Chem. Chem. Phys.*, **19** (2017), 9575 [3] A. Salis et al., *Chem. Soc. Rev.*, **43**(2014), 7358 [4] D. Niether et al., *J. Chem. Phys.*, **149** (2018), 44506 [5] R. Piazza et al., *Eur. Phys. J. E*, **19** (2006), 59 [6] G. H. Gao et al., *Fluid Phase Equilib.*, **256** (2007), 105 [7] G. E. Zaikov et al., *In Advances in kinetics and mechanism of chemical reactions*, Apple Academic Press, 2013; pp 21–33. [8] S. Mohanakumar et al., *J. Chem. Phys.*, **154** (2021), 084506