



# Active participation of “inert” YSZ substrates on interface formation in $\text{Fe}_3\text{O}_4/\text{YSZ}$ heterostructures

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## ABSTRACT

The bulk and emerging interface properties of magnetite  $\text{Fe}_3\text{O}_4/\text{YSZ}(001)$  heterostructures grown by pulsed laser deposition are investigated.  $\text{Fe}_3\text{O}_4$  thin films (4–38 nm) grow epitaxially in (111) orientation and undergo a Verwey transition at max.  $T_V = 117 \pm 0.6\text{K}$ . Surprisingly, the formation of interfacial  $\text{Fe}_2\text{O}_3$  phase is observed albeit the quasi-inert properties of yttrium-stabilized zirconia (YSZ). Possible mechanisms include either (i) thermodynamically induced interfacial redox reaction at the YSZ substrate surfaces or (ii) oxygen diffusion from the outside atmosphere, as YSZ is a very good oxygen ion conductor. Hence, substrate-assisted oxygen supply may enable the control of emerging interface functionalities.

## 1. Introduction

The atomic scale growth control of complex oxide thin films and heterostructures is essential for realizing advanced oxide electronic applications. Recent developments in high quality oxide growth allowed studying emergent physical properties at oxide interfaces, such as variable conductivity, magnetism or ferroelectricity [1–5]. The ability to create and even tune interface functionalities motivates the search of novel design schemes for oxide heterostructures [6,7]. The overall challenge is to find appropriate ways to control the oxygen content in thin oxide films and at their interfaces, which in turn allow to set their respective functional properties at will.

Iron oxide thin films are promising candidates for application in spin-based oxide electronics due to their variable magnetic properties and conductivity [8–11]. For example, magnetite  $\text{Fe}_3\text{O}_4$  consists of a mixture of  $\text{Fe}^{3+}$  and  $\text{Fe}^{2+}$  cations occupying the tetrahedral and octahedral sites of an inverse spinel lattice, and reveals simultaneously ferrimagnetic and electrically half metallic properties. The ferrimagnetic state persist up to a high Curie temperature of  $T_C = 860\text{K}$ , while the conductivity undergoes an insulator-to-metal (Verwey) transition at  $T_V = 120\text{K}$  [12, 13]. However, it had become evident that the properties of  $\text{Fe}_3\text{O}_4$  ultrathin films are not only controlled by the particular growth conditions but are strongly dependent on the choice of the substrate material [14,

15]. Various previous studies report on a – mostly unintentional – formation of either reduced [16,17] or oxidized [18] phases at substrate–film interfaces.

Recently, we demonstrated that the interactions between substrate and  $\text{Fe}_3\text{O}_4$  ultrathin films can stabilize functional iron oxide phases far from equilibrium which otherwise are not accessible by the standard synthesis routes. In particular, we investigated the impact of  $\text{SrTiO}_3$  substrates on the resulting properties of ultrathin  $\text{Fe}_3\text{O}_4$  films [18] and observed the formation of a 2 u.c. intralayer of ferrimagnetic and insulating  $\gamma\text{-Fe}_2\text{O}_3$ . An interfacial redox process promotes the transport of oxygen ions from the  $\text{SrTiO}_3$  substrate towards  $\text{Fe}_3\text{O}_4$ . Moreover, the thermal activation of the mobility of the oxygen leads to a phase transformations between  $\text{Fe}_3\text{O}_4$ ,  $\gamma\text{-Fe}_2\text{O}_3$  and  $\text{FeO}$  ultrathin iron oxide films.

In this study, we investigate  $\text{Fe}_3\text{O}_4$  thin films interfaced to  $\text{YSZ}(100)$ , which is a transparent non-magnetic insulator at room temperature, but has considerable ionic conductivity at elevated temperatures. The goal is to elucidate the impact of  $\text{YSZ}(100)$  substrates on the bulk and emerging interfacial magnetic, structural and physicochemical properties.  $\text{YSZ}$  shows a much higher resilience to reduction than  $\text{SrTiO}_3$ . Hence, it is often considered as quasi-inert substrate with negligible film-substrate interaction.

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## 2. Experimental details

A series of  $\text{Fe}_3\text{O}_4$  thin films was grown by pulsed laser deposition (PLD) on yttria-stabilized zirconia (001) (YSZ) substrates with 9.5%  $\text{Y}_2\text{O}_3$  doping level. The oxygen partial pressure was kept at  $2 \times 10^{-6}$  mbar. The laser fluence was set to  $1.5 \text{ J/cm}^2$  with a repetition rate of 5 Hz producing ionized particles from a  $\text{Fe}_2\text{O}_3$  rotating target. Here, we vary two growth parameters: the growth temperature ( $T_{\text{growth}} = 300^\circ\text{--}500^\circ$ ) and the film thickness ( $d = 4\text{--}38 \text{ nm}$ ).

The crystalline structure, film thickness and surface roughness of the as-grown  $\text{Fe}_3\text{O}_4$  films were examined using X-ray diffraction (XRD) and X-ray reflectivity (XRR). Both, XRR and XRD experiments were performed at room temperature on a Philips XPert MRD using  $\text{Cu-K}\alpha$ -radiation. Magnetic properties, i.e. saturation magnetization and Verwey transition of the  $\text{Fe}_3\text{O}_4/\text{YSZ}$  heterostructures were investigated by a vibrating sample magnetometer (VSM) using a Quantum Design Dynacool physical properties measurement system (PPMS). Magnetic moment versus temperature  $M(T)$  was measured in zero-field cooling (ZFC) mode with 500 Oe applied field. Hysteresis loops were recorded at a magnetic field of  $H = \pm 5 \text{ T}$  along the in-plane [100]-axis.

Element-selective and depth-dependent chemical information of the films and buried interfaces were analyzed by hard X-ray photoelectron spectroscopy (HAXPES) [19] at the KMC-1 beamline at BESSY II (Berlin, Germany) using the HIKE endstation. The photon energy used in this study is  $h\nu = 4 \text{ keV}$  which results in an information depth of  $\lambda = 20\text{--}25 \text{ nm}$  for  $\text{Fe}_3\text{O}_4$  [20,21]. The samples are adjusted to normal emission geometry.

## 3. Growth temperature: structural and magnetic characterization

### 3.1. Structural properties

First, we study the effect of growth temperature on the structural properties of  $\text{Fe}_3\text{O}_4$  thin films. Using XRD, we performed out-of-plane  $\theta - 2\theta$  scans ranging from  $15^\circ$  to  $110^\circ$ , not shown here. All XRD intensities are normalized to their YSZ (002) reflection. In addition to the two peaks corresponding to the YSZ (002) and (004) reflections, we observe four peaks which represent the  $\text{Fe}_3\text{O}_4$  (111), (222), (333) and (444) reflections. These reflections confirm the out-of-plane growth of  $\text{Fe}_3\text{O}_4$  (111) on YSZ (001). The lattice constant of  $\text{Fe}_3\text{O}_4$  is  $8.395 \text{ \AA}$  [22] and of YSZ equals  $5.16 \text{ \AA}$  [23], and the interplanar spacing  $d_{(111)}$  of  $\text{Fe}_3\text{O}_4$  is  $4.846 \text{ \AA}$ . Therefore, the mismatch between the YSZ (100) and  $\text{Fe}_3\text{O}_4$  (111) is  $+6\%$ . This can explain the preference of the  $\text{Fe}_3\text{O}_4$  thin films to grow parallel to the (111) planes on the YSZ (100).

Fig. 1 shows the out-of-plane  $\theta - 2\theta$  scans of the  $\text{Fe}_3\text{O}_4$  (222) and YSZ (002) reflections. For the films grown at temperatures of  $500^\circ$  and

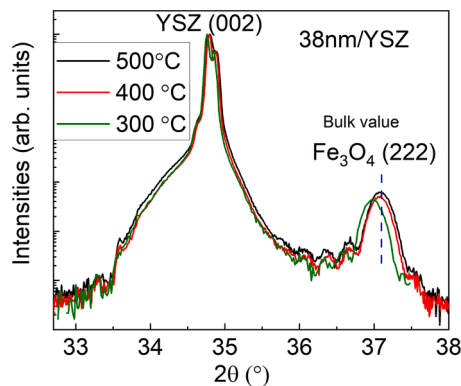


Fig. 1. Out-of-plane  $\theta - 2\theta$  scans of the  $\text{Fe}_3\text{O}_4$  (222) and the YSZ (002) reflections for 38 nm thick films grown at different growth temperatures  $T_{\text{growth}} = 500^\circ, 400^\circ$  and  $300^\circ$ .

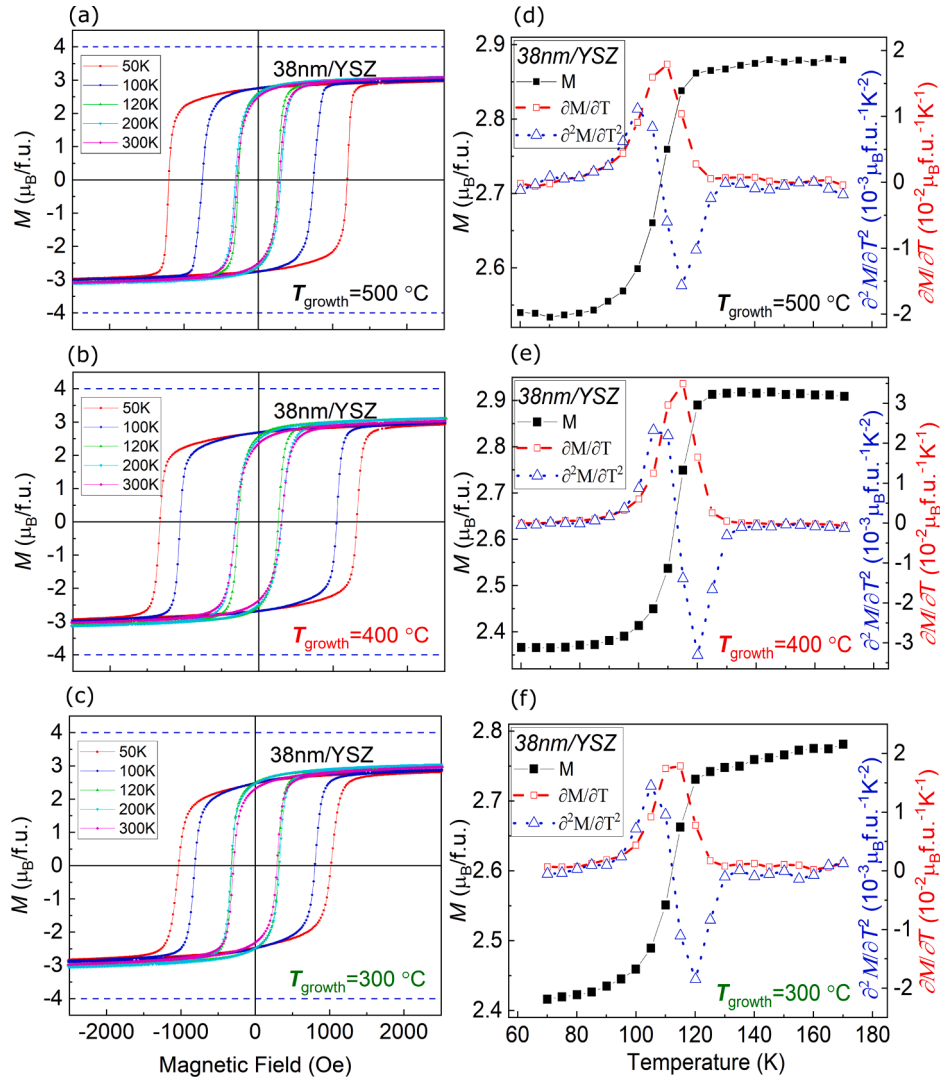
$400^\circ$ , the  $\text{Fe}_3\text{O}_4$  (222) peak positions are very close to the bulk position of  $\text{Fe}_3\text{O}_4$ . However, for the film grown at a temperature of  $300^\circ$ , the peak position is slightly shifted to smaller angles. The calculated out-of-plane lattice constant  $c_{\text{oop}}$  of all the films is very close to the  $\text{Fe}_3\text{O}_4$  bulk value, where  $d_{(111)}$  of the films grown at  $500^\circ$  and  $400^\circ$  is  $4.849 \text{ \AA}$  and for the film grown at  $300^\circ$  equals  $4.864 \text{ \AA}$ . This indicates that all films grow fully relaxed on the YSZ substrates. Moreover, the appearance of the Laue oscillations of the (222)  $\text{Fe}_3\text{O}_4$  reflections reflects the high crystallinity of the films in addition to the structural homogeneity of the films and low density of defects [24].

### 3.2. Saturation magnetization

Next, we study the effect of growth temperature on the magnetic properties of  $\text{Fe}_3\text{O}_4$  thin films. Hysteresis loops have been measured by VSM in order to determine the saturation magnetization of the  $\text{Fe}_3\text{O}_4$  films. Hysteresis loops recorded at temperatures between 5 K and room temperature are depicted in Fig. 2 (a–c). We investigate the saturation magnetization as an important parameter to quantify the magnetic properties of  $\text{Fe}_3\text{O}_4$  films. The saturation magnetization of all  $\text{Fe}_3\text{O}_4$  films is similar within the respective errors ( $3.3\text{--}3.4 \mu_B/\text{f.u.}$ ). The saturation magnetization is less than the ideal theoretical bulk  $\text{Fe}_3\text{O}_4$  value of  $4 \mu_B/\text{f.u.}$  [25], which could be due to the formation of antiferromagnetic anti-phase boundaries and structural dislocations. Moreover, the coercivity of all the films grown at different temperatures increases due to the structural transition below the Verwey transition from cubic to monoclinic [26]. The slight differences can be explained by the formation of dislocations and anti-phase boundaries.

### 3.3. Verwey transition

Another approach to characterize the stoichiometry of the  $\text{Fe}_3\text{O}_4$  film is the Verwey transition. The Verwey temperature  $T_V$  is strongly influenced by the oxygen content. If  $\delta > 0.045$  for nonstoichiometric  $\text{Fe}_{3-\delta}\text{O}_4$ , the Verwey transition could not be detected [27,28]. Additionally,  $T_V$  is influenced by many parameters like the growth conditions, substrate choice and film thicknesses [29–36]. We can detect the Verwey transition by measuring the magnetization vs. temperature curve. Below the Verwey transition, a transition from cubic structure to a lower symmetry monoclinic phase occurs. This transition causes a decrease of the magnetization at temperature lower than the Verwey transition. Fig. 2 (d–f) depicts the  $M(T)$  curve of the films grown at different temperatures. The  $M(T)$  curves were measured in zero-field cooling mode with an applied field of 500 Oe. In Fig. 2 (d–f), the solid black square represent the magnetization as a function of temperature  $M(T)$  of the  $\text{Fe}_3\text{O}_4$  films, which drops around 120 K corresponding to the Verwey transition. The maximum  $\partial M/\partial T$  (open red square) and the transition width  $\Delta T_V$  (the sharpness of the transition) are calculated in order to exactly determine the Verwey transition  $T_V$  and the transition width  $\Delta T_V$ , respectively. All  $\text{Fe}_3\text{O}_4$  films show almost identical  $M(T)$  curves but the magnetic properties in particular the Verwey transition of the  $\text{Fe}_3\text{O}_4$  film grown at  $T_{\text{growth}} = 400^\circ$  is the closest to the ideal bulk properties. The magnetization of this  $\text{Fe}_3\text{O}_4$  film drops by 17% at the Verwey transition of  $T_V = 117 \pm 0.6 \text{ K}$ . In addition, the transition width  $\Delta T_V$  of this film equals  $14 \pm 0.7 \text{ K}$ . For the films grown at  $T_{\text{growth}} = 500^\circ$  and  $T_{\text{growth}} = 300^\circ$ , the decrease of  $T_V$  and broadening of  $\Delta T_V$  can be attributed to the epitaxial strain, growth defects, oxygen content and reduced domain sizes [31,37]. In our work, we suggest that the decrease of  $T_V$  and broadening of  $\Delta T_V$  are mainly attributed to oxygen content and growth defects. The domain sizes are supposed to increase with the growth temperature, but for film grown at  $T_{\text{growth}} = 500^\circ$ , the  $T_V$  is reduced and  $\Delta T_V$  is broadened. Those observations suggested that the film grown at  $T_{\text{growth}} = 400^\circ$  has the highest quality in terms of oxygen content. In Table 1, the values of  $T_V$  and  $\Delta T_V$  of the  $\text{Fe}_3\text{O}_4$  films grown at different temperatures are depicted.



**Fig. 2.** Hysteresis loops for 38nm thick  $\text{Fe}_3\text{O}_4$  films grown at different growth temperatures of (a) 500°, (b) 400°, and (c) 300°. Magnetization vs. temperature (solid square), the first and second derivatives for the magnetization curve (open square and open triangle respectively) of  $\text{Fe}_3\text{O}_4$  films grown at temperatures (d) 500°, (e) 400°, and (f) 300°.

**Table 1**

Magnetization parameters of  $\text{Fe}_3\text{O}_4/\text{YSZ}$  heterostructures grown at different growth temperatures determined from the  $M(T)$  curves.

$T_{\text{growth}}$ (°C)	$T_V$ (K)	$\Delta T_V$ (K)	Magnetization drop (%)
300	115±0.8	15±0.9	12
400	117±0.6	14±0.7	17
500	113±0.8	16±0.8	13

In summary, for 38nm thick  $\text{Fe}_3\text{O}_4$  films grown on YSZ substrates, the structural properties are independent of the growth temperature. We show that the films are grown parallel to the (111) plane on the YSZ (100) with high crystalline quality indicated by Laue oscillations. However, for the magnetic measurements, there are slight differences between the films grown with different temperatures, where the  $\text{Fe}_3\text{O}_4$  film grown at  $T_{\text{growth}} = 400^\circ$  has the closest  $T_V$  to the ideal bulk value of 120K. Therefore, we chose the growth temperature of  $T_{\text{growth}} = 400^\circ$  for the subsequent thickness study.

#### 4. Film thickness: physicochemical properties

In the next step, we study the chemical properties of  $\text{Fe}_3\text{O}_4/\text{YSZ}$  heterostructures and the influence of reduced film thickness down to 4nm. To precisely investigate the stoichiometry of the films, we conducted the bulk element-sensitive HAXPES for  $\text{Fe}_3\text{O}_4$  films of thicknesses between 4 and 38nm grown on YSZ substrates at a growth temperature of 400°.

##### 4.1. HAXPES

Fig. 3 (a) shows Fe 2p core level spectra of the 4, 12 and 38 nm thick films. For the 38 nm thick  $\text{Fe}_3\text{O}_4$  film, the two satellites of the  $\text{Fe}^{2+}$  and  $\text{Fe}^{3+}$  cations, at binding energies of 715.5 eV and 719 eV [38,39], respectively, are merged. Moreover, in Fig. 3 (b), the Fe  $2p_{3/2}$  core level is depicted. The Fe  $2p_{3/2}$  core level is composed of the  $\text{Fe}^{3+}$  peak and a shoulder referring to  $\text{Fe}^{2+}$ . From both observations, a  $\text{Fe}_3\text{O}_4$  bulk-like stoichiometry of the 38 nm thick  $\text{Fe}_3\text{O}_4$  film is confirmed. Similarly, for 12 nm thick film, the two satellites are merged and a slight reduction of the Fe  $2p_{3/2}$  peak shoulder intensity are observed. On the other hand, for  $d=4$  nm, the  $\text{Fe}^{3+}$  satellite peak appears and the Fe  $2p_{3/2}$  peak shoulder intensity decreases with decreasing film thickness. This

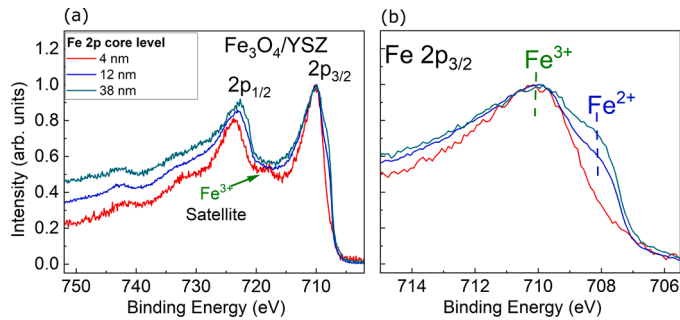


Fig. 3. (a) Fe 2p core level spectra of  $\text{Fe}_3\text{O}_4(111)/\text{YSZ}(001)$  for varying film thicknesses recorded at  $h\nu = 4 \text{ keV}$ . (b) Zoom into the  $\text{Fe } 2p_{3/2}$  core level.

indicates a decrease of the  $\text{Fe}^{2+}$  contribution and the formation of an  $\text{Fe}_2\text{O}_3$  phase in addition to the  $\text{Fe}_3\text{O}_4$  phase.

The observation of the  $\text{Fe}_2\text{O}_3$  phase may indicate an oxygen supply from the underlying YSZ substrate. This possibly causes a redox reaction with the iron oxide films at the interface as we observed also previously for  $\text{Fe}_3\text{O}_4$  films grown on  $\text{Nb:SrTiO}_3$  [18,40]. We explained the formation of the oxidized  $\gamma\text{-Fe}_2\text{O}_3$  phase at the interface by the reduction of the  $\text{TiO}_2$ -terminated  $\text{SrTiO}_3$  substrate, which in turn causes the oxidation of  $\text{Fe}_3\text{O}_4$  to form the  $\gamma\text{-Fe}_2\text{O}_3$  intralayer. However, in case of YSZ substrate, the reducibility of the YSZ is much lower compared to  $\text{SrTiO}_3$ . Moreover, the oxygen ion conductivity of the YSZ also depends on the doping percentage [41,42]. Therefore, the thermodynamics play an important role as we discuss in the next section.

## 5. Substrate-assisted interface oxidation

From bulk thermodynamics, the formation enthalpies have nearly similar values: the Gibbs free energy  $\Delta G_f^\circ$  equals  $-1200 \text{ kJ/mol}$  [43] and  $-1164 \text{ kJ/mol}$  [44] for YSZ (with  $\text{Y}_2\text{O}_3$  doping level of 10%) and for  $\text{Fe}_3\text{O}_4$ , respectively. Therefore, no clear expectation of which phase may be reduced which can be drawn. It should be noted, however, that since YSZ is such a good oxygen ion conductor, any formation of additional oxygen vacancies (concomitant with reduction of Zr) at the interface will be followed by a quick redistribution, making the concentration of excess oxygen vacancies at the interface quite small. Additionally, excess oxygen vacancies can be replenished by oxygen gas from the atmosphere.

Additionally, the YSZ is known to be a good electrolyte because of its high oxygen conductivity [45] which is influenced by the doping level as well [46]. Therefore, oxygen can diffuse from the outer atmosphere through the YSZ substrate and cause the formation of an  $\text{Fe}_2\text{O}_3$  intralayer (Fig. 4). Therefore, the oxygen conductivity of YSZ is a key factor for the formation of an oxidized  $\text{Fe}_3\text{O}_4$  intralayer. However, the perturbation of oxygen vacancy concentration caused by this redistribution within the substrate crystal plays a negligible role.

## 6. Conclusion

In summary, we investigate the bulk and emerging structural, magnetic and physicochemical interface properties of  $\text{Fe}_3\text{O}_4$  thin films ( $d = 4\text{--}38 \text{ nm}$ ) grown on  $\text{YSZ}(001)$  substrates. All  $\text{Fe}_3\text{O}_4$  thin films grow epitaxially in (111) orientation and reveal Laue oscillations as an indication for high crystalline quality. A Verwey transition was observed for all  $\text{Fe}_3\text{O}_4/\text{YSZ}$  samples.  $T_V = 117 \pm 0.6 \text{ K}$  was found for  $38 \text{ nm}$   $\text{Fe}_3\text{O}_4$  grown at  $T_{\text{growth}} = 400^\circ$ , which is closest to bulk  $T_V$ . The chemical properties were investigated by HAXPES and surprisingly reveal the formation of a few unit cells (u.c.)  $\text{Fe}_2\text{O}_3$  intralayer in  $4 \text{ nm}$   $\text{Fe}_3\text{O}_4/\text{YSZ}$  – although YSZ is considered as a quasi-inert substrate with negligible oxygen diffusivity. The formation of this  $\text{Fe}_2\text{O}_3$  intralayer can be explained by two scenarios: Either (i) a supply of oxygen from the YSZ

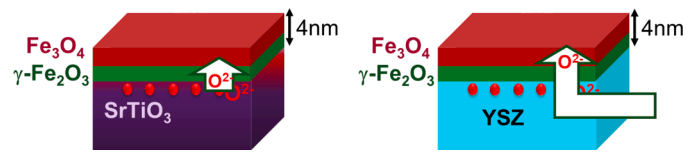


Fig. 4. The formation process of  $\text{Fe}_2\text{O}_3$  at the interfaces between the oxides substrates ( $\text{SrTiO}_3$  and  $\text{YSZ}$ ) and  $\text{Fe}_3\text{O}_4$  films has different mechanisms: (1) redox process at the interfaces  $\text{Fe}_3\text{O}_4/\text{SrTiO}_3$  and (2) oxygen diffusion through the substrate from the outer atmosphere into  $\text{Fe}_3\text{O}_4/\text{YSZ}$ .

substrate or (ii) the diffusion of oxygen through the substrate. In the first case, the resulting excess vacancies are distributed fast enough through the whole YSZ substrate, making the amount of excess vacancies very small. In the second scenario, the diffusion of oxygen is possible because YSZ is a very good oxygen conductor. In general, these findings emphasize the complex substrate–film interaction in oxide heterostructures, and the resulting opportunity to control the emerging interface functionalities—even with seemingly inert substrates.

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